

References

1. T. Nezakati, A. Seifalian, A. Tan, A. M. Seifalian. Conductive Polymers: Opportunities and Challenges in Biomedical Applications. *Chemical Reviews*, 2018, 118 (14), 6766-6843. DOI: 10.1021/acs.chemrev.6b00275.
2. J. Pecher, S. Mecking. Nanoparticles of Conjugated Polymers. *Chemical Reviews*, 2010, 110 (10), 6260-6279. DOI: 10.1021/cr100132y.
3. S. Palaniappan, A. John. Polyaniline materials by emulsion polymerization pathway. *Progress in Polymer Science*, 2008, 33(7), 732-758. <https://doi.org/10.1016/j.progpolymsci.2008.02.002>.
4. V. Nayana, B. Kandasubramanian. Polycarbazole and its derivatives: progress, synthesis, and applications. *Journal of Polymer Research*, 2020, 285(27). <https://doi.org/10.1007/s10965-020-02254-7>.
5. A. Kumar, M. Tiwari, R. Prakash. Electrochemical Study of Interfacially Synthesized Polycarbazole with Different Oxidants. *ChemElectroChem* 2015, 2, 2001. <https://doi.org/10.1002/celec.201500318>.
6. B. Gupta, A. K. Singh, R. Prakash. Electrolyte effects on various properties of polycarbazole. *Thin Solid Films*, 2010, 519(3), 1016-1019. <https://doi.org/10.1016/j.tsf.2010.08.034>.
7. W. Sangwan, N. Paradee, A. Sirivat. Polycarbazole by chemical oxidative interfacial polymerization: Morphology and electrical conductivity based on synthesis conditions. *Polymer International*, 2016, 65, 1232–1237. doi: 10.1002/pi.5186.
8. T. V. Vernitskaya, O. N. Efimov. Polypyrrole: a conducting polymer; its synthesis, properties and applications. 1997 *Russian Chemical Review*, 66, 443. DOI 10.1070/RC1997v066n05ABEH000261.
9. S. Panero, S. Passerini, B. Scrosati, Conducting Polymers: New electrochromic materials for advanced optical devices. *Mol. Cryst. Liq. Cryst.* 1993, 229 (1), 97–109.
10. A. Wang, W. Zhao and W. Yu, Effect of acid/base on the third-order optical nonlinearity of polypyrrole. *J. Mol. Struct.*, 2015, 1099, 291–296.
11. T. Onggar, I. Kruppke, C. Cherif. Techniques and Processes for the Realization of Electrically Conducting Textile Materials from Intrinsically Conducting Polymers and Their Application Potential. *Polymers*, 2020, 12, 2867. <https://doi.org/10.3390/polym12122867>.
12. L. Zhang, W. Du, , A. Nautiyal, Z. Liu, X. Zhang. Recent progress on nanostructured conducting polymers and composites: synthesis, application and future aspects. *Sci. China Mater.* 2018, 61, 303–352. <https://doi.org/10.1007/s40843-017-9206-4>.
13. M. Bharti, A. Singh, S. Samanta, D.K. Aswal. Conductive polymers for thermoelectric power generation. *Progress in Materials Science*, 2018, 93, 270-310. <https://doi.org/10.1016/j.pmatsci.2017.09.004>.

14. Nalwa, H.S. ‘ Handbook of organic conductive molecules and polymers’ vols. 1-4, John Wiley & sons, 1997.
15. Kroon R, Mengistie DA, Kiefer D, Hynynen J, Ryan JD, Yu L, et al. Thermoelectric plastics: from design to synthesis, processing and structure–property relationships. *Chem Soc Rev* 2016;45:6147–64.
16. Russ B, Glaudell A, Urban JJ, Chabynec ML, Segalman RA. Organic thermoelectric materials for energy harvesting and temperature control. *Nat Rev Mater* 2016;1:1–13.
17. Namsheer, K; Chandra Sekhar Rout. Conducting polymers: a comprehensive review on recent advances in synthesis, properties and applications. *RSC Adv.*, 2021, 11, 5659-5697. DOI: <https://doi.org/10.1039/D0RA07800J>.
18. S. Raza, X. Li, F. Soyekwo, D. Liao, Y. Xiang, C. Liu. A comprehensive overview of common conducting polymer-based nanocomposites; Recent advances in design and applications. *European Polymer Journal*, 2021, 160, 110773. <https://doi.org/10.1016/j.eurpolymj.2021.110773>.
19. M. H. Naveen, N. G. Gurudatt, Y-B. Shim. Applications of conducting polymer composites to electrochemical sensors: A review. *Applied Materials Today*, 2017, 9, 419-433. <https://doi.org/10.1016/j.apmt.2017.09.001>.
20. J. Wang, J. Dai, T. Yarlagadda. Carbon nanotube-conducting-polymer composite nanowires. *Langmuir*, 21 (2005), pp. 9-12.
21. J. G. Ibanez, M.. E. Rincón, S. Gutierrez-Granados, M’hamed Chahma, O. A. Jaramillo-Quintero, B. A. Frontana-Urbe. Conducting Polymers in the Fields of Energy, Environmental Remediation, and Chemical–Chiral Sensors. *Chemical Reviews*, 2018, 118 (9), 4731-4816. DOI: 10.1021/acs.chemrev.7b00482.
22. S. Sharma, P. Sudhakara, A.A.B. Omran, J. Singh, R.A. Ilyas. Recent Trends and Developments in Conducting Polymer Nanocomposites for Multifunctional Applications. *Polymers*, 2021, 13, 2898. <https://doi.org/10.3390/polym13172898>.
23. G. Wegner. Polymers with Metal-Like Conductivity—A Review of their Synthesis, Structure and Properties. *Angew. Chem. Int. Ed. Engl.*, 1981, 20, 361-381. <https://doi.org/10.1002/anie.198103611>
24. A.F. Diaz, Juan I. Castillo, J.A. Logan, Wen-Yaung Lee. Electrochemistry of conducting polypyrrole films. *Journal of Electroanalytical Chemistry and Interfacial Electrochemistry*, 1981, 129, 1–2, 115-132. [https://doi.org/10.1016/S0022-0728\(81\)80008-3](https://doi.org/10.1016/S0022-0728(81)80008-3).
25. Nguyen Cong, H.; El Abbassi, K.; Gautier, J. L.; Chartier, P. Oxygen Reduction on Oxide/Polypyrrole Composite Electrodes: Effect of Doping Anions. *Electrochim. Acta* 2005, 50, 1369– 1376, DOI: 10.1016/j.electacta.2004.08.025.
26. Dong, Y.-T.; Feng, J.-X.; Li, G.-R. Transition Metal Ion-Induced High Electrocatalytic Performance of Conducting Polymer for Oxygen and Hydrogen Evolution Reactions. *Macromol. Chem. Phys.* 2017, 218, 1700359, DOI: 10.1002/macp.201700359.

27. Song, F.; Li, W.; Han, G.; Sun, Y. Electropolymerization of Aniline on Nickel-Based Electrocatalysts Substantially Enhances Their Performance for Hydrogen Evolution. *ACS Appl. Energy Mater.* 2018, *1*, 3–8, DOI: 10.1021/acsaem.7b00005.
28. McQuade, D. T.; Pullen, A. E.; Swager, T. M. Conjugated Polymer-Based Chemical Sensors. *Chem. Rev.* 2000, *100*, 2537–2574, DOI: 10.1021/cr9801014.
29. Yang, G.; Kampstra, K. L.; Abidian, M. R. High Performance Conducting Polymer Nanofiber Biosensors for Detection of Biomolecules. *Adv. Mater.* 2014, *26*, 4954–4960, DOI: 10.1002/adma.201400753.
30. Zhan, L.; Chen, H.; Fang, J.; Wang, S.; Ding, L.X.; Li, Z.; Ashman, P.J.; Wang, H. Coaxial Co₃O₄ polypyrrole core-shell nanowire arrays for high performance lithium ion. *Electrochim. Acta* 2016, *209*, 192–200.
31. Chu, S., Majumdar, A. Opportunities and challenges for a sustainable energy future. *Nature* **488**, 294–303 (2012). <https://doi.org/10.1038/nature11475>
32. Hongxia Wang, Yanwei Wang, Lixing Tan, Ling Fang, Xiaohui Yang, Zhengyong Huang, Jian Li, Huijuan Zhang, Yu Wang. Component-controllable cobalt telluride nanoparticles encapsulated in nitrogen-doped carbon frameworks for efficient hydrogen evolution in alkaline conditions. *Applied Catalysis B: Environmental*, 2019, 244, 568-575. <https://doi.org/10.1016/j.apcatb.2018.11.081..>
33. Q. Gao, W. Zhang, Z. Shi, L. Yang and Y. Tang. Structural Design and Electronic Modulation of Transition-Metal-Carbide Electrocatalysts toward Efficient Hydrogen Evolution. *Adv. Mater.*, 2019, 31, 1802880.
34. Z. Yang, J. Zhang, M. C. Kintner-Meyer, X. Lu, D. Choi, J. P. Lemmon and J. Liu. Electrochemical Energy Storage for Green Grid. *Chem. Rev.*, 2011, 111, 3577–3613.
35. N. Mahmood, Y. Yao, J. W. Zhang, L. Pan, X. Zhang and J. J. Zou. Electrocatalysts for Hydrogen Evolution in Alkaline Electrolytes: Mechanisms, Challenges, and Prospective Solutions. *Adv. Sci.*, 2018, 5, 1700464.
36. Y. Hou, X. Zhuang and X. Feng. Recent Advances in Earth-Abundant Heterogeneous Electrocatalysts for Photoelectrochemical Water Splitting. *Small Methods*, 2017, 1, 1700090.
37. S. Anantharaj, S. R. Ede, K. Karthick, S. Sam Sankar, K. Sangeetha, P. E. Karthik and S. Kundu. Precision and correctness in the evaluation of electrocatalytic water splitting: revisiting activity parameters with a critical assessment. *Energy Environ. Sci.*, 2018, 11, 744–771.
38. T.T. Cheng, E.L. Gyenge. Novel catalyst-support interaction for direct formic acid fuel cell anodes: Pd electrodeposition on surface-modified graphite felt. *J. Appl. Electrochem.*, 39 (2009), pp. 1925-1938, 10.1007/s10800-009-9901-7.
39. C. Hu, L. Zhang and J. Gong. Recent progress made in the mechanism comprehension and design of electrocatalysts for alkaline water splitting. *Energy Environ. Sci.*, 2019, 12, 2620–2645.

40. Heinze, J.; Frontana-Uribe, B. A.; Ludwigs, S. Electrochemistry of Conducting Polymers—Persistent Models and New Concepts. *Chem. Rev.* 2010, 110, 4724 DOI: 10.1021/cr900226k.
41. Hao, F.; Dong, P.; Luo, Q.; Li, J. B.; Lou, J.; Lin, H. Recent advances in alternative cathode materials for iodine-free dye-sensitized solar cells. *Energy Environ. Sci.* 2013, 6, 2003 DOI: 10.1039/c3ee40296g.
42. Yun, S. N.; Hagfeldt, A.; Ma, T. L. Pt-Free Counter Electrode for Dye-Sensitized Solar Cells with High Efficiency. *Adv. Mater.* 2014, 26, 6210 DOI: 10.1002/adma.201402056.
43. Qinqin Zhou, Gaoquan Shi. Conducting Polymer-Based Catalysts. *Journal of the American Chemical Society* 2016 138 (9), 2868-2876. DOI: 10.1021/jacs.5b12474.
44. Albertas Malinauskas. Electrocatalysis at conducting polymers. *Synthetic Metals.* 107, 2, 1999, 75-83. [https://doi.org/10.1016/S0379-6779\(99\)00170-8](https://doi.org/10.1016/S0379-6779(99)00170-8).
45. Bard, Allen J.; Larry R. Faulkner (2001). *Electrochemical methods: fundamentals and applications* (Second ed.). Hoboken, NJ. ISBN 0-471-04372-9. OCLC 43859504.
46. Richard L. McCreery. Advanced Carbon Electrode Materials for Molecular Electrochemistry. *Chemical Reviews* 2008 108 (7), 2646-2687. DOI: 10.1021/cr068076m.
47. Staffell, I.; Scamman, D.; Abad, A.V.; Balcombe, P.; Dodds, P.E.; Ekins, P.; Shah, N.; Ward, K.R. The role of hydrogen and fuel cells in the global energy system. *Energy Environ. Sci.* 2019, 12, 463–491.
48. Saleem, F.; Ni, B.; Yong, Y.; Gu, L.; Wang, X. Ultra-small Tetrametallic Pt-Pd-Rh-Ag Nanoframes with tunable behavior for direct formic acid/methanol oxidation. *Small* 2016, 12, 5261–5268.
49. Das, S.; Dutta, K.; Kundu, P.P.; Bhattacharya, S.K. Nanostructured polyaniline: An efficient support matrix for Platinum-Ruthenium anode Catalyst in direct Methanol fuel cell. *Fuel Cell* 2018, 18, 369–378.
50. Ghosh, S.; Das, S.; Mosquera, M.E.G. Conducting Polymer-Based Nanohybrids for Fuel Cell Application. *Polymers* 2020, 12, 2993. <https://doi.org/10.3390/polym12122993>.
51. Muhammad G. Abd El-Moghny, Hafsa H. Alalawy, Ahmad M. Mohammad, Amina A. Mazhar, Mohamed S. El-Deab, Bahgat E. El-Anadouli. Conducting polymers inducing catalysis: Enhanced formic acid electrooxidation at a Pt/polyaniline nanocatalyst. *International Journal of Hydrogen Energy*, 42, 16, 2017, 11166-11176, <https://doi.org/10.1016/j.ijhydene.2017.01.157>.
52. R.B. Moghaddam, P.G. Pickup. Influences of aniline, carbazole, indole, and pyrrole monomers and polymers on formic acid oxidation at Pt electrodes. *Electrochim Acta*, 107 (2013), pp. 225-230.

53. Ashish Kumar, Avinash C. Pandey, Rajiv Prakash. Electro-oxidation of formic acid using polyindole-SnO₂ nanocomposite. *Catal. Sci. Technol.*, 2012,2, 2533-2538. <https://doi.org/10.1039/C2CY20382K>.
54. Weiqiang Zhou, Jingkun Xu, Yukou Du, Ping Yang. Polycarbazole as an efficient promoter for electrocatalytic oxidation of formic acid on Pt and Pt–Ru nanoparticles. *International Journal of Hydrogen Energy*, 36, 3, 2011, 1903-1912, <https://doi.org/10.1016/j.ijhydene.2010.11.023>.
55. N.R. Avery. Adsorption of formic acid on clean and oxygen covered Pt(111). *Appl. Surf Sci.*, 11(1982), PP. 774-783.
56. G.A. El-Nagar, A.M. Mohammad. Enhanced electrocatalytic activity and stability of platinum, gold, and nickel oxide nanoparticles-based ternary catalyst for formic acid electrooxidation. *Int J Hydrogen Energy*, 39 (2014), pp. 11955-11962.
57. Gejun Liu, Haipeng Bai, Prof. Dr. Bo Zhang, Prof. Dr. Huisheng Peng. Role of Organic Components in Electrocatalysis for Renewable Energy Storage. *Chem. Eur. J.* 2018, 24, 18271.
58. Weimo Li, Ce Wang, Xiaofeng Lu. Conducting polymers-derived fascinating electrocatalysts for advanced hydrogen and oxygen electrocatalysis. *Coordination Chemistry Reviews*, 464, 2022, 214555. <https://doi.org/10.1016/j.ccr.2022.214555>.
59. Jing Zhu, Liangsheng Hu, Pengxiang Zhao, Lawrence Yoon Suk Lee, Kwok-Yin Wong. Recent Advances in Electrocatalytic Hydrogen Evolution Using Nanoparticles. *Chemical Reviews* 2020 120 (2), 851-918. DOI: 10.1021/acs.chemrev.9b00248.
60. Peng Yu, Fengmei Wang, Tofik Ahmed Shifa, Xueying Zhan, Xiaoding Lou, Fan Xia, Jun He. Earth abundant materials beyond transition metal dichalcogenides: A focus on electrocatalyzing hydrogen evolution reaction. *Nano Energy*, 58, 2019, 244-276. <https://doi.org/10.1016/j.nanoen.2019.01.017>.
61. Wei, J., Zhou, M., Long, A., Xue, Y., Liao, H., Wei, C., & Xu, Z. J. Heterostructured electrocatalysts for hydrogen evolution reaction under alkaline conditions. *Nano-micro letters* 2018, 10(4), 1-15. <https://doi.org/10.1007/s40820-018-0229-x>.
62. Ivana Matanovic, Fernando H. Garzon. Nitrogen electroreduction and hydrogen evolution on cubic molybdenum carbide: a density functional study. *Phys. Chem. Chem. Phys.*, 2018,20, 14679-14687. <https://doi.org/10.1039/C8CP01643G>.
63. Chen Zhao, Xiaoteng Jia, Kewei Shu, Changchun Yu, Gordon G. Wallace, Caiyun Wang. Conducting polymer composites for unconventional solid-state supercapacitors. *J. Mater. Chem. A*, 2020,8, 4677-4699. <https://doi.org/10.1039/C9TA13432H>.
64. Ye Shi, Lele Peng, Yu Ding, Yu Zhao, Guihua Yu. Nanostructured conductive polymers for advanced energy storage. *Chem. Soc. Rev.*, 2015,44, 6684-6696. <https://doi.org/10.1039/C5CS00362H>.

65. Graeme A. Snook, Pon Kao, Adam S. Best. Conducting-polymer-based supercapacitor devices and electrodes, *Journal of Power Sources*, 196, 1, 2011, 1-12. <https://doi.org/10.1016/j.jpowsour.2010.06.084>.
66. I. Shown, A. Ganguly, L. C. Chen and K. H. Chen. Conducting polymer-based flexible supercapacitor. *Energy Sci. Eng.*, 2015, 3, 2–26.
67. Cheng Zhong, Yida Deng, Wenbin Hu, Jinli Qiao, Lei Zhang, JiuJun Zhang. A review of electrolyte materials and compositions for electrochemical supercapacitors. *Chem. Soc. Rev.*, 2015,44, 7484-7539. <https://doi.org/10.1039/C5CS00303B>.
68. Qiufeng Meng, Kefeng Cai, Yuanxun Chen, Lidong Chen. Research progress on conducting polymer based supercapacitor electrode materials. *Nano Energy*, 36, 2017, 268-285. <https://doi.org/10.1016/j.nanoen.2017.04.040>.
69. Skoog, D. A. Holler, F. J. and Crouch, S. R. “Instrumental Analysis, 6 th , Indian Reprint” (2010).
70. Harvey, D. “Modern Analytical Chemistry, A Division of the MC–Graw-Hill Companies,” 1 st Edn, (1956).
71. B.J. Inkson. 2 - Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) for materials characterization, Editor(s): Gerhard Hübschen, Iris Altpeter, Ralf Tschuncky, Hans-Georg Herrmann. *Materials Characterization Using Nondestructive Evaluation (NDE) Methods*, Woodhead Publishing, 2016, Pages 17-43. <https://doi.org/10.1016/B978-0-08-100040-3.00002-X>.
72. Stephen Brunauer, P. H. Emmett, and Edward Teller. Adsorption of Gases in Multimolecular Layers. *Journal of the American Chemical Society* 1938 60 (2), 309-319. DOI: 10.1021/ja01269a023.
73. Leech, M.C., Lam, K. A practical guide to electrosynthesis. *Nat Rev Chem* 6, 275–286 (2022). <https://doi.org/10.1038/s41570-022-00372-y>.
74. Owen J. Guy, Kelly-Ann D. Walker. Chapter 4 - Graphene Functionalization for Biosensor Applications. Editor(s): Stephen E. Sadow, *Silicon Carbide Biotechnology (Second Edition)*. Elsevier, 2016, 85-141. <https://doi.org/10.1016/B978-0-12-802993-0.00004-6>.
75. Wang, S., Zhang, J., Gharbi, O. *et al.* Electrochemical impedance spectroscopy. *Nat Rev Methods Primers* 1, 41 (2021). <https://doi.org/10.1038/s43586-021-00039-w>
76. K Mandal, D Bhattacharjee, P.S Roy., S.K Bhattacharya. S Dasgupta. Room temperature synthesis of Pd–Cu nanoalloy catalyst with enhanced electrocatalytic activity for the methanol oxidation reaction. *Applied Catalysis a: General* 2015, 100–6. <http://dx.doi.org/10.1016/j.apcata.2014.12.012>.

77. S Jeon, W Han, H An, S Im, C Yoon. Polypyrrole-modified graphitized carbon black as catalyst support for methanol oxidation. *Applied Catalysis A: General* 2011, 156-161. doi:10.1016/j.apcata.2011.09.044
78. J. Xu, D. Yuan, F. Yang, D. Mei, Z. Zhang, Y.X. Chen, On the mechanism of the direct pathway for formic acid oxidation at a Pt(111) electrode, *Physical Chemistry Chemical Physics*. 15 (2013) 4367–4376. <https://doi.org/10.1039/c3cp44074e>.
79. Y.X. Chen, M. Heinen, Z. Jusys, R.J. Behm, Kinetics and mechanism of the electrooxidation of formic acid - Spectroelectrochemical studies in a flow cell, *Angewandte Chemie - International Edition*. 45 (2006) 981–985. <https://doi.org/10.1002/anie.200502172>.
80. Y.X. Chen, M. Heinen, Z. Jusys, R.J. Behm, Bridge-bonded formate: Active intermediate or spectator species in formic acid oxidation on a Pt film electrode?, *Langmuir*. 22 (2006) 10399–10408. <https://doi.org/10.1021/la060928q>.
81. M. Baldauf, D.M. Kolb, Formic acid oxidation on ultrathin Pd films on Au(hkl) and Pt(hkl) electrodes, *Journal of Physical Chemistry*. 100 (1996) 11375–11381. <https://doi.org/10.1021/jp952859m>.
82. N. Kristian, Y. Yan, X. Wang, Highly efficient submonolayer Pt-decorated Au nanocatalysts for formic acid oxidation, *Chemical Communications*. 1 (2008) 353–355. <https://doi.org/10.1039/b714230g>.
83. G. Zhang, Y. Wang, X. Wang, Y. Chen, Y. Zhou, Y. Tang, L. Lu, J. Bao, T. Lu, Preparation of Pd-Au/C catalysts with different alloying degree and their electrocatalytic performance for formic acid oxidation, *Applied Catalysis B: Environmental*. 102 (2011) 614–619. <https://doi.org/10.1016/j.apcatb.2010.12.049>.
84. J. H. Choi, K. J. Jeong, Y. Dong, J. Han, T. H. Lim, J. S. Lee, Y.-E. Sung, Electro-oxidation of methanol and formic acid on PtRu and PtAu for direct liquid fuel cells, *Journal of Power Sources*. 163 (2006) 71–75. <https://doi.org/10.1016/j.jpowsour.2006.02.072>.
85. Q. Yang, L. Shi, B. Yu, J. Xu, C. Wei, Y. Wang, H. Chen, Facile synthesis of ultrathin Pt-Pd nanosheets for enhanced formic acid oxidation and oxygen reduction reaction, *Journal of Materials Chemistry A*. 7 (2019) 18846–18851. <https://doi.org/10.1039/c9ta03945g>.
86. M. Arenz, V. Stamenkovic, T.J. Schmidt, K. Wandelt, P.N. Ross, N.M. Markovic, The electrooxidation of formic acid on Pt-Pd single crystal bimetallic surfaces, *Physical Chemistry Chemical Physics*. 5 (2003) 4242–4251. <https://doi.org/10.1039/b306307k>.
87. Y. Tang, Y. Chen, P. Zhou, Y. Zhou, L. Lu, J. Bao, T. Lu, Electrocatalytic performance of PdCo bimetallic hollow nano-spheres for the oxidation of formic acid, *Journal of Solid State Electrochemistry*. 14 (2010) 2077–2082. <https://doi.org/10.1007/s10008-010-1018-8>.
88. D. Chen, P. Sun, H. Liu, J. Yang, Bimetallic Cu-Pd alloy multipods and their highly electrocatalytic performance for formic acid oxidation and oxygen reduction, *Journal of Materials Chemistry A*. 5 (2017) 4421–4429. <https://doi.org/10.1039/c6ta10476b>.

89. Q. Yi, A. Chen, W. Huang, J. Zhang, X. Liu, G. Xu, Z. Zhou, Titanium-supported nanoporous bimetallic Pt-Ir electrocatalysts for formic acid oxidation, *Electrochemistry Communications*. 9 (2007) 1513–1518. <https://doi.org/10.1016/j.elecom.2007.02.014>.
90. B. Habibi, N. Delnavaz, Carbon-ceramic supported bimetallic Pt-Ni nanoparticles as an electrocatalyst for oxidation of formic acid, *International Journal of Hydrogen Energy*. 36 (2011) 9581–9590. <https://doi.org/10.1016/j.ijhydene.2011.05.062>.
91. B.S. Choi, J. Song, M. Song, B.S. Goo, Y.W. Lee, Y. Kim, H. Yang, S.W. Han, Core-Shell Engineering of Pd-Ag Bimetallic Catalysts for Efficient Hydrogen Production from Formic Acid Decomposition, *ACS Catalysis*. 9 (2019) 819–826. <https://doi.org/10.1021/acscatal.8b04414>.
92. W. Chen, J. Kim, L.P. Xu, S. Sun, S. Chen, Langmuir-Blodgett thin films of Fe₂₀Pt₈₀ nanoparticles for the electrocatalytic oxidation of formic acid, *Journal of Physical Chemistry C*. 111 (2007) 13452–13459. <https://doi.org/10.1021/jp072385e>.
93. J. Pei, J. Mao, X. Liang, Z. Zhuang, C. Chen, Q. Peng, D. Wang, Y. Li, Ultrathin Pt-Zn Nanowires: High-Performance Catalysts for Electrooxidation of Methanol and Formic Acid, *ACS Sustainable Chemistry and Engineering*. 6 (2018) 77–81. <https://doi.org/10.1021/acssuschemeng.7b03234>.
94. X.M. Wang, Y.Y. Xia, The influence of the crystal structure of TiO₂ support material on Pd catalysts for formic acid electrooxidation, *Electrochimica Acta*. 55 (2010) 851–856. <https://doi.org/10.1016/j.electacta.2009.09.037>.
95. K. Ding, C. Li, Y. Zhang, L. Wang, B. Wei, X. Shi, X. He, Using PdO and PbO as the starting materials to prepare a multi-walled carbon nanotubes supported composite catalyst (Pd_xPb_y/MWCNTs) for ethanol oxidation reaction (EOR), *International Journal of Hydrogen Energy*. 43 (2018) 1523–1528. <https://doi.org/10.1016/j.ijhydene.2017.11.142>.
96. Y. Wang, S. Wang, X. Wang, CeO₂ promoted electrooxidation of formic acid on PDC nano-electrocatalysts, *Electrochemical and Solid-State Letters*. 12 (2009) 0–3. <https://doi.org/10.1149/1.3086263>.
97. S. Ramani, S. Sarkar, V. Vemuri, S.C. Peter, Chemically designed CeO₂ nanoboxes boost the catalytic activity of Pt nanoparticles toward electrooxidation of formic acid, *Journal of Materials Chemistry A*. 5 (2017) 11572–11576. <https://doi.org/10.1039/c6ta06339j>.
98. A. Altamirano-Gutiérrez, A.M. Fernández, K.K. Aruna, R. Manoharan, P. Karthikeyan, A. Siller-Ceniceros, P. Meléndez-González, P. Bartolo-Pérez, F.J. Rodríguez-Varela, Evaluation of supported and unsupported Pd–CeO₂ nanostructured anode electrocatalysts for the formic acid and the glycerol oxidation reactions in acid media, *Journal of Applied Electrochemistry*. 45 (2015) 1195–1204. <https://doi.org/10.1007/s10800-015-0858-4>.
99. S. Fierro, A. Kapałka, C. Comninellis, Electrochemical comparison between IrO₂ prepared by thermal treatment of iridium metal and IrO₂ prepared by thermal decomposition of H₂IrCl₆ solution, *Electrochemistry Communications*. 12 (2010) 172–174. <https://doi.org/10.1016/j.elecom.2009.11.018>.

100. M.S. El-Deab, Platinum nanoparticles-manganese oxide nanorods as novel binary catalysts for formic acid oxidation, *Journal of Advanced Research*. 3 (2012) 65–71. <https://doi.org/10.1016/j.jare.2011.04.002>.
101. Y.M. Asal, I.M. Al-Akraa, A.M. Mohammad, M.S. El-Deab, A competent simultaneously co-electrodeposited Pt-MnOx nanocatalyst for enhanced formic acid electrooxidation, *Journal of the Taiwan Institute of Chemical Engineers*. 96 (2019) 169–175. <https://doi.org/10.1016/j.jtice.2018.10.026>.
102. I.M. Al-Akraa, A.M. Mohammad, A spin-coated TiOx/Pt nanolayered anodic catalyst for the direct formic acid fuel cells, *Arabian Journal of Chemistry*. 13 (2020) 4703–4711. <https://doi.org/10.1016/j.arabjc.2019.10.013>.
103. A Rehman, Sk Hossain, S Rahman, S Ahmed, M Hossain. WO3 modification effects on Pt-Pd_WO3-OMC electrocatalysts for formic acid oxidation. *Applied Catalysis A: General* 2014, 482:309-317. <http://dx.doi.org/10.1016/j.apcata.2014.06.008>
104. R.K. Pandey, V. Lakshminarayanan, Electro-oxidation of formic acid, methanol, and ethanol on electrodeposited Pd-polyaniline nanofiber films in acidic and alkaline medium, *Journal of Physical Chemistry C*. 113 (2009) 21596–21603. <https://doi.org/10.1021/jp908239m>.
105. R. Yue, F. Jiang, Y. Du, J. Xu, P. Yang, Electrosynthesis of a novel polyindole derivative from 5-aminoindole and its use as catalyst support for formic acid electrooxidation, *Electrochimica Acta*. 77 (2012) 29–38. <https://doi.org/10.1016/j.electacta.2012.05.150>.
106. A. Kumar, L. Joshi, R. Prakash, Electrocatalytic performance of interfacially synthesized Au-polyindole composite toward formic acid oxidation, *Industrial and Engineering Chemistry Research*. 52 (2013) 9374–9380. <https://doi.org/10.1021/ie400915s>.
107. W. Zhou, Y. Du, H. Zhang, J. Xu, P. Yang, High efficient electrocatalytic oxidation of formic acid on Pt/polyindoles composite catalysts, *Electrochimica Acta*. 55 (2010) 2911–2917. <https://doi.org/10.1016/j.electacta.2010.01.017>.
108. M.A. del Valle, M. Gacitua, F.R. Diaz, F. Armijo, J.P. Soto, Electro-synthesis and characterization of polythiophene nano-wires/platinum nanoparticles composite electrodes. Study of formic acid electrocatalytic oxidation, *Electrochimica Acta*. 71 (2012) 277–282. <https://doi.org/10.1016/j.electacta.2012.04.001>.
109. V. Selvaraj, M. Alagar, I. Hamerton, Nanocatalysts impregnated polythiophene electrodes for the electrooxidation of formic acid, *Applied Catalysis B: Environmental*. 73 (2007) 172–179. <https://doi.org/10.1016/j.apcatb.2006.07.020>.
110. K. Ding, H. Jia, S. Wei, Z. Guo, Electrocatalysis of sandwich-structured Pd/polypyrrole/Pd composites toward formic acid oxidation, *Industrial and Engineering Chemistry Research*. 50 (2011) 7077–7082. <https://doi.org/10.1021/ie102392n>.

111. İ. Becerik, F. Kadirgan, Electro-oxidation of Formic Acid on Highly Dispersed Platinum and Perchlorate Doped Polypyrrole Electrodes, *Journal of The Electrochemical Society*. 148 (2001) D49. <https://doi.org/10.1149/1.1360186>.
112. V. Selvaraj, M. Alagar, K.S. Kumar, Synthesis and characterization of metal nanoparticles-decorated PPY-CNT composite and their electrocatalytic oxidation of formic acid and formaldehyde for fuel cell applications, *Applied Catalysis B: Environmental*. 75 (2007) 129–138. <https://doi.org/10.1016/j.apcatb.2007.03.012>.
113. S. Yang, C. Shen, Y. Liang, H. Tong, W. He, X. Shi, X. Zhang, H.J. Gao, Graphene nanosheets-polypyrrole hybrid material as a highly active catalyst support for formic acid electrooxidation, *Nanoscale*. 3 (2011) 3277–3284. <https://doi.org/10.1039/c1nr10371g>.
114. R.B. Moghaddam, P.G. Pickup, Oxidation of formic acid at polycarbazole-supported Pt nanoparticles, *Electrochimica Acta*. 97 (2013) 326–332. <https://doi.org/10.1016/j.electacta.2013.02.133>.
115. R.B. Moghaddam, P.G. Pickup, Mechanistic studies of formic acid oxidation at polycarbazole supported Pt nanoparticles, *Electrochimica Acta*. 111 (2013) 823–829. <https://doi.org/10.1016/j.electacta.2013.08.098>.
116. F. Wang, C. di Valentin, G. Pacchioni, Electronic and structural properties of WO₃: A systematic hybrid DFT study, *Journal of Physical Chemistry C*. 115 (2011) 8345–8353. <https://doi.org/10.1021/jp201057m>.
117. G. Leftheriotis, S. Papaefthimiou, P. Yianoulis, A. Siokou, D. Kefalas, Structural and electrochemical properties of opaque sol-gel deposited WO₃ layers, *Applied Surface Science*. 218 (2003) 276–281. [https://doi.org/10.1016/S0169-4332\(03\)00616-0](https://doi.org/10.1016/S0169-4332(03)00616-0).
118. Yusuf O. Ibrahim, M.A. Gondal, A. Alaswad, R.A. Moqbel, M. Hassan, E. Cevik, T.F. Qahtan, M.A. Dastageer, A. Bozkurt, Laser-induced anchoring of WO₃ nanoparticles on reduced graphene oxide sheets for photocatalytic water decontamination and energy storage, *Ceramics International*, 46 (2020) 444–451. <https://doi.org/10.1016/j.ceramint.2019.08.281>.
119. J. Georgieva, S. Sotiropoulos, E. Valova, S. Armyanov, A. Hubin, O. Steenhaut, M. Raes, Papaderakis, Pt-doped TiO₂/WO₃ bi-layer catalysts on graphite substrates with enhanced photoelectrocatalytic activity for methanol oxidation under visible light, *Journal of Photochemistry and Photobiology A: Chemistry*. 346 (2017) 70–76. <https://doi.org/10.1016/j.jphotochem.2017.05.049>
120. S.S. Kalanur, Y.J. Hwang, S.Y. Chae, O.S. Joo, Facile growth of aligned WO₃ nanorods on FTO substrate for enhanced photoanodic water oxidation activity, *Journal of Materials Chemistry A*. 1 (2013) 3479–3488. <https://doi.org/10.1039/c3ta01175e>
121. V. Lokhande, A. Lokhande, G. Namkoong, J.H. Kim, T. Ji, Charge storage in WO₃ polymorphs and their application as supercapacitor electrode material, *Results in Physics*. 12 (2019) 2012–2020. <https://doi.org/10.1016/j.rinp.2019.02.012>

122. S.A. Alqarni, M.A. Hussein, A.A. Ganash, A. Khan, Composite Material–Based Conducting Polymers for Electrochemical Sensor Applications: a Mini Review, *BioNanoScience*. 10 (2020) 351–364. <https://doi.org/10.1007/s12668-019-00708-x>
123. S. Ghosh, T. Maiyalagan, R.N. Basu, Nanostructured conducting polymers for energy applications: Towards a sustainable platform, *Nanoscale*. 8 (2016) 6921–6947. <https://doi.org/10.1039/c5nr08803h>
124. T.K. Das, S. Prusty, Review on Conducting Polymers and Their Applications, *Polymer - Plastics Technology and Engineering*. 51 (2012) 1487–1500. <https://doi.org/10.1080/03602559.2012.710697>
125. K. S.V. Santhanam, conducting polymers for based on models biosensors: Rationale based on models, *Pure & Appl. Chem.*, 70(1998), 1259-1262
126. Petr Novák, Klaus Müller, K. S. V. Santhanam, and Otto Haas, Electrochemically Active Polymers for Rechargeable Batteries, *Chemical Reviews* 1997 97 (1), 207-282 DOI: 10.1021/cr941181o
127. F. Bekkar, F. Bettahar, I. Moreno, R. Meghabar, M. Hamadouche, E. Hernáez, J.L. Vilas-Vilela, L. Ruiz-Rubio, Polycarbazole and Its Derivatives: Synthesis and Applications. A Review of the Last 10 Years, *Polymers*. 12 (2020) 2227. <https://doi.org/10.3390/polym12102227>
128. M. Kosmulski, The pH dependent surface charging and points of zero charge. VIII. Update, *Advances in Colloid and Interface Science*. 275 (2020). <https://doi.org/10.1016/j.cis.2019.102064>
129. F. Zhao, L. Yi, R. Deng, K. You, J. Song, J. Jian, P. Liu, Q. Ai, H. Luo, Supported WO₃/Γ-Al₂O₃ as bifunctional catalyst for liquid-phase highly selective oxidation of cyclohexylamine to cyclohexanone oxime under solvent-free conditions, *Molecular Catalysis*. 475 (2019). <https://doi.org/10.1016/j.mcat.2019.110494>
130. F. Zhan, J. Li, W. Li, Y. Liu, R. Xie, Y. Yang, Y. Li, Q. Chen, In situ formation of CuWO₄/WO₃ heterojunction plates array films with enhanced photoelectrochemical properties, *International Journal of Hydrogen Energy*. 40 (2015) 6512–6520. <https://doi.org/10.1016/j.ijhydene.2015.03.131>
131. S. Bai, K. Zhang, L. Wang, J. Sun, R. Luo, D. Li, A. Chen, Synthesis mechanism and gas-sensing application of nanosheet-assembled tungsten oxide microspheres, *Journal of Materials Chemistry A*. 2 (2014) 7927–7934. <https://doi.org/10.1039/c4ta00053f>
132. O. Winjobi, Z. Zhang, C. Liang, W. Li, Carbon nanotube supported platinum-palladium nanoparticles for formic acid oxidation, *Electrochimica Acta*. 55 (2010) 4217–4221. <https://doi.org/10.1016/j.electacta.2010.02.062>
133. X. Zhou, C. Liu, J. Liao, T. Lu, W. Xing, Platinum-macrocyclic co-catalysts for electrooxidation of formic acid, *Journal of Power Sources*. 179 (2008) 481–488. <https://doi.org/10.1016/j.jpowsour.2008.01.025>

134. R.S. Jayashree, J.S. Spendelow, J. Yeom, C. Rastogi, M.A. Shannon, P.J.A. Kenis, Characterization and application of electrodeposited Pt, Pt/Pd, and Pd catalyst structures for direct formic acid micro fuel cells, *Electrochimica Acta*. 50 (2005) 4674–4682. <https://doi.org/10.1016/j.electacta.2005.02.018>
135. C. Masarapu, H.F. Zeng, K.H. Hung, B. Wei, Effect of temperature on the capacitance of carbon nanotube supercapacitors, *ACS Nano*. 3 (2009) 2199–2206. <https://doi.org/10.1021/nn900500n>
136. A. Kumar, R. Prakash, Synthesis of nano ground nutshell-like polyindole by supramolecular assembled salts of ss-DNA assisted chloroauric acid, *Chemical Physics Letters*. 511 (2011) 77–81. <https://doi.org/10.1016/j.cplett.2011.05.057>
137. J. Masud, M.T. Alam, M.R. Miah, T. Okajima, T. Ohsaka, Enhanced electrooxidation of formic acid at Ta₂O₅-modified Pt electrode, *Electrochemistry Communications*. 13 (2011) 86–89. <https://doi.org/10.1016/j.elecom.2010.11.020>
138. V. Mazumder, S. Sun, Oleylamine-mediated synthesis of monodisperse Pd-based composite nanoparticles for catalytic formic acid oxidation, *ACS National Meeting Book of Abstracts*. (2009) 4588–4589
139. D. Morales-Acosta, J. Ledesma-Garcia, L.A. Godinez, H.G. Rodríguez, L. Álvarez-Contreras, L.G. Arriaga, Development of Pd and Pd-Co catalysts supported on multi-walled carbon nanotubes for formic acid oxidation, *Journal of Power Sources*. 195 (2010) 461–465. <https://doi.org/10.1016/j.jpowsour.2009.08.014>
140. J. Lee, J. Shim, J. Lee, Y. Ye, J. Hwang, S.K. Kim, T.H. Lim, U. Wiesner, One-pot synthesis of intermetallic electrocatalysts in ordered, large-pore mesoporous carbon/silica toward formic acid oxidation, *ACS Nano*. 6 (2012) 6870–6881. <https://doi.org/10.1021/nn301692y>
141. H. Ali, F.K. Kanodarwala, I. Majeed, J.A. Stride, M.A. Nadeem, La₂O₃ promoted Pd/rGO electro-catalysts for formic acid oxidation, *ACS Applied Materials and Interfaces*. 8 (2016) 32581–32590. <https://doi.org/10.1021/acsami.6b09645>
142. H. Xu, K. Zhang, B. Yan, J. Wang, C. Wang, S. Li, Z. Gu, Y. Du, P. Yang, Ultra-uniform PdBi nanodots with high activity towards formic acid oxidation, *Journal of Power Sources*. 356 (2017) 27–35. <https://doi.org/10.1016/j.jpowsour.2017.04.070>
143. M.R.A. Ramírez, M.A. del Valle, F. Armijo, F.R. Díaz, M. Angélica Pardo, E. Ortega, Enhancement of electrodes modified by electrodeposited PEDOT-nanowires with dispersed Pt nanoparticles for formic acid electrooxidation, *Journal of Applied Polymer Science*. 134 (2017). <https://doi.org/10.1002/app.44723>
144. M. v. Lebedeva, N.A. Yashtulov, V.R. Flid, Metal–Polymer Nanocomposites with Carbon Fillers for the Catalytic Oxidation of Formic Acid, *Kinetics and Catalysis*. 59 (2018) 498–503. <https://doi.org/10.1134/S0023158418040043>

145. M. Sönmez Çelebi, A.N. Yılmaz, PVF-PPy Composite as Support Material for Facile Synthesis of Pt@PVF-PPy Catalyst and Its Electrocatalytic Activity Towards Formic Acid Oxidation, *Journal of New Materials for Electrochemical Systems*. 21 (2018) 157–162. <https://doi.org/10.14447/jnmes.v21i3.502>
146. S.R. Chowdhury, T. Maiyalagan, Enhanced Electro-catalytic Activity of Nitrogen-doped Reduced Graphene Oxide Supported PdCu Nanoparticles for Formic Acid Electro-oxidation, *International Journal of Hydrogen Energy*. 44 (2019) 14808–14819. <https://doi.org/10.1016/j.ijhydene.2019.04.025>
147. Z. Liu, L. Tian, S. Xi, 1, 10-Phenanthroline: A new highly effective promoter for formic acid electrooxidation, *Materials Chemistry and Physics*. 222 (2019) 263–266. <https://doi.org/10.1016/j.matchemphys.2018.10.017>
148. L. Huang, C.Y. Zheng, B. Shen, C.A. Mirkin, High-Index-Facet Metal-Alloy Nanoparticles as Fuel Cell Electrocatalysts, *Advanced Materials*. 32 (2020). <https://doi.org/10.1002/adma.202002849>.
149. Zhu, Y.; Lin, Q.; Zhong, Y.; Tahini, H. A.; Shao, Z.; Wang, H. Metal Oxide-Based Materials as an Emerging Family of Hydrogen Evolution Electrocatalysts. *Energy and Environmental Science* 2020, 13 (10), 3361–3392. <https://doi.org/10.1039/d0ee02485f>.
150. Wang, H.; Fu, W.; Yang, X.; Huang, Z.; Li, J.; Zhang, H.; Wang, Y. Recent Advancements in Heterostructured Interface Engineering for Hydrogen Evolution Reaction Electrocatalysis. *Journal of Materials Chemistry A* 2020, 8 (15), 6926–6956. <https://doi.org/10.1039/c9ta11646j>.
151. Panwar, N. L.; Kaushik, S. C.; Kothari, S. Role of Renewable Energy Sources in Environmental Protection: A Review. *Renewable and Sustainable Energy Reviews* 2011, 15 (3), 1513–1524. <https://doi.org/10.1016/j.rser.2010.11.037>.
152. Ahmed, M.; Dincer, I. A Review on Photoelectrochemical Hydrogen Production Systems: Challenges and Future Directions. *International Journal of Hydrogen Energy* 2019, 44 (5), 2474–2507. <https://doi.org/10.1016/j.ijhydene.2018.12.037>.
153. Anantharaj, S.; Noda, S. Amorphous Catalysts and Electrochemical Water Splitting: An Untold Story of Harmony. *Small* 2020, 16 (2), 1–24. <https://doi.org/10.1002/sml.201905779>.
154. Chen, A.; Holt-hindle, P. Platinum-Based Nanostructured Materials: Synthesis, Properties, and Applications. *Chemical Reviews* 2010, 110 (6), 3767–3804. DOI: 10.1021/cr9003902.
155. Sun, H.; Yan, Z.; Liu, F.; Xu, W.; Cheng, F.; Chen, J. Self-Supported Transition-Metal-Based Electrocatalysts for Hydrogen and Oxygen Evolution. *Advanced Materials* 2020, 32 (3). <https://doi.org/10.1002/adma.201806326>.
156. Sun, H.; Xu, X.; Yan, Z.; Chen, X.; Jiao, L.; Cheng, F.; Chen, J. Superhydrophilic Amorphous Co-B-P Nanosheet Electrocatalysts with Pt-like Activity and Durability for the

Hydrogen Evolution Reaction. *Journal of Materials Chemistry A* 2018, 6 (44), 22062–22069. <https://doi.org/10.1039/C8TA02999G>.

157. Yang, L.; Xu, H.; Liu, H.; Cheng, D.; Cao, D. Active Site Identification and Evaluation Criteria of In Situ Grown CoTe and NiTe Nanoarrays for Hydrogen Evolution and Oxygen Evolution Reactions. *Small Methods* 2019, 3 (5), 1–11. <https://doi.org/10.1002/smt.201900113>.

158. Nemiwal, M.; Zhang, T. C.; Kumar, D. Graphene-Based Electrocatalysts: Hydrogen Evolution Reactions and Overall Water Splitting. *International Journal of Hydrogen Energy* 2021, 46 (41), 21401–21418. <https://doi.org/10.1016/j.ijhydene.2021.04.008>.

159. Peng, S.; Li, N.; Han, X.; Sun, W.; Srinivasan, M.; Mhaisalkar, S. G.; Cheng, F.; Yan, Q.; Chen, J.; Ramakrishna, S. Cobalt Sulfide Nanosheet/Graphene/Carbon Nanotube Nanocomposites as Flexible Electrodes for Hydrogen Evolution. *Angewandte Chemie - International Edition* 2014, 53 (46), 12594–12599. <https://doi.org/10.1002/anie.201408876>.

160. Chen, Z.; Duan, X.; Wei, W.; Wang, S.; Ni, B. J. Recent Advances in Transition Metal-Based Electrocatalysts for Alkaline Hydrogen Evolution. *Journal of Materials Chemistry A* 2019, 7 (25), 14971–15005. <https://doi.org/10.1039/c9ta03220g>.

161. Aydın, R.; Köleli, F. Hydrogen Evolution on Conducting Polymer Electrodes in Acidic Media. *Progress in Organic Coatings* 2006, 56 (1), 76–80. <https://doi.org/10.1016/j.porgcoat.2006.02.004>.

162. Xu, L.; Zhang, Y.; Feng, L.; Li, X.; Cui, Y.; An, Q. Active Basal Plane Catalytic Activity via Interfacial Engineering for a Finely Tunable Conducting Polymer/MoS₂Hydrogen Evolution Reaction Multilayer Structure. *ACS Applied Materials and Interfaces* 2021, 13 (1), 734–744. <https://doi.org/10.1021/acsami.0c20176>.

163. Zhang, G.; Lan, Z. A.; Wang, X. Conjugated Polymers: Catalysts for Photocatalytic Hydrogen Evolution. *Angewandte Chemie - International Edition* 2016, 55 (51), 15712–15727. <https://doi.org/10.1002/anie.201607375>.

164. Zheng, H.; Yang, F.; Xiong, T.; Adekoya, D.; Huang, Y.; Balogun, M. S. J. T. Polypyrrole Hollow Microspheres with Boosted Hydrophilic Properties for Enhanced Hydrogen Evolution Water Dissociation Kinetics. *ACS Applied Materials and Interfaces* 2020, 12 (51), 57093–57101. <https://doi.org/10.1021/acsami.0c16938>.

165. Dan, L. I.; Huang, J.; Kaner, R. B. Polyaniline Nanofibers: A Unique Polymer Nanostructure for Versatile Applications. *Accounts of Chemical Research* 2009, 42 (1), 135–145. <https://doi.org/10.1021/ar800080n>.

166. Ng, C. H.; Winther-Jensen, O.; Ohlin, C. A.; Winther-Jensen, B. Enhanced Catalytic Activity towards Hydrogen Evolution on Polythiophene via Microstructural Changes. *International Journal of Hydrogen Energy* 2017, 42 (2), 886–894. <https://doi.org/10.1016/j.ijhydene.2016.10.105>.

167. Kurys, Y. I.; Mazur, D. O.; Koshechko, V. G.; Pokhodenko, V. D. Electrocatalysis of Electrochemical Hydrogen Evolution from Water in Acid Media Using N-Containing Conjugated Polymers. *Theoretical and Experimental Chemistry* 2016, 52 (3), 163–169. <https://doi.org/10.1007/s11237-016-9464-8>.
168. Valiollahi, R.; Vagin, M.; Gueskine, V.; Singh, A.; Grigoriev, S. A.; Pushkarev, A. S.; Pushkareva, I. v.; Fahlman, M.; Liu, X.; Khan, Z.; Berggren, M.; Zozoulenko, I.; Crispin, X. Electrochemical Hydrogen Production on a Metal-Free Polymer. *Sustainable Energy and Fuels* 2019, 3 (12), 3387–3398. <https://doi.org/10.1039/c9se00687g>.
169. Yuan, X.; Zeng, X.; Zhang, H. J.; Ma, Z. F.; Wang, C. Y. Improved Performance of Proton Exchange Membrane Fuel Cells with P-Toluenesulfonic Acid-Doped Co-PPy/C as Cathode Electrocatalyst. *J Am Chem Soc* 2010, 132 (6), 1754–1755. <https://doi.org/10.1021/ja909537g>.
170. Jiang, L.; Wang, Z.; Geng, D.; Wang, Y.; An, J.; He, J.; Li, D.; Liu, W.; Zhang, Z. Carbon-Encapsulated Fe Nanoparticles Embedded in Organic Polypyrrole Polymer as a High Performance Microwave Absorber. *Journal of Physical Chemistry C* 2016, 120 (49), 28320–28329. <https://doi.org/10.1021/acs.jpcc.6b09445>.
171. Kumar, S.; Ranjeeth, R.; Mishra, N. K.; Prakash, R.; Singh, P. NASICON-Structured Na₃Fe₂PO₄(SO₄)₂: A Potential Cathode Material for Rechargeable Sodium-Ion Batteries. *Dalton Transactions* 2022, 51 (15), 5834–5840. <https://doi.org/10.1039/d2dt00780k>.
172. Håkansson, E.; Lin, T.; Wang, H.; Kaynak, A. The Effects of Dye Dopants on the Conductivity and Optical Absorption Properties of Polypyrrole. *Synthetic Metals* 2006, 156 (18–20), 1194–1202. <https://doi.org/10.1016/j.synthmet.2006.08.006>.
173. Ramesan, M. T. Synthesis, Characterization, and Conductivity Studies of Polypyrrole/Copper Sulfide Nanocomposites. *Journal of Applied Polymer Science* 2013, 128 (3), 1540–1546. <https://doi.org/10.1002/app.38304>.
174. Omastová, M.; Trchová, M.; Kovářová, J.; Stejskal, J. Synthesis and Structural Study of Polypyrroles Prepared in the Presence of Surfactants. *Synthetic Metals* 2003, 138 (3), 447–455. [https://doi.org/10.1016/S0379-6779\(02\)00498-8](https://doi.org/10.1016/S0379-6779(02)00498-8).
175. Zuo, W.; Zang, L.; Wang, X.; Liu, Q.; Qiu, J.; Liang, C.; Liu, X.; Yang, C. Flexible Polypyrrole@Fe₂O₃@Stainless Steel Yarn Composite Electrode for Symmetric Thread-Like Supercapacitor with Extended Operating Voltage Window in Li₂SO₄-Based Aqueous Electrolyte. *Advanced Sustainable Systems* 2020, 4 (11), 1–9. <https://doi.org/10.1002/adsu.202000173>.
176. Liu, Y.; Zhou, Y.; Zhang, J.; Xia, Y.; Chen, T.; Zhang, S. Monoclinic Phase Na₃Fe₂(PO₄)₃: Synthesis, Structure, and Electrochemical Performance as Cathode Material in Sodium-Ion Batteries. *ACS Sustainable Chemistry and Engineering* 2017, 5 (2), 1306–1314. <https://doi.org/10.1021/acssuschemeng.6b01536>.
177. Li, S.; Song, X.; Kuai, X.; Zhu, W.; Tian, K.; Li, X.; Chen, M.; Chou, S.; Zhao, J.; Gao, L. A Nanoarchitected Na₆Fe₅(SO₄)₈/CNTs Cathode for Building a Low-Cost 3.6 v Sodium-

Ion Full Battery with Superior Sodium Storage. *Journal of Materials Chemistry A* 2019, 7 (24), 14656–14669. <https://doi.org/10.1039/c9ta03089a>.

178. Dubal, D. P.; Lee, S. H.; Kim, J. G.; Kim, W. B.; Lokhande, C. D. Porous Polypyrrole Clusters Prepared by Electropolymerization for a High Performance Supercapacitor. *Journal of Materials Chemistry* 2012, 22 (7), 3044–3052. <https://doi.org/10.1039/c2jm14470k>.

179. Lattach, Y.; Fortage, J.; Deronzier, A.; Moutet, J. C. Polypyrrole-Ru(2,2'-Bipyridine)₃²⁺/MoS_x Structured Composite Film As a Photocathode for the Hydrogen Evolution Reaction. *ACS Applied Materials and Interfaces* 2015, 7 (8), 4476–4480. <https://doi.org/10.1021/acsami.5b00401>.

180. Feng, J. X.; Xu, H.; Ye, S. H.; Ouyang, G.; Tong, Y. X.; Li, G. R. Silica–Polypyrrole Hybrids as High-Performance Metal-Free Electrocatalysts for the Hydrogen Evolution Reaction in Neutral Media. *Angewandte Chemie - International Edition* 2017, 56 (28), 8120–8124. <https://doi.org/10.1002/anie.201702934>.

181. Stejskal, J.; Acharya, U.; Bober, P.; Hajná, M.; Trchová, M.; Mičušík, M.; Omastová, M.; Pašti, I.; Gavrilo, N. Surface Modification of Tungsten Disulfide with Polypyrrole for Enhancement of the Conductivity and Its Impact on Hydrogen Evolution Reaction. *Applied Surface Science* 2019, 492, 497–503. <https://doi.org/10.1016/j.apsusc.2019.06.175>.

182. Brijesh, K.; Bindu, K.; Shanbhag, D.; Nagaraja, H. S. Chemically Prepared Polypyrrole/ZnWO₄ Nanocomposite Electrodes for Electrocatalytic Water Splitting. *International Journal of Hydrogen Energy* 2019, 44 (2), 757–767. <https://doi.org/10.1016/j.ijhydene.2018.11.022>.

183. Jayaseelan, S. S.; Bhuvanendran, N.; Xu, Q.; Su, H. Co₃O₄ Nanoparticles Decorated Polypyrrole/Carbon Nanocomposite as Efficient Bi-Functional Electrocatalyst for Electrochemical Water Splitting. *International Journal of Hydrogen Energy* 2020, 45 (7), 4587–4595. <https://doi.org/10.1016/j.ijhydene.2019.12.085>.

184. Guo, S.; Jian, X.; Hou, X.; Wu, J.; Tian, B.; Tian, Y. Low-Pt Amount Supported Polypyrrole/MXene 1D/2D Electrocatalyst for Efficient Hydrogen Evolution Reaction. *Electrocatalysis* 2022, 13, 469–478. <https://doi.org/10.1007/s12678-022-00731-9>

185. Iqbal, M. F.; Yang, Y.; Hassan, M. U.; Zhang, X.; Li, G.; Hui, K. N.; Esmat, M.; Zhang, M. Polyaniline Grafted Mesoporous Zinc Sulfide Nanoparticles for Hydrogen Evolution Reaction. *International Journal of Hydrogen Energy* 2022, 47 (9), 6067–6077. <https://doi.org/10.1016/j.ijhydene.2021.11.255>.

186. Zhao, G. Q., Rui, K., Dou, S. X., Sun, W. P., Heterostructures for Electrochemical Hydrogen Evolution Reaction: A Review. *Advanced Functional Materials* 2018, 28, 1803291. <https://doi.org/10.1002/adfm.201803291>.

187. Jaiswal, A.; Pal, S.; Kumar, A.; Prakash, R.; Metal free triad from red phosphorous, reduced graphene oxide and graphitic carbon nitride (red P-rGO-g-C₃N₄) as robust electrocatalysts for hydrogen evolution reaction. *Electrochimica Acta* 2020, 338135851. <https://doi.org/10.1016/j.electacta.2020.135851>.

188. Laursen, A. B.; Kegnæs, S.; Dahl, S.; Chorkendorff, Ib; Molybdenum sulfides—efficient and viable materials for electro - and photoelectrocatalytic hydrogen evolution. *Energy Environ. Sci.* 2012, 5, 5577-5591. <https://doi.org/10.1039/C2EE02618J>.
189. Afif, A.; Rahman, S. M.; Tasfiah Azad, A.; Zaini, J.; Islan, M. A.; Azad, A. K. Advanced Materials and Technologies for Hybrid Supercapacitors for Energy Storage – A Review. *Journal of Energy Storage* 2019, 25, 100852. <https://doi.org/10.1016/j.est.2019.100852>.
190. Moradi, R; Groth, K. M. Hydrogen storage and delivery: Review of the state of the art technologies and risk and reliability analysis, *International Journal of Hydrogen Energy* 2019, 44 (23), 12254-12269. DOI: 10.1016/j.ijhydene.2019.03.041.
191. Li, C; Baek, Jong-Beom. Recent Advances in Noble Metal (Pt, Ru, and Ir)-Based Electrocatalysts for Efficient Hydrogen Evolution Reaction. *ACS Omega* 2020 5 (1), 31-40. DOI: 10.1021/acsomega.9b03550
192. Jaiswal, A; Kumar, R; Prakash, R. Iron/Iron Carbide (Fe/Fe₃C) Encapsulated in S, N Codoped Graphitic Carbon as a Robust HER Electrocatalyst. *Energy & Fuels* 2021, 35 (19), 16046-16053. DOI: 10.1021/acs.energyfuels.1c02125.
193. Ziliang Chen, Huilin Qing, Kun Zhou, Dalin Sun, Renbing Wu. Metal-organic framework-derived nanocomposites for electrocatalytic hydrogen evolution reaction. *Progress in Materials Science* 2020, 108, 100618. DOI: 10.1016/j.pmatsci.2019.100618.
194. Urhan, B. K; Doğan, H. Ö; Özer, T. Ö; Demir, Ü. Palladium-coated polyaniline nanofiber electrode as an efficient electrocatalyst for hydrogen evolution reaction. *International Journal of Hydrogen Energy* 2022, 47(7), 4631-4640. DOI: 10.1016/j.ijhydene.2021.11.101.
195. Jana, S.; Mondal, A.; Manam, J.; Das, S. Pr³⁺ doped BaNb₂O₆ reddish orange emitting phosphor for solid-state lighting and optical thermometry applications. *Journal of Alloys and Compounds* 2020, 821, 153342. DOI: 10.1016/j.jallcom.2019.153342.
196. Shalom, M; Ressnig, D; Yang, X; Clavel, G; Fellingner T. P; Antonietti, M. Nickel nitride as an efficient electrocatalyst for water splitting. *Journal of Materials Chemistry A* 2015, 3, 8171-8177. DOI: 10.1039/C5TA00078E.
197. Hongbin Zeng, Shiqi Chen, Yan Qi Jin, Jiawang Li, Jidong Song, Zhichen Le, Guofeng Liang, Hao Zhang, Fangyan Xie, Jian Chen, Yanshuo Jin, Xiaobo Chen, and Hui Meng. Electron Density Modulation of Metallic MoO₂ by Ni Doping to Produce Excellent Hydrogen Evolution and Oxidation Activities in Acid. *ACS Energy Letters* 2020, 5 (6), 1908-1915. DOI: 10.1021/acsenenergylett.0c00642.
198. Chen, Z., Cui, Y., Ye, C., Sun, Y., Zhang, J., Lv, H., Deng, L., Xu, W., Zhang, Q., Chen, G., Electrocatalytic hydrogen evolution of conducting coordination polymers based on 1,1,2,2-ethenetetrathiolate. *Journal of Polymer Science* 2022, 60(14), 2069. DOI: 10.1002/pol.20220098.

199. Tamboli, M. S.; Patil, S. A.; Tamboli, A. M.; Patil S. S.; Truong, N. T. N.; Lee, K.; Praveen, C. S.; Shrestha, N. K.; Park, C.; Kale, B. B. Polyaniline-wrapped MnMoO₄ as an active catalyst for hydrogen production by electrochemical water splitting†. *Dalton Trans.*, 2022, 51, 6027-6035. DOI: 10.1039/D2DT00032F.
200. Xie, J.; Zhang, J.; Li, S.; Grote, F.; Zhang, X.; Zhang, H.; Wang, R.; Lei, Y.; Pan, B.; Xie, Y. Controllable Disorder Engineering in Oxygen-Incorporated MoS₂ Ultrathin Nanosheets for Efficient Hydrogen Evolution. *Journal of the American Chemical Society* 2013, 135 (47), 17881-17888. DOI: 10.1021/ja408329q.
201. Xie, L.; Qu, F.; Liu, Z.; Ren, X.; Hao, S.; Ge, R.; Du, G.; Asiri, A. M.; Sun, X.; Chen, L. *In situ* formation of a 3D core/shell structured Ni₃N@Ni–Bi nanosheet array: an efficient non-noble-metal bifunctional electrocatalyst toward full water splitting under near-neutral conditions. *Journal of Materials Chemistry A* 2017, 5, 7806-7810. DOI: 10.1039/C7TA02333B.
202. Merki, D.; Hu, X. Recent developments of molybdenum and tungsten sulfides as hydrogen evolution catalysts. *Energy and Environmental Science* 2011, 4, 3878-3888. DOI: 10.1039/C1EE01970H.
203. Greeley, J.; Jaramillo, T.; Bonde, J.; Jaramillo, T. F.; Bonde, J.; Chorkendorff, I.; Nørskov, J. K. Computational high-throughput screening of electrocatalytic materials for hydrogen evolution. *Nature Materials* 2006, 5, 909–913. DOI: 10.1038/nmat1752.
204. Morales-Guio, C. G.; Stern, L.-A.; Hu, X. Nanostructured hydrotreating catalysts for electrochemical hydrogen evolution. *Chemical Society Reviews* 2014, 43, 6555-6569. DOI: 10.1039/C3CS60468C.
205. McCrory, C. C. L.; Jung, S.; Ferrer, I. M.; Chatman, S. M.; Peters, J. C.; Jaramillo, T. F. Benchmarking Hydrogen Evolving Reaction and Oxygen Evolving Reaction Electrocatalysts for Solar Water Splitting Devices. *Journal of the American Chemical Society* 2015, 137 (13), 4347-4357. DOI: 10.1021/ja510442p.
206. Chen, H.; Zheng, J.; Ballesteros-Barrientos, A.; Bing, J.; Liao, C.; Yuen, A. K. L.; Fois, C. A. M.; Valtchev, P.; Proschogo, N.; Bremner, S. P.; Atwater, H. A.; Boyer, C.; Maschmeyer, T.; Ho-Baillie, A. W. Y. Solar-Driven Co-Production of Hydrogen and Value-Add Conductive Polyaniline Polymer. *Advanced Functional Materials* 2022, 2204807. DOI: 10.1002/adfm.202204807.
207. Pei, H.; Zhang, L.; Zhi, G.; Kong, D.; Wang, Y.; Huang, S.; Zang, J.; Xu, T.; Wang, H.; Li, X. Rational construction of hierarchical porous FeP nanorod arrays encapsulated in polypyrrole for efficient and durable hydrogen evolution reaction. *Chemical Engineering Journal* 2022, 433(3), 133643, DOI: 10.1016/j.cej.2021.133643.
208. Mao, H.; Guo, X.; Fan, Q.; Fu, Y.; Yang, H.; Liu, D.; Wu, S.; Wu, Q.; Song, X.-M. Improved hydrogen evolution activity by unique NiS₂-MoS₂ heterostructures with misfit lattices supported on poly(ionic liquid)s functionalized polypyrrole/graphene oxide nanosheets. *Chemical Engineering Journal* 2021, 404, 126253. DOI: 10.1016/j.cej.2020.126253.

209. Simon, P., Gogotsi, Y. Materials for electrochemical capacitors. *Nature Mater* 7, 845–854 (2008). <https://doi.org/10.1038/nmat2297>.
210. Manopriya Samtham, Diwakar Singh, K. Hareesh, Rupesh S. Devan. Perspectives of conducting polymer nanostructures for high-performance electrochemical capacitors. *Journal of Energy Storage*, Volume 51, 2022, 104418. <https://doi.org/10.1016/j.est.2022.104418>.
211. Martin Winter and Ralph J. Brodd. What Are Batteries, Fuel Cells, and Supercapacitors? *Chemical Reviews*, 2004 104 (10), 4245-4270. DOI: 10.1021/cr020730k.
212. Yao, G., Zhang, N., Zhang, Y. *et al.* Nanostructured transition metal vanadates as electrodes for pseudo-supercapacitors: a review. *J Nanopart Res* 23, 57 (2021). <https://doi.org/10.1007/s11051-021-05158-9>.
213. Guoping Wang, Lei Zhang, JiuJun Zhang. A review of electrode materials for electrochemical supercapacitors. *Chem. Soc. Rev.*, 2012,41, 797-828. <https://doi.org/10.1039/C1CS15060J>.
214. Andy Rudge, Ian Raistrick, Shimshon Gottesfeld, John P. Ferraris. A study of the electrochemical properties of conducting polymers for application in electrochemical capacitors. *Electrochimica Acta*, Volume 39, Issue 2, 1994, Pages 273-287. [https://doi.org/10.1016/0013-4686\(94\)80063-4](https://doi.org/10.1016/0013-4686(94)80063-4).
215. Mike, J.F. and Lutkenhaus, J.L. (2013), Recent advances in conjugated polymer energy storage. *J. Polym. Sci. B Polym. Phys.*, 51: 468-480. <https://doi.org/10.1002/polb.23256>.
216. R. Ramya, R. Sivasubramanian, M.V. Sangaranarayanan. Conducting polymers-based electrochemical supercapacitors—Progress and prospects. *Electrochimica Acta*, Volume 101, 2013, Pages 109-129. <https://doi.org/10.1016/j.electacta.2012.09.116>.
217. Volfkovich, Y.M. Electrochemical Supercapacitors (a Review). *Russ J Electrochem* 57, 311–347 (2021). <https://doi.org/10.1134/S1023193521040108>.
218. Liangxu Lin, Wen Lei, Shaowei Zhang, Yuqing Liu, Gordon G. Wallace, Jun Chen. Two-dimensional transition metal dichalcogenides in supercapacitors and secondary batteries. *Energy Storage Materials*, Volume 19, 2019, Pages 408-423. <https://doi.org/10.1016/j.ensm.2019.02.023>.
219. G. Veerapandi, S. Prabhu, R. Ramesh, R. Govindan, C. Sekar. Pseudo spin-ladder CaCu₂O₃ nanostructures as potential electrode material for asymmetric supercapacitors. *Journal of Energy Storage*, Volume 48, 2022, 104051. <https://doi.org/10.1016/j.est.2022.104051>.
220. Sushmitha Veeralingam, Sivagaami Sundari Gunasekaran, Sushmee Badhulika. Bifunctional NiFe LDH as a piezoelectric nanogenerator and asymmetric pseudo-supercapacitor. *Mater. Chem. Front.*, 2022,6, 2297-2308. <https://doi.org/10.1039/D2QM00275B>

List of Publications

Patent:

1. Composites of Polypyrrole and NASICON-structured-based compounds for electrocatalytic hydrogen evolution reaction (Application No.: 202221075078).

Publications:

1. **Ajay Kumar**, Ashish Kumar, Subhajit Jana, Rajiv Prakash. Electro-Oxidation of Formic Acid on Composites from Polycarbazole and WO_3 . *Mater. Chem. Phys.* 2022, 282, 125958. <https://doi.org/10.1016/j.matchemphys.2022.125958>.

2. **Ajay Kumar**, Saurabh Kumar, Subhajit Jana, Rajpal, and Rajiv Prakash. Facile Polypyrrole/NASICON (PPy/ $\text{Na}_3\text{Fe}_2(\text{SO}_4)_2(\text{PO}_4)$) Electrode 2 Materials for the Hydrogen Evolution Reaction. *ACS Energy & Fuels* 2022, 36 (18), 11142-11153. DOI: 10.1021/acs.energyfuels.2c01893.

3. **Ajay Kumar**, Saurabh kumar, Subhajit Jana and Rajiv Prakash. Investigation of Synergistic Effect in Polypyrrole/Ni-doped NASICON Composites for Enhanced Hydrogen Evolution Reaction. *ACS Energy & Fuels* 2023, 37 (6), 4552-4565. DOI: 10.1021/acs.energyfuels.2c04178.

4. **Ajay Kumar**, Saurabh kumar and Rajiv Prakash. Investigation of Redox mediated Super capacitive property of PPy/NFS and PPy/Ni-doped NFS composites. (communicated).

5. **Ajay Kumar**, Ravi Prakash Ojha, Subhajit Jana, Radhe Shyam and Rajiv Prakash. To study the effect of Alkyl Chain on Supercapacitive Behaviour of Co-Polymer derived from Aniline and Thiophene Using electrochemical and FTM Techniques. (communicated).

6. Shaili Pal, **Ajay Kumar**, Arup Kumar De, Rajiv Prakash and Indrajit Sinha. Visible Light Photocatalysis on Magnetically Recyclable $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{O}$ Nanostructures. *Catalysis Letter*, 2022. <https://doi.org/10.1007/s10562-021-03893-1>.

7. Subham Kumar Singh, **Ajay Kumar**, Gopal Ji and Rajiv Prakash. Electrochemical and Computational Examination of Camellia Sinensis Assamica Biomolecules Ability to Retard Mild Steel Corrosion in Sodium Chloride Solutions. *J Bio Tribo Corros*, 2022, 8, 10. <https://doi.org/10.1007/s40735-021-00611-7>.

List of Publications

8. Deepti Gangwar, Priyanka Tiwari, **Ajay Kumar**, Rajiv Prakash, Chandana Rath. Effect of Dy on electrochemical supercapacitive behaviour of α -MnO₂ nanorods. *Electrochimica Acta*, 2019, 328, 135027. <https://doi.org/10.1016/j.electacta.2019.135027>.
9. Uttam Kumar, Jyoti Kuntail, **Ajay Kumar**, Rajiv Prakash, Mrinal R. Pai, Indrajit Sinha. In-situ H₂O₂ production for tetracycline degradation on Ag/s-(Co₃O₄/NiFe₂O₄) visible light magnetically recyclable photocatalyst. *Applied Surface Science*, 2022, 589, 153013. <https://doi.org/10.1016/j.apsusc.2022.153013>.
10. Shaili Pal, Sunil Kumar, Alkadevi Verma, **Ajay Kumar**, Tim Ludwig, Michael Frank, Sanjay Mathur, Rajiv Prakash, Indrajit Sinha. Development of magnetically recyclable visible light photocatalysts for hydrogen peroxide production. *Materials Science in Semiconductor Processing*, 2020, 112, 105024. <https://doi.org/10.1016/j.mssp.2020.105024>.
11. Shaili Pal, Prakash Narayan Singh, Alkadevi Verma, **Ajay Kumar**, Dhanesh Tiwary, Rajiv Prakash, Indrajit Sinha. Visible light photo-Fenton catalytic properties of starch functionalized iron oxyhydroxide nanocomposites. *Environmental Nanotechnology, Monitoring & Management*. 2020, 14, 100311. <https://doi.org/10.1016/j.enmm.2020.100311>.

List of Conferences/workshops/symposiums attended

1. International Conference on Advances in Polymer Science and Technology (APA 2018), Nepal
2. International Conference on Beyond Fossil Fuel: The Future of Alternative Energy Technologies (B-FAT 2020), IIT(BHU), Varanasi, India.
3. International Conference on Emerging Trends in Chemical Science (ICETCS 2018), Gorakhpur University, Gorakhpur.
4. National Symposium on contemporary trends and future prospects of functional materials (CTFM 2019), Department of Chemistry, Banaras Hindu University, Varanasi.
5. National Conference on Innovative Approaches Towards Sustainable Developments (NCIATSD 2020), KNPG College, Gyanpur, Bhadohi, U.P.
6. 45th National Seminar on Crystallography (NSC 45 - 2017), IIT(BHU).
7. Workshop on Research Writing and Publishing (2019). IIT(BHU).