# **CERTIFICATE**

It is certified that the work contained in the thesis titled "Studies on Magnetic and Dielectric Properties of Mn-doped CaCu<sub>3</sub>Ti<sub>4</sub>O<sub>12</sub>

*synthesized through semi-wet route*" by "*Santosh Pandey*" has been carried out under my supervision and that this work has not been submitted elsewhere for a degree.

It is further certified that the student has fulfilled all the requirements of Comprehensive examination, Candidacy and SOTA for the award of Ph.D. degree.

Prof. K, D. Mandal (Supervisor) Department of chemistry IIT (BHU), Varanasi

#### **DECLARATION BY THE CANDIDATE**

I, "Santosh Pandey", certify that the work embodied in this thesis is my own bona fide work and carried out by me under the supervision of "Prof. K. D. Mandal" from "2017" to "2022", at the "Department of Chemistry", Indian Institute of Technology (Banaras Hindu University), Varanasi. The matter embodied in this thesis has not been submitted for the award of any other degree/diploma. I declare that I have faithfully acknowledged and given credits to the research workers wherever their works have been cited in my work in this thesis. I further declare that I have not will fully copied any other's work, paragraphs, text, data, results, *etc.*, reported in journals, books, magazines, reports dissertations, theses, *etc.*, or available at websites and have not included them in this thesis and have not cited as my own work.

Date: 08/03/2022 Place: Varanasi

Santoih Panduy

Santosh Pandey

#### **DECLARATION BY THE CANDIDATE**

I, "Santosh Pandey", certify that the work embodied in this thesis is my own bona fide work and carried out by me under the supervision of "Prof. K. D. Mandal" from "2017" to "2022", at the "Department of Chemistry", Indian Institute of Technology (Banaras Hindu University), Varanasi. The matter embodied in this thesis has not been submitted for the award of any other degree/diploma. I declare that I have faithfully acknowledged and given credits to the research workers wherever their works have been cited in my work in this thesis. I further declare that I have not will fully copied any other's work, paragraphs, text, data, results, *etc.*, reported in journals, books, magazines, reports dissertations, theses, *etc.*, or available at websites and have not included them in this thesis and have not cited as my own work.

Date: 08/03/2022 Place: Varanasi

Santoih Panduy

Santosh Pandey

# CERTIFICATE BY THE SUPERVISOR/CO-SUPERVISOR(S)

It is certified that the above statement made by the student is correct to the best of my

knowledge.

Prof. Y. C. Sharma

(Head of department

Prof. K

(Supervisor)

विभागाध्यक्ष /HEAD रसायन विज्ञान विभाग Department of Chemistry भारतीय प्रौद्योगिकी संस्थान (का.हि.वि.वि.) Indian Institute of Technology (B.H.U.) वाराणसी-२२१००५ /Varanasi-221005

### COPYRIGHT TRANSFER CERTIFICATE

**Title of the Thesis:** Studies on Magnetic and Dielectric Properties of Mndoped  $CaCu_3Ti_4O_{12}$  synthesized through semi-wet route"

Name of the Student: Santosh Pandey

#### **Copyright Transfer**

The undersigned hereby assigns to the Indian Institute of Technology (Banaras Hindu University) Varanasi all rights under copyright that may exist in and for the above thesis submitted for the award of the

"Doctor of Philosophy".

Date: 00/03/2022 Place: Varanasi

Santosh Panduy Santosh Pandey 02/03/2022

**Note:** However, the author may reproduce or authorize others to reproduce material extracted verbatim from the thesis or derivative of the thesis for author's personal use provided that the source and the Institute's copyright notice are indicated.

V



It's my great pleasure to thank my *supervisor Prof. K. D. Mandal*, Department of Chemistry, Indian Institute of Technology, (B.H.U) for allowing me to join his research group, and for providing the knowledge base for my future. *Prof. Mandal* is an excellent advisor and good friend. He was always available for discussions and regularly took the time to answer my many questions. As a supervisor, he was very active in my continuing education and scientific growth, allowing me to explore different aspects of materials science independently and with guidance. In addition, he continually provided opportunities for collaborations with other scientists, which allowed me to learn more about related fields.

My sincere gratitude is also expressed to **Prof. Y.C. Sharma**, Head, Department of Chemistry, Indian Institute of Technology, (B.H.U) Varanasi, for his fruitful suggestions and providing research facilities available in the Department.

It's my great pleasure to thanks my RPEC members **Prof. D.Tiwary**, Department of Chemistry, Indian Institute of Technology, (B.H.U) and Prof. **V. L. Yadav**, Department of chemical engineering, Indian Institute of Technology, (B.H.U) for helping me and giving valuable suggestions throughout my Ph.D program.

I also want to acknowledge the members of central instrumentation facility center (CIFC) and office staff of the chemistry department for their help and support.

I would also like to thank to my lab mates Vinod Kumar and Manish Kumar Verma, shruti singh, Vishnu shankar rai and Dinesh parajapati for their active support.

I am very much thankful and grateful to all faculty members of the Department for their encouragement and support.

*I* take great pleasure in thanking my friend Manish kr. Tirpathi, Suresh kumar Pandey, Himanshu kr.singh. Department of Chemistry BHU for sharing with me all the happiness and difficulties throughout my research work.

Parents are the God on earth and I am very lucky to have very caring and loving parents Mr. Swaminath Pandey, and Mrs. Kesh kumari I would like to put in words my gratitude to my family members for making my life happy with their constant love. I wish to express my special thanks to the funding agency, IIT(BHU) for providing the financial support to carry out this work.

Last but not least, I thank almighty God for providing me strength and courage to do this work.

Date: Place: Varanasi

(Santosh Pandey)

| Title of Thesis   | i            |
|---|--------------|
| Certificate   | ii           |
| Declaration by the Candidate & Certificate by the Supervisor    | iii          |
| Copyright Transfer Certificate                                  | iv           |
| Dedication  | v            |
| Acknowledgement   | vi-vii       |
| Contents  | viii-xiii    |
| List of Figures   | xiv- xviii   |
| List of Tables  | xix          |
| List of symbols/ Abbreviation                                   | xx-xxi       |
| Preface   | xxii-xxvi    |
| CHAPTER – 1   | <b>1-5</b> 2 |
| Introduction of Perovskite                                      | 1            |
| 1.2. Perovskite substitution                                    | 3            |
| 1.2.1. Isovalent substitution                                   | 4            |
| 1.2.2. Heterovalent substitution                                | 4            |
| (a) Acceptors substitutions                                     | 4            |
| (b) Donors substitutions  | 4            |
| 1.2.3. Valence compensated substitution                         | 5            |
| 1.3. ABO <sub>3</sub> type high dielectric constant Perovskites | 5            |
| 1.3.1. CaTiO <sub>3</sub>                                       | 5            |
| (a) Applications  | 7            |
| 1.3.2. Barium Titanate (BaTiO <sub>3</sub> )                    | 7            |
| 1.3.3. SrTiO <sub>3</sub>                                       | 10           |
| 1.4. Complex Peroskite  | 13           |

| 1.4.1. CaCu <sub>3</sub> Ti <sub>4</sub> O <sub>12</sub> (CCTO)                 | 13 |
|---|----|
| 1.4.2. Y <sub>2/3</sub> Cu <sub>3</sub> Ti <sub>4</sub> O <sub>12</sub> (YCTO)  | 15 |
| 1.4.3. Bi <sub>2/3</sub> Cu <sub>3</sub> Ti <sub>4</sub> O <sub>12</sub> (BCTO) | 16 |
| 1.5. Application of Perovskite  | 18 |
| 1.6. Composite Materials  | 17 |
| 1.6.1. Natural composites   | 17 |
| 1.6.2. Early composites   | 17 |
| 1.6.3. Making composites  | 18 |
| 1.6.4. Nanocomposite  | 19 |
| 1.6.5. Ceramic matrices Nanocomposite   | 21 |
| 1.6.6. Application of composites  | 22 |
| 1.7. Ceramic Dielectrics  | 22 |
| 1.7.1. Dielectric Behaviour   | 23 |
| (a) Basic concept of dielectric capacitor                                       | 24 |
| 1.8. Polarization   | 26 |
| 1.8.1. Types of polarization:   | 26 |
| (a) Electronic polarization:  | 26 |
| (b) Ionic polarization:   | 26 |
| (c) Dipolar polarization:   | 26 |
| (d) Space charge polarization:  | 27 |
| 1.9. Dielectric constant  | 28 |
| 1.10. Dielectric loss   | 28 |
| 1.11. Basic Principles of Impedance Spectroscopy                                | 31 |
| 1.12. Conductivity  | 34 |

| 1.13. Magnetic Properties                    | 34    |
|--|-------|
| 1.13.1. Magnetism                            | 36    |
| (a) Electron magnetic moment                 | 37    |
| (b) Spin magnetic moments                    | 37    |
| 1.13.2. Diamagnetic                          | 38    |
| 1.13.3. Paramagnetic                         | 39    |
| 1.13.4. Ferromagnetism                       | 40    |
| 1.13.5. Magnetic Domain                      | 40    |
| 1.13.6. Ant ferromagnetism                   | 41    |
| 1.13.7. Ferrimagnetism                       | 41    |
| 1.13.8. Superparamagnetism                   | 42    |
| (a) Applications                             | 42    |
| 1.14. Aim of study                           | 43    |
| References                                   | 46    |
| CHAPTER – 2                                  | 53-67 |
| 2.1. Experimental                            | 53    |
| 2.2. Material used                           | 54    |
| 2.2.1 Chemicals                              | 54    |
| 2.3. Preparation of materials                | 55    |
| 2.3.1. Preparation of metal nitrate solution | 55    |
| 2.4. Synthesis methods                       | 55    |
| 2.4.1. Semi wet route                        | 55    |
| 2.4.2. Solid State Route                     | 57    |
| 2.4.2. Calcination Process                   | 57    |
| 2.4.3 Sintering Process                      | 57    |

## **CONTENTS**

| 2.5. Characterization techniques for synthesized ceramic materials: | 58     |
|---|--------|
| 2.5.1. Phase and Crystal Structure Analysis:                        | 58     |
| 2.5.3. Scanning Electron Microscopy (SEM) Analysis                  | 60     |
| 2.5.4. Energy Dispersive X-ray Analysis (EDX)                       | 61     |
| 2.5.5. Transmission Electron Microscopy (TEM) Analysis              | 62     |
| 2.5.6. Atomic force microscopy (AFM)                                | 63     |
| 2.5.7. Superconducting quantum interference device (SQUID)          | 64     |
| 2.5.8. Electric and Dielectric Measurement:                         | 65     |
| 2.5.9. Impedance and Conductivity                                   | 67     |
| CHAPTER – 3   | 68-85  |
| 3.1. Introduction   | 68     |
| 3.2. Experimental   | 69     |
| 3.2.1. Synthesis  | 69     |
| 3.2.2. Characterization   | 70     |
| 3.3. Results and discussion   | 71     |
| 3.3.1. X-ray diffraction studies                                    | 71     |
| 3.3.2. Microstructural studies                                      | 72     |
| 3.3.3. Dielectric studies   | 75     |
| 3.3.4. Magnetic studies   | 78     |
| 3.4. Conclusion   | 81     |
| References  | 82     |
| CHAPTER – 4   | 86-102 |
| 4.1. Introduction   | 86     |
| 4.2. Experimental   | 88     |
| 4.2.1. Synthesis  | 88     |

| 4.2.2. Characterization                                | 89      |
|--|---------|
| 4.3. Results and Discussion                            | 89      |
| 4.3.1. X-ray diffraction studies                       | 89      |
| 4.3.2. Microstructural studies                         | 91      |
| 4.3.3. Dielectric Studies                              | 95      |
| 4.4. Conclusion  | 98      |
| References   | 99      |
| CHAPTER – 5  | 103-120 |
| 5.1. Introduction                                      | 103     |
| 5.2. Experimental                                      | 104     |
| 5.2.1. Synthesis                                       | 104     |
| 5.2.2. Characterization                                | 105     |
| 5.3. Results and Discussion                            | 106     |
| 5.3.1. X-ray diffraction studies                       | 106     |
| 5.3.2. Microstructural studies                         | 107     |
| 5.3.3. Magnetic studies                                | 111     |
| 5.3.4. Dielectric studies                              | 114     |
| 5.4. Conclusion  | 117     |
| References   | 118     |
| CHAPTER – 6  | 121-135 |
| 6.1. Introduction                                      | 121     |
| 6.2. Experimental                                      | 123     |
| 6.2.1. Synthesis of materials                          | 123     |
| 6.2.2. Structural and microstructural characterization | 124     |
| 6.3. Result and Discussions                            | 124     |

| 6.3.1. X-ray diffraction studies | 124     |
|----------------------------------|---------|
| 6.3.2. Microstructural studies   | 125     |
| 6.3.3. Dielectric studies        | 129     |
| 6.4. Conclusion                  | 130     |
| References                       | 132     |
| Summary and future scope         | 136-138 |
| List of publications             | 139     |

| <b>Figure 1.1</b> A perovskite unit cell that displays the titanium ion off-centered.                         | 6      |
|---|--------|
| Figure 1.2 Schematic of the perovskite structure of BaTiO <sub>3</sub> (A) Cubic lattice (above               |        |
| Curie temperature, 120°C)   |        |
| ( <b>B</b> ) Tetragonal lattice (below Curie temperature, 120°C 8   |        |
| Figure 1.3. Reversal toward spontaneous polarization in BaTiO <sub>3</sub> by reversal of the appli           |        |
| ed field direction.   | 9      |
| Figure 1.4. Atomic structure of SrTiO <sub>3</sub> at Room Temperature.                                       | 12     |
| Figure 1.5. Atomic arrangements for the <100>, <110> and <111> axial direction                                | in     |
| SrTiO <sub>3</sub> . 12   |        |
| Figure 1.6 Crystal structure of CCTO compound.  | 14     |
| Figure 1.7. Crystal structure of BCTO, in which 1/3Bi sites are vacant. The Ti atoms sit                      |        |
| at the center of the $TiO_6$ Octahedra.   | 17     |
| Figure 1.8. A polarized dielectric material.  | 23     |
| Figure 1.9. Typical response of the total polarizability of a crystal as a function of                        |        |
| electric field Frequency.   | 27     |
| Figure 1.10. Types of polarization mechanisms   | 28     |
| Figure 1.11.Simulated IS data for two RC elements connected in series, presented                              | in     |
| different formats: Z'' $[\Omega]$ vs. frequency (f), M'' vs. f, Z'' $[\Omega]$ vs. Z' $[\Omega]$ and C' [Fara | d] vs. |
| f.  | 33     |
| Figure 1.12. Types of Magnetism.  | 36     |
| Figure 1.13. Flow chart of Magnetism.   | 38     |
| Figure 1.14. Ferromagnetic and super paramagnetic.42  |        |
| Figure 2.1. Flow chart for the synthesis of materials by semi-wet route54                                     |        |

| Figure 2.2. Powder XRD instrument, RigakuMiniflex600 (Japan)                                  | 59         |
|---|------------|
| Figure 2.3 Braggs law of diffraction  | 59         |
| Figure 2.4 Scanning Electron microscopy (SEM, ZEISS model, EVO-18 Ger                         | many) and  |
| EDX Analysis instrument (Oxford instrument; USA)  | 62         |
| Figure 2.5 Transmission electron microscope (TEM, FEI Tecnai-20G2)                            | used for   |
| determination of particle structure.  | 63         |
| Figure 2.6 Superconducting quantum interference device (SQUID) (Quantu                        | m Design,  |
| MPMS 3)   | 65         |
| Figure 2.7 LCR Meter (PSM 1735, Newton 4th Ltd, U.K.) used for                                | dielectric |
| measurement.  | 67         |
| Figure 3.1 XRD diffraction pattern of sintered CCTMO at 1223K for 8 h.                        | 72         |
| Figure 3.2 (a) Bright field TEM images (b) Selected area diffraction patter                   | ern        |
| (SEAD) of CCTMO ceramics sintered at 1223K at 8h. 73  |            |
| Figure 3.3 SEM images of CCTMO ceramics sintered at 1223K for 8h,                             | (a)        |
| SEM images of CCTMO (b) EDX spectra of CCTMO ceramic. 74                                      |            |
| Figure 3.4 AFM images of CCTMO ceramics sintered at 1223 K for 8 h, (a) 2                     | 2-         |
| dimentional structure (b) 3-dimenional structure  |            |
| (c) bar diagram of particle size. 7   | 75         |
| Figure 3.5 Temperature dependent (a) dielectric constant ( $\epsilon_r$ ), and (b) dielectric | ic         |
| loss (tan $\delta$ ) at few selected frequencies. 77  | 7          |
| Figure 3.6 Frequency dependent (a) dielectric constant ( $\epsilon_r$ ) and (b) Dielectric    |            |

loss (tan  $\delta$ ) at few selected Temperature. 77

Figure 3.7 Temperature dependent (a) magnetic moment noted at  $\pm$  2 T and magnetic field at 100 Oe. (b) M-H hysteresis at 5 and 300 K for CCTMO ceramic. 79

Figure 3.8 Magnetic susceptibility as a function of temperature recorded at ± 2T and applied magnetic field (H) at 100 Oe.80

Figure 3.9 P-E hysteresis loop for CaCu<sub>3</sub>TiMnO<sub>12</sub> sintered at 1223 K for 8 h. 81

 Figure 4.1 XRD patterns of CaCu<sub>3</sub>Ti<sub>4-x</sub>Mn<sub>x</sub>O<sub>12</sub> (a) x = 0.25 (b) x = 0.50 (c) x = 1.00 

 sintered at 1223 K for 8 h.

Figure 4.2 Bright field TEM images and their corresponding SEAD patterns of sintered

$$CaCu_{3}Ti_{4-x}Mn_{x}O_{12}$$
 ceramic (0.25, 0.50 and 1.00). 92

Figure 4.3 SEM micrograph of CaCu<sub>3</sub>Ti<sub>4-x</sub>Mn<sub>x</sub>O<sub>12</sub> ceramics (a) x = 0.25 (b) x = 0.50 (c) x = 1.00 and EDX spectra of CaCu<sub>3</sub>Ti<sub>4-x</sub>Mn<sub>x</sub>O<sub>12</sub> ceramics (d) x = 0.25 (e) x = 0.50 (f) x = 1.00 sintered at 1223 K for 8 h. 93

Figure 4.4 AFM images of  $CaCu_3Ti_{4-x}Mn_xO_{12}$  (x=1.00) ceramics sintered at 1223 K for 8 h (a) 2-dimentional structure (b) 3-dimensional structure (c) bar diagram of particle size. 94

**Figure 4.5** Dielectric constant ( $\varepsilon_r$ ) as the function of frequency for CaCu<sub>3</sub>Ti<sub>4-x</sub>Mn<sub>x</sub>O<sub>12</sub> ceramics (x=0.25, 0.50 and 1.00) sintered at 1223 K for 8 h. 96

**Figure 4.6** Dielectric loss (tan  $\delta$ ) as the function frequency CaCu<sub>3</sub>Ti<sub>4-x</sub>Mn<sub>x</sub>O<sub>12</sub> ceramics (x=0.25, 50 and 1.00) sintered at 1223 K for 8 h. 96

Figure 4.7 P-E hysteresis loop for  $CaCu_3Ti_{4-x}Mn_xO_{12}$  ceramics (x= 0.25, 0.50 and 1.00)at room temperature.97

Figure 5.1 XRD patterns of CCTMO sintered at (a) 1223 K (b) 1323 K (c) 1373 K for 8 h. 106 Figure 5.2 Bright-field TEM images at (a) 1223 K (b) 1323 K (c) 1373 K and (d)Selected area diffraction pattern of CCTMO at 1373 K.108

Figure 5.3 (a-c) SEM micrograph and (d-e) EDX spectra of CCTMO ceramic sinteredat 1223 K, 1323 K, and 1373 K, respectively for 8 h.110

**Figure 5.4** AFM images of CCTMO ceramics sintered at 1373 K for 8 hours (a) 2dimensional structure (b) 3-dimensional structure

(c) bar diagram of particle size. 112

Figure 5.5 Temperature-dependent (a) magnetic moment noted applied magnetic fieldat 100 Oe. (b) M-H hysteresis at 5 and 300 K for CCTMO ceramic sintered at 1323 Kfor 8 h.112

**Figure 5.6** Magnetic susceptibility as a function of temperature recorded at  $\pm 2$  T and applied magnetic field (H) at 100 Oe. 113

Figure 5.7 The polarization versus electric field (P-E) hysteresis loop ofCaCu3Ti3.5Mn0.5O12 at sintered at 1223 K, 1323 K, and 1373 K, respectively.114

**Figure 5.8** Temperature-dependent (a) dielectric constant ( $\epsilon r$ ) and (b) loss tangent (tan  $\delta$ ) at 10 kHz for sintered at 1223 K, 1323 K, and 1373 K,

respectively for 8 h. 116

**Figure 5.9** Frequency-dependent (a) dielectric constant ( $\epsilon r$ ) and (b) loss tangent (tan  $\delta$ ) at room temperature for sintered at 1223 K, 1323 K, and 1373 K, respectively. 117

Figure 6.1 XRD patterns of CaCu<sub>3</sub>Ti<sub>3.5</sub>X<sub>0.5</sub>O<sub>12</sub> (a) X = Mn (b) X = Nb (c) X = W sintered at at 1223 K, 1323 K and 1373 K, respectively for 8 h. 125

**Figure 6.2** Bright-field TEM images and their corresponding SEAD patterns of CaCu<sub>3</sub>Ti<sub>3.5</sub>X<sub>0.5</sub>O<sub>12</sub> ceramics (a-b) X= Mn (c-d) X= Nb (e-f) X= W (Mn, Nb and W) sintered at 1223 K, 1323 K and 1373 K, respectively for 8 h. 126

Figure 6.3 SEM micrograph of CaCu<sub>3</sub>Ti<sub>3.5</sub>X<sub>0.5</sub>O<sub>12</sub> ceramics (a) X = Mn (b) X = Nb (c)X = W and EDX spectra of CaCu<sub>3</sub>Ti<sub>3.5</sub>X<sub>0.5</sub>O<sub>12</sub> ceramics (d) X = Mn (e) X = Nb (f) X = Wsintered at 1223 K 1323 K and 1373 K, respectively for 8 h.

**Figure 6.4** X-ray photo emission spectroscopy (a) Mn 2p, (b) Nb 3d, (c) W 4f, for CCTXO (X= Mn, Nb, W) sintered at 1223 K, 1323 K and 1373 K,

128

respectively for 8 h.

Figure 6.5(a). Dielectric constant (b) dielectric loss dependent on the frequency at a fewselected temperatures of CCTMO sintered at 1223 K for 8h.130

Figure 6.6(a). Dielectric constant (b) dielectric loss dependent on the frequency at a fewselected temperatures of CCTNO sintered at 1323 K for 8h.130

Figure 6.7(a). Dielectric constant (b) dielectric loss dependent of frequency at a fewselected temperatures of CCTWO sintered at 1373 K for 8h.130

| <b>Table 1.1</b> : Summary of the physical properties of SrTiO <sub>3</sub> .  |    |
|--|----|
| Table 2.1 Specification of the chemical used.  |    |
| Table 4.1 Atomic percentage of elements for CaCu <sub>3</sub> Ti <sub>(4-x)</sub> Mn <sub>x</sub> O <sub>12</sub> ceramics                   |    |
| (x= 0.25, 0.50 and 1.00) sintered at 1223 K for 8 h.   | 93 |
| <b>Table 5.1</b> Atomic percentage of elements for CaCu <sub>3</sub> Ti <sub>3.5</sub> Mn <sub>0.5</sub> O <sub>12</sub> sintered at 1223 K, |    |
| 1323 K and 1373 K, respectively for 8 h. 109   |    |

| E              | Permittivity or dielectric constant        |
|----------------|--|
| ε*             | Complex Quantity of dielectric constant    |
| ٤'             | real components of dielectric constant     |
| ٤''            | Imaginary component of dielectric constant |
| i              | Imaginary number                           |
| ε <sub>0</sub> | Permittivity of free space                 |
| ε <sub>r</sub> | relative dielectric constant               |
| С              | Capacitance                                |
| F              | Farad                                      |
| tan δ          | tangent loss                               |
| σ              | conductivity                               |
| f              | frequency                                  |
| λ              | Wavelength                                 |
| θ              | Angle theta                                |
| °C             | Degree centigrade                          |
| Κ              | Kelvin                                     |
| k <sub>B</sub> | Boltzmann constant                         |
| Тв             | Blocking temperature                       |
| X              | Magnetic susceptibility                    |
| С              | Curie constant                             |
| М              | Magnetization                              |
| Н              | Magnetic field                             |
| Oe             | Oersted                                    |
| Р              | Density                                    |
| В              | Induced magnetic field                     |
| E              | Electric field                             |

| Р               | Net polarization              |
|-----------------|-------------------------------|
| P electronic    | Electronic polarization       |
| P ionic         | Ionic polarization            |
| P molecular     | Molecular polarization        |
| P interfacial   | Interfacial polarization      |
| Hz              | hertz                         |
| ω               | angular frequency             |
| τ               | Relaxation time               |
| t               | tolerance factor              |
| Å               | angstrom                      |
| R               | Resistance                    |
| С               | Capacitance                   |
| R <sub>b</sub>  | Resistance of bulk            |
| C <sub>b</sub>  | Capacitance of bulk           |
| R <sub>gb</sub> | Resistance of grain boundary  |
| $C_{gb}$        | Capacitance of grain boundary |
| eV              | electron Volt                 |

A capacitor material which has high dielectric constant and low dielectric loss is an interesting topic in materials science. It may be used as multilayer capacitor (MLCC), dynamic random access memory (DRAMs), microwave devices, electronic devices in automobiles andaircrafts.ACu<sub>3</sub>Ti<sub>4</sub>O<sub>12</sub> (A = Ca, Bi, Sr) type oxides had complex perovskite structure and discovered in 1967 by Subramanian *et al.* It produces a high dielectric constant ( $\varepsilon_r \sim 10^4$ ) and nearly constant in the temperature range of 100–600 K. which has led to many important applications. The high dielectric loss of CCTO ceramics (tan  $\delta > 0.05$  at 1 kHz) is still the most serious problem for applications requiring capacitive components. Presently we use simple perovskite BaTiO<sub>3</sub>, which are not environmentally friendly as capacitor materials. The problem with BaTiO<sub>3</sub> is that it is quite unstable at higher and shows phase transition. Therefore, it is not suitable for use at high temperature. Therefore, Development of excellent dielectric materials with good stability over wide temperature and frequency ranges are highly desired.

The modern age technology needs for the development of electrical composite, that properties are not available in individual single component materials. The required combination properties are possible to tailor composite by combining two or more components. Composites perovskite play a very important role in various areas of chemistry, physics, biology and materials science because of their interesting properties. When two or more perovskites are mixed together either by physical or by chemical methods to fabricate composite, a novel set of physical and chemical properties may be obtained that would be completely different from that of the individual constituents.

Electrical and dielectric properties of the composite are also very important. Which leads to data storage, tunnel junction, and spin valves. The composite has also been increasing interest in flexible, high dielectric constant and a polymer for use in

#### **PREFACE**

high-density energy storage and capacitor applications. With the smaller sizes of nanoparticle less than 100 nm surfaces to volume ratio increases resulting in the number of atoms on the surface of nanocrystals, therefore variation in electrical properties with change in structure in the nanoscale region is observed in comparison to the bulk material. The electrical and dielectric properties of nanoparticle are affected by particle size, morphology, and chemical composition.

In the present work synthesis of composite perovskite with different composition using semi-wet route. All the synthesized composites were characterized by various physicochemical techniques to study the crystal structure, particle size and shape whereas electrical and dielectric, properties of materials were studied in detail. And it also studied the effect of sintering duration of these composite.

The present work aims to investigate (a) crystal structure (b) microstructure (c) elemental analysis (d) particle size (e) electrical and dielectric behavior of the following compound prepared by semi-wet route.

1. CaCu<sub>3</sub>Ti<sub>3</sub>MnO<sub>12</sub>

- 2. CaCu<sub>3</sub>Ti<sub>3.5</sub>Mn<sub>0.5</sub>O<sub>12</sub>
- 3. CaCu<sub>3</sub>Ti<sub>3.75</sub>Mn<sub>0.25</sub>O<sub>12</sub>
- 4. CaCu<sub>3</sub>Ti<sub>3.5</sub>W<sub>0.5</sub>O<sub>12</sub>
- 5. CaCu<sub>3</sub>Ti<sub>3.5</sub>Nb<sub>0.5</sub>O<sub>12</sub>

**Chapter I** This chapter contains a brief introduction of the subject describing briefly the technical investigations reported in the field of perovskite oxides and composite materials. Polarization also describes which related to dielectric properties as well as the

frequency of perovskite. It contains basic knowledge of Impedance spectroscopy which separates the contributions of the grains and grain boundaries, and electrode specimen interface observed RC elements of the composite. This includes the effect of isovalent, heterovalent and valence compensated substitutions on the electrical and dielectric properties.

**Chapter II** This chapter describes the details of experimental procedure used for the synthesis, characterization, and application of composite materials. The crystalline phases of composite sintered samples were identified by using the X-ray diffraction analysis (Rigaku, miniflex-600, Japan) employing Cu-k $\alpha$  radiation.Scanning Electron Microscopy gives an idea of formation of the microstructure of these materials. Transmission Electron Microscopy (TEM) has been used for determination of their size and shape of the particle. Atomic force microscopy analyzed the surface morphology. Electrical and dielectric properties which are characteristic of all the composite were measured as a function of temperature (300-500 K) in the frequency range 100Hz-5 MHz with the help of PSM 1735 (NumetriQ 4<sup>th</sup> U.K Limited) LCR Meter.

**Chapter III** The detailed synthesis, characterization and application of the CaCu<sub>3</sub>Ti<sub>4</sub>MnO<sub>12</sub> (CCTMO) perovskite were described in this chapter. CCTMO was synthesized using a semi-wet method through sintering at 1223 K for 8 h. The structural and microstructural details were studied by X-ray diffraction (XRD), scanning electron microscope (SEM) and transmission electron microscope (TEM) techniques. XRD analysis confirmed the existence of CaCu<sub>3</sub>Ti<sub>4</sub>O<sub>12</sub> (CCTO) as the primary phases along with TiO<sub>2</sub> as the minor aspects. The average grain sizes obtained by SEM analysis were found to be around 1.46 µm sintering for 8 h, respectively. TEM analysis showed the particle size in the range of 43.76  $\pm$  10 nm. The surface morphology was analyzed by

xxiv

atomic force microscopy (AFM). The sample sintered for 8 h exhibited very high dielectric constant ( $\epsilon_r \sim 100$ ) at 1 kHz and 303 K. The presence of semiconducting grains with the insulating grain boundaries significantly attributes to such a high dielectric constant value, supporting the internal barrier layer capacitance (IBLC) mechanism operative in CCTMO perovskite.

**Chapter IV** In this chapter, the CaCu<sub>3</sub>Ti<sub>(4-x)</sub>Mn<sub>x</sub>O<sub>12</sub> (CCTMO) was synthesized by a semi-wet method at 1223 K for 8 h. X-ray diffraction (XRD) analysis confirms the presence of CCTMO and CaCu<sub>3</sub>Ti<sub>4</sub>O<sub>12</sub> both phases in the perovskite ceramic. Transmission electron microscope (TEM) analysis of the composite demonstrates the formation of nanoparticles with average particle size  $23 \pm 10$  nm,  $31 \pm 10$  nm and  $24 \pm 10$  nm at a different doping concentration of Mn (x= 0.25, 0.50 and 1.00) in CaCu<sub>3</sub>Ti<sub>4</sub>-xMn<sub>x</sub>O<sub>12</sub> ceramic.. The surface morphology of the composite sintered at 1223 K for 8 h obtained by SEM analysis indicate the formation of large and small grains with bimodal structure. The average and root mean square roughness is found to be 72 nm and 90 nm, respectively by Atomic force microscopy studies of the ceramic. The dielectric constant of CCTMO ceramic is found to be 150 at 100 Hz and 500 K respectively. The presence of semiconducting grains and insulating grain boundaries in the composite supporting the internal barrier layer capacitance (IBLC) mechanism operative in Mn-doped CCTO type of perovskites of different composition.

**Chapter V** The synthesis, characterization and application of the CaCu<sub>3</sub>Ti<sub>3.5</sub>Mn<sub>0.5</sub>O<sub>12</sub> ceramic were discussed in this chapter. A nano-composite ceramic with the chemical composition CaCu<sub>3</sub>Ti<sub>3.5</sub>Mn<sub>0.5</sub>O<sub>12</sub> was synthesized by a semi-wet method at CCTMO sintered at 950 °C, 1050 °C, and 1100 °C, respectively for 8 h. X-ray diffraction analysis confirms the presence of CCTO and TiO<sub>2</sub> phases in the composite ceramic.

Transmission electron microscope analysis of the formation of nano-particles (98.49  $\pm$  10 nm, 92.95  $\pm$  10 nm and 145.50  $\pm$  10 nm at 950 °C, 1050 °C, and 1100 °C, respectively. Further, scanning electron microscope (SEM) images show that the morphology consists of large and small grains (1.0–10 µm) with a bimodal distribution. The surface morphology of composite was studied by atomic force microscope using tapping mode of measurement also substantiates the results obtained by SEM analysis. The sample sintered for 8 h exhibits very high dielectric constant ( $\varepsilon_r \approx 130$ ) at 100 Hz and room temperature. The presence of semiconducting grains with insulating grain boundaries significantly attributes to such a high dielectric constant value, supporting the internal barrier layer capacitance mechanism operative in Mn-doped CCTO of different composition.

Chapter VI the CCTMO, CCTWO and CCTNO perovskites were successfully synthesized by semi-wet route. Powder X-Ray Diffraction confirms the formation of CCTO as main phase along with minor TiO<sub>2</sub> phase in CCTMO, CCTWO and CCTNO at sintered at 950°C, 1050°C, and 1100°C, for 8 h. Particle size observed by TEM is 44 nm, 101 nm, and 51 nm, respectively. Atomic force microscopy shows statistically significant changes in the surface roughness. The nano-composite exhibits improvement in dielectric loss (tan  $\delta \approx 0.9$ ) at 1 kHz. The low-frequency performance of the doped CCTO was estimated by measuring the frequency dispersion of the dielectric constant ( $\epsilon$ ') and dielectric loss (tan  $\delta$ ).