Chapter 4

STRUCTURAL ANALYSIS OF H02Ge2O7 AND EFFECT OF NEGATIVE CHEMICAL PRESSURE UPON ITS MAGNETIC PROPERTIES.

4.1 Introduction

The ground states of magnetically frustrated systems (transition metal oxides) are essentially understood by considering two energy scales. The on-site Coulomb repulsion- Hubbard U and the spin-orbit coupling λ [88]. Both of these parameters largely depend upon the topology of ligands. Hence first it is essential to establish a proper structural analysis of a system, since the arrangement of ligands in such systems imparts a strong crystal field effect upon the rareearth (RE) ion. The ground spin-state is decided by the low-temperature magnetic interaction between two magnetic RE ions which is described through the Dipolar Spin-Ice Model (DSIM) as discussed in chapter 1. It involves the complexity of low-temperature ferromagnetic (FM) dipolar (D_{nn}) and antiferromagnetic (AFM) exchange interaction (J_{nn}). The effect of these magnetic interactions is generally visible at a temperature (T) ≈ 2 K which leads to RE ion-dependent nontrivial magnetic ground states.

Pyrogermanates, having an entirely different crystallographic structure than the titanates, also belong to the class of frustrated system, but its place in spin-ice family is yet a matter of investigation. This is because the magnetic ground state for Ho₂Ge₂O₇ resembles that of the spiral arrangement of spins rather than ice-like spin arrangement for conventional cubic pyrochlores, though both exhibits similar spin relaxation feature as seen in ac-susceptibility

response, discussed in chapter 1 [62].

The exact Curie-Weiss temperature as well as the magnetic ground state is yet unknown in $Ho_2Ge_2O_7$ as could be seen from **Table 4.1** given below.

Table 4.1 Curie-Weiss temperature, magnetic ground state, lattice parameter and anisotropy exhibited in few A2B2O7. [59]

A(B) (A ₂ B ₂ O ₇)	$\theta_{\rm C}$ or $\theta_{\rm N}$	Anisotropy	Anisotropy Ground State	
Pr(Pb)	?	Easy axis	Quantum spin ice?	10.8721(9)
Nd(Pb)	0.41 K	Easy axis?	Antiferromagnet?	10.8336(4)
Gd(Pb)	0.8 K	Isotropic	Ferromagnet?	10.7292(8)
Tb(Ge)	?	Easy axis	Spin liquid	9.9617(1)
Dy(Ge)	?	Easy axis	Spin ice	9.9290(5)
Ho(Ge)	?	Easy axis	Spin ice	9.9026(6)
Er(Ge)	1.41 K	Easy plane	Antiferromagnet?	9.8767(3)
Yb(Ge)	0.61 K	Easy plane	Antiferromagnet?	9.8306(6)

Making an appreciable enhancement in the pyrochlore lattice parameter by replacing Ge^{4+} with Ti^{4+} in Ho₂Ge₂O₇ matrix at B site (ionic radii of $Ge^{4+} < Ti^{4+}$), it is possible to tune J_{nn} since it is more sensitive to change in the lattice parameter as well as the distance between the rare earth spins. Induction of this negative chemical pressure effect in this magnetically frustrated system facilitates to study the nature of ground state spin dynamics through M-T, M-H, and ac-susceptibility measurements. This chapter presents a comprehensive structural analysis of powdered Ho₂Ge₂O₇ along with the magnetic response. Consequential efforts were made to consolidate and compare pyrochlore's structural and magnetic properties with pyrogermanates.

4.2 Structural analysis

The crystal structure of Ho₂Ge₂O₇ follows tetragonal non-centrosymmetric P4₁2₁2 space group symmetry. The Wyckoff positions for the P4₁2₁2 tetragonal space group, as obtained from the International Tables for Crystallography (2006). Vol. A Space group 92, pp. 368–369. is mentioned in **Table 4.2** [89].

Multiplicity	Wyckoff letter	Site symmetry	Coordinates
8	b	1	$\begin{array}{l} (x,y,z) (-x,-y,z+1/2) (-y+1/2,x+1/2,z+1/4) \\ (y+1/2,-x+1/2,z+3/4) (-x+1/2,y+1/2,-z+1/4) \\ (x+1/2,-y+1/2,-z+3/4) (y,x,-z) (-y,-x,-z+1/2) \end{array}$
4	а	2	(x,x,0) (-x,-x,1/2) (-x+1/2,x+1/2,1/4) (x+1/2,- x+1/2,3/4)

 Table 4.2 Wyckoff positions of space group = P41212. [89]
 P41212. [89]

4.2.1 Rietveld refinement of obtained HRXRD pattern of Ho₂Ge₂O₇

There are four independent sites of oxygen atoms for this particular space group, three of whose does not restrict *x*, *y*, and *z* positions. The fourth oxygen at 4a Wyckoff site poses restriction in coordinates with x=y and z=0. Using these designated Wyckoff positions for the tetragonal system as listed in **Table 4.2** Rietveld refinement of the obtained HRXRD pattern of Ho₂Ge₂O₇ was obtained. **Figure 4.1** presents the room temperature HRXRD pattern of powdered Ho₂Ge₂O₇, with no extra reflections indicating the absence of any impurity phases. We established a satisfactory agreement between the calculated and observed data with the value of $R_{wp}=15.7$, $R_e= 8.26$, $R_p=12.7$, and $\chi^2 = 3.62$. The results of the structural investigations (unit cell parameters and Wyckoff positions) of Ho₂Ge₂O₇ are summarized in *School of Materials Science and Technology, IIT (BHU), Varanasi*

Table 4.3. The lattice constants obtained are a = b = 6.8066(5) Å; c = 12.3803(1) Å and unit cell volume V = 573.584 ± 0.008 Å³.



Figure 4.1 Room-temperature high-resolution x-ray diffraction pattern of powdered Ho₂Ge₂O₇ along with the fitted [P4₁2₁2(tetragonal) space group] and the difference between the observed and calculated fit.

Table 4.3 Wyckoff positions of Ho₂Ge₂O₇ obtained from refinement using space group = P4₁2₁2; Lattice parameters (a = b) = 6.80662(5) Å, c = 12.3803(1) Å; Obtained values of χ^2 =3.62, R_{wp}=15.7, R_p= 12.7 and R_e=8.26.

Atomic	Wyckoff	A	Thermal			
site	symbol	x (error)	y (error)	z (error)	parameter B (error)	
Но	8b	0.8778(1)	0.3524(1)	0.1351(6)	0.89470	
Ge	8b	0.9009(1)	0.1528(2)	0.6194(1)	0.01108	
01	4a	0.301(1)	0.301(1)	0.00000	0.00000	
O2	8b	0.0682(8)	0.9668(9)	0.6225(7)	0.62250	
03	8b	0.0582(9)	0.336(1)	0.5790(5)	0.90340	
04	8b	0.6897(9)	0.152(1)	0.5482(4)	0.07088	

The basic units of the asymmetric structure of Ho₂Ge₂O₇ consist of a pentagonal bipyramidal of Ho³⁺ atom coordinated with seven oxygen atoms, five in *ab* plane, and two obliques to the plane. The network contains a secluded $[Ge_2O_7]^{6-}$ unit in which Ge⁺⁴ ion is coordinated with four dis-equivalent oxygen atoms forming a tetrahedron with Ge-O₄-Ge unit in common with a bridging angle of 133.3(4) ° as shown in **Figure 4.2**.



Figure 4.2 3D-structure of Ho₂Ge₂O₇ depicting [Ge₂O₇]⁶⁻ unit in which Ge⁺⁴ ion is coordinated with four dis-equivalent oxygen with Ge-O4-Ge unit in common having a bridging angle of 133.3(4) °.

This is in good agreement with other lanthanide pyrogermanates, where the Ge-O-Ge angle lies between 130.0° to 136.0° . 3D structure of Ho₂Ge₂O₇ as obtained from the Visualization for electronic structural analysis (VESTA), using the structural information, is shown in **Figure 4.3 (a)**. The structure shows the 8 Ho atoms with 2 Ho atoms in each of the four layers

parallel to the crystallographic ab plane. Within a unit cell, the HoO_7 sublattice forms an infinite helical chain of magnetic rare-earth atom parallel to the 4₁ screw axis through the corner-sharing triangle. The pentagonal bipyramidal structure constituting the HoO_7 unit along with GeO₄ tetrahedra could be easily seen in **Figure 4.3 (b)**.



Figure 4.3 Crystal structure of Ho₂Ge₂O₇ showing (a) infinite helical chain of magnetic rare-earth atom in a single unit cell having 8 Ho³⁺ atoms with 2 Ho³⁺ atom in each of the four layers parallel to the crystallographic ab plane. (b) The pentagonal bipyramidal structure around the rare-earth ion is constituting the HoO₇ unit along with GeO₄ tetrahedra.

Within a helix, four Ho³⁺ ions are arranged in a pattern of common edge-sharing two triangles further corner joined with next Ho₄ plaquette. This arrangement is shown in the **Figure 4.4**. The average Ho-O bond length turns out to be 2.365 Å, which is linearly and conclusively correlated with increasing ionic radius of the rare-earth ion as the rare earth moves along the lanthanide period. The bond length and bond angle for this structure is listed in **Table 4.4**.



Figure 4.4 Position of magnetic ion within a helix of a unit cell of Ho₂Ge₂O₇. It shows four Ho³⁺ ions arranged in a pattern of common edge-sharing two triangles further corner joined with the next Ho₄ plaquette. Shared edges have been marked with a red line.

Bond	Bond Angle	Bond Length	Bond Length	Bond Angle
Length (Å)	(°)	(Å)	(Å)	(°)
Ho-O3	O4-Ge-O3	Ge-O1	O2-O1	O3-Ho-O4
2.494(7)	112.300(4)	1.780(4)	2.878(10)	76.200(3)
Ho-O2	O3-Ge-O2	Ge-O2	O3-O2	O3-Ho-O3
2.209(7)	97.500(4)	1.704(7)	2.542(11)	133.000(3)
Ho-O3	O2-Ge-O1	Ge-O3	O3-O4	O3-Ho-O4
2.565(7)	111.400(4)	1.718(8)	2.828(10)	166.400(3)
Ho-O2	O1-Ge-O4	Ge-O4	O4-O1	O3-Ho-O2
2.442(7)	98.600(4)	1.687(7)	2.630(7)	70.200(3)
Ho-O4	O1-Ge-O3		O4-O2	O3-Ho-O2
2.297(7)	112.000(4)		3.014(10)	89.400(3)
Ho-O3	O4-Ge-O2		01-03	(within HoO7
2.272(8)	125.500(4)		2.916(2)	polyhedron)
Ho-O4 2.278(7)			(within GeO4 tetrahedron)	

Table 4.4 Bond length and bond angles as obtained from the 3D structure of Ho₂Ge₂O₇ generated from VESTA.

4.2.2 Role of negative chemical pressure on the crystal structure of Ho₂Ge₂O₇

Followed by the structural analysis of Ho₂Ge₂O₇ using the correct space group and coordinates, we have tried to understand the spin dynamics of this particular spin frustrated system by applying negative chemical pressure effect in the Ho₂Ge₂O₇. **Figure 4.5** shows the Rietveld refinement of the room temperature HRXRD pattern of Ho₂Ti_{0.1}Ge_{1.9}O₇.



Figure 4.5 Room temperature refined high-resolution x-ray diffraction pattern of powdered Ho₂Ti_{0.1}Ge_{1.9}O₇ using the P4₁2₁2 space group.

Wyckoff positions of Ho₂Ti_{0.1}Ge_{1.9}O₇ obtained after refinement using space group = P4₁2₁2 are shown in **Table 4.5**. The bong length and bond angles are listed in the **Table 4.6**. Considering the higher ionic radii of Ti⁴⁺, its inclusion in the Ho₂Ge₂O₇, causes a lattice

volume expansion. The obtained lattice volume for $Ho_2Ge_2O_7$ is 573.584 ± 0.008 Å³, and that

for Ho₂Ti_{0.1}Ge_{1.9}O₇ is 573.990 ± 0.015 Å³.

Table 4.5 Wyckoff positions of Ho₂Ti_{0.1}Ge_{1.9}O₇ obtained after refinement using space group = P4₁2₁2; Lattice parameters (a = b) = 6.8080(1) Å, c = 12.3840(2) Å; Obtained values of χ^2 =15.64, R_{wp}=35.2, R_p= 23.9 and R_e=8.91.

Atomic	Wyckoff		Thermal		
Site	symbol	x (error)	y (error)	z (error)	parameter B (error)
Но	8b	0.8792(1)	0.3518(4)	0.1350(3)	0.7006(8)
Ge	8b	0.9016(4)	0.1546(2)	0.6187(6)	0.0110(8)
01	4a	0.2989(7)	0.2989(7)	0.00000	0.0237(4)
02	8b	0.0600(5)	0.9587(4)	0.6242(4)	0.0091(3)
O3	8b	0.0495(9)	0.3255(6)	0.5810(4)	0.0106(6)
04	8b	0.6858(4)	0.1520(4)	0.5447(6)	0.0708(8)

Table 4.6 Bond	length and	l bond	angles of	Ho2Ti0.1Ge1.9	O7 as o	btained	from	using	the
tools VESTA.									

Bond Length	Bond Angle	Bond Length	Bond Length	Bond Angle
(Å)	(°)	(Å)	(Å)	(°)
Ho-O4	O4-Ge-O3	Ge-O1	O2-O1	O3-Ho-O4
2.2520(14)	112.6(9)	1.797(10)	2.88(3)	86.0(7)
Ho-O2	O3-Ge-O2	Ge-O2	O3-O2	O3-Ho-O3
2.5290(14)	100.3(8)	1.715(15)	2.56(3)	142.0(3)
Ho-O3	O2-Ge-O1	Ge-O3	O3-O4	O3-Ho-O4
2.5440(16)	110.2(10)	1.613(18)	2.78(3)	86.0(7)
Ho-O2	O1-Ge-O4	Ge-O4	O4-O1	O3-Ho-O2
2.1600(14)	98.7(8)	1.730(14)	2.67(16)	77.7(7)
Ho-O3	O1-Ge-O3		O4-O2	O3-Ho-O2
2.3400(2)	112.3(9)		3.03(3)	71.8(5)
Ho-O4 2.2620(14)	O4-Ge-O2 123.1(9)		O1-O3 2.83(3)	(within HoO ₇ polyhedron)
Ho-O3 2.582(16)			(within GeO4 tetrahedron)	

The expansion is also confirmed through the shift of (hkl) peaks towards a lower angle with Ti^{4+} substitution, as could be seen from a few of the prominent diffraction peaks shown in **Figure 4.6**.



Figure 4.6 Room temperature high-resolution x-ray diffraction pattern for (a) Ho₂Ge₂O₇ and (b) Ho₂Ti_{0.1}Ge_{1.9}O₇ indicating the shift of (201), (004), (212), (105), and (321) peaks towards lower angle side.

4.3 Magnetic properties of Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇

In this section we will be describing the magnetic properties of Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇ through the temperature dependent magnetization, magnetic field dependent magnetization and temperature dependent ac-susceptibility measurements. These will be subsequently discussed in section in 4.3.1, 4.3.2 and 4.3.3, respectively.

4.3.1 Temperature dependent magnetization.

Figure 4.7 (a) & Figure 4.7 (b) present the dc-susceptibility (M-T) of both Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇ at an applied field of 100 & 1000 Oe, respectively. The sample was cooled till 2 K, and zero-field cooled (ZFC) data were collected in the range of 2-200 K under the application of the magnetic field (H= 100 & 1000 Oe) during the warming sequence. Field cooled (FC) data is also collected in the presence of that very same field and recording the data in warming sequence.



Figure 4.7 Temperature dependence of dc-susceptibility curve of Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇ in the range of (2-200) K for (a) H = 100 Oe & (b) H = 1000 Oe. Inset contains the corresponding inverse susceptibility curve to obtain θ_{CW} fitted through Curie-Weiss law in the linear temperature range of (25-50) K with further extrapolating the χ^{-1} vs. T plot.

A gradual increase in magnetization without any abrupt characteristic implies no evidence of any magnetic phase transition down to the 2 K, which indicates a slow spin alignment process with decreasing temperature. FC and ZFC dc-susceptibility data entirely overlap which excludes the possibility of the formation of a spin-glass state at a lower temperature [45]. This response gives the clear inference of partially ordered, ordered, or ice like ground state, indicating that holmium germanates behave as a soft magnetic material at a lower temperature. Inverse susceptibility curve as shown in the inset of **Figure 4.7 (a) & Figure 4.7 (b)** shows a linear response against temperature above 70 K extending till 200 K. Below 70 K; χ^{-1} shows deviation from ideal Curie-Weiss law might be due to the formation of shortrange magnetic correlations due to the increase in spin flipping time at lower temperatures. However, no interactions between spins are established until 2 K, which could lead to longrange ordering. Identical behavior is manifested in Tb₂Ti₂O₇ and Tb₂Ge₂O₇ systems, where the compound is in a typical paramagnetic state with the appearance of short-range spin correlations at a lower temperature [19].

Inverse susceptibility vs. temperature plot has been fitted using equation 3.1 to obtain the Curie-Weiss temperature [90]. In the similar fashion, three different range of temperature, (20-45), (25-50), and (30-55) K, has been used. For the fitting temperature range (20-45) K, θ cw increases from -1.44 K to -1.28 K for H = 100 Oe, and from -1.35 K to -1.32 K for H = 1000 Oe. For the temperature range of (25-50) K, θ cw increases from -2.04 K to -1.84 K for H = 100 Oe, and from -1.95 K to -1.84 K for H = 1000 Oe. Similarly, for temperature range (30-55), θ cw increases from -2.77 K to -2.64 K (H = 100 Oe) and from -2.85 K to -2.50 K (H = 1000 Oe). The values are summarized in **Table 4.7**. Similar trend in θ cw was obtained with increase in negative chemical pressure i.e., for sample having composition of

Ho₂Ti_{0.1}Ge_{1.9}O₇. The fitting range of (25-50) K has been used for further discussion. The value of θ_{cw} is -2.04 K for Ho₂Ge₂O₇. θ_{cw} moves towards a lesser negative value of -1.89 K for Ho₂Ti_{0.1}Ge_{1.9}O₇ (H=100 Oe) suggesting a decrement in antiferromagnetic exchange interaction between Ho⁺³ magnetic moments with increase in negative chemical pressure. Saez-Puche *et al.* reported the Curie-Weiss Temperature of powdered tetragonal Ho₂Ge₂O₇ to be -4.7 K [91]. The value of obtained Curie-Weiss temperature for H =1000 Oe also evolves, in the same manner, being -1.95 K for Ho₂Ge₂O₇ and -1.84 K for Ho₂Ti_{0.1}Ge_{1.9}O₇.

Table 4.7 Curie Weiss temperature for Ho₂Ge₂O₇ & Ho₂Ti_{0.1}Ge_{1.9}O₇ obtained from Curie Weiss fitting at 100 and 1000 Oe. The values in bracket represent the error.

Compound	θ _{cw} (K); H=100 Oe	θ _{cw} (K); H=1000 Oe	θ _{cw} (K); H=100 Oe	θ _{cw} (K); H=1000 Oe	θ _{cw} (K); H=100 Oe	θ _{cw} (K); H=1000 Oe
	Fitting range = (20-45)K		Fitting range = (25-50)K		Fitting range = (30-55)K	
Ho ₂ Ge ₂ O ₇	-1.44(0.48)	-1.35(0.46)	-2.04(0.54)	-1.95(0.54)	-2.77(0.54)	-2.85 (0.63)
$Ho_2Ti_{0.1}Ge_{1.9}O_7$	-1.28(0.44)	-1.32(0.58)	-1.84(0.61)	-1.84(0.58)	-2.64(0.43)	-2.50 (0.74)

4.3.2 Field dependent magnetization

Isothermal magnetization (T=2 K) of Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇ as a function of the magnetic field (0-70 kOe) is shown in **Figure 4.8.** Magnetization increases sharply with field till 5 kOe, which proceeds towards saturation with a very gradual increment till the applied field of 70 kOe. The value of maximum magnetization (M_{max}) of Ho₂Ge₂O₇ within the applied field range is 84.14 emu/gm, i.e., 4.3 μ b/Ho³⁺ ion, indicating ferromagnetic alignment of Ho³⁺ spins at 2 K, and is little less than half the expected value of free Ho³⁺ ion moment (μ _{sat} = 10 μ B) due to the angular averaging of the powdered sample.



Figure 4.8 Magnetic field dependence isothermal magnetization of Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇ at T = 2 K for H ranging from -70 to 70 kOe.

Interestingly it also follows the similar M-H behavior exhibited for cubic pyrochlore say, $Ho_2Ti_2O_7$ spin ice [27]. The magnetic moment of $Dy_2Ge_2O_7$ (tetragonal) and $Dy_2Ti_2O_7$ (cubic) has already been compared by Ke. X *et al.*, where $Dy_2Ge_2O_7$ has a slightly lower value of saturation magnetization in comparison to $Dy_2Ti_2O_7$ [92]. In $Ho_2Ge_2O_7$ and $Ho_2Ti_{0.1}Ge_{1.9}O_7$, similar behavior could be associated with the strong crystal electric field's anisotropy resulting in a local "Ising behavior" manifested through Ho-Ho coordination in a single Ho₄ plaquette.

Lattice expansion with negative chemical pressure reduces both dipolar (D_{nn}) and exchange interactions (J_{nn}). Even though both interactions decrease, the evolution of θ_{cw} suggests that reduction in exchange interaction dominates over the reduction in dipolar interaction. It could be justified since the dipolar energy D_{nn} is proportional to $1/r_{nn}^{3}$, and the exchange energy is simply electrostatic interaction varying as 1/rnn. The separation of the moment (r_{nn}) is of the order of 3.5 Å, which justifies more reduction in J_{nn} in comparison to that of the D_{nn}. T. Sloter *et al.* had established the positive chemical pressure effect in the Dy₂Ge₂. _xSi_xO₇ system where both D_{nn} and J_{nn} increases, but an increase of AFM exchange interaction dominates over that of FM dipolar interaction [21]. Hence, in Ho₂Ti_{0.1}Ge_{1.9}O₇, this is possibly accredited to the altered sequence of chemical pressure in an entirely different lattice. It can also be possibly associated with the different electronic ground states (singlet in Ho₂Ge₂O₇ and doublet in Ho₂Ti₂O₇) due to differences in local crystal field symmetry of Ho^{3+} , which is D_{5h} (Ho₂Ge₂O₇) and D_{3d} (Ho₂Ti₂O₇). In Ho₂Ti₂O₇ and Dy₂Ti₂O₇ spin ice, FM dipolar interaction D_{nn} dominates over nearest-neighbor AFM exchange interaction J_{nn} (magnetic ground state); the effective nearest neighbor interaction being given by $J_{eff}S_i^z S_j^z$ where $J_{eff} = m^2(5D_{nn}-J_{nn})/3$ (m=magnetic moment of Ho³⁺); J_{nn} and D_{nn} being the exchange and dipolar coupling constants; $S_i^z = \sigma_i \vec{z}_i$; $\sigma_i = \pm 1$ and \vec{z}_i are four in equivalent unit vectors pointing from one tetrahedral sublattice to other) [49]. For this J_{eff} , the spins are highly frustrated and end up with a "2 in-2 out" configuration. But for Ho₂Ge₂O₇, the obtained value of θ_{cw} (-1.95 K) indicates that there should be long-range AFM ordering in such systems.

4.3.3 Temperature dependent ac-susceptibility

The response of ac-susceptibility with frequency, temperature and applied dc field bias gives better insight of low temperature spin dynamics and how interactions had been tuned in for such a complex spin ice Ho₂Ge₂O₇ systems. **Figure 4.9** shows the temperature dependence of the real part of the ac-susceptibility of Ho₂Ge₂O₇ down up to 2 K in the absence of dc magnetic field at applied characteristic frequencies of 50, 100, 300, 500, 700, and 900 Hz.



Figure 4.9 Temperature dependence of real part of the ac-susceptibility of Ho₂Ge₂O₇ without any application of magnetic field at the characteristic frequency of 50, 100, 300, 500, 700, and 900 Hz.

Without any applied magnetic field, Ho₂Ge₂O₇ shows a typical paramagnetic behavior with a monotonous increase in susceptibility having no signature of magnetic ordering down to 2 K. **Figure 4.10** shows ac-susceptibility plot for Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇ under the application of external magnetic field (0.5, 1.0, 5.0, 7.5, 10, 20, and 50 kOe) for an applied frequency of 500 Hz. Contrary to the dc-susceptibility, which rises uniformly with decrease

in temperature followed by the sharp increment below 3 K, ac-susceptibility reveals two spin relaxation features on the application of an external magnetic field. Lower temperature relaxation feature starts appearing at a minimum applied field of 5 kOe centered at 3.2 K, which shifts towards a higher temperature side with an increase in the magnetic field, firmly defying its AFM origin.



Figure 4.10 Real part of ac-susceptibility of for (a) Ho₂Ge₂O₇ and (b) Ho₂Ti_{0.1}Ge_{1.9}O₇ as a function of temperature (2-40) K at an applied magnetic field of 0.5, 1.0, 5.0, 7.5, 10, 20 and 50 kOe (applied frequency = 500 Hz).

Had this feature been of AFM origin, an increase in the magnetic field's magnitude would have suppressed it towards the lower temperature side. This shifting with field enhancement suggests it to be of ferromagnetic origin and is most likely associated with the emergence of local short-range spin correlation [19]. Specific heat measurement on single-crystal Ho₂Ge₂O₇ had already established the development of short-range correlations above the 3d ordering temperature, which is of ferromagnetic nature, while long-range ordering reveals significant antiferromagnetic nature [28]. Such a field-driven magnetic transition in Ho₂Ge₂O₇ can be characterized as a metamagnetic transition (MMT). It results due to the reorientation of the spins at the anisotropic magnetic site in response to the applied field. Similar short-range local ferromagnetic spin correlations are also manifested in Tb₂Ge₂O₇, whereas Dy₂Ge₂O₇ exhibits long-range antiferromagnetic ordering at 2.3 K [92]. Relative lowering in peak intensity corresponding to this feature is manifested. With an increase in the magnetic field to 7.5 kOe, the second relaxation peak emerges towards the higher temperature side at 14 K (single ion freezing; T_s), which is a characteristic thermally induced peak due to initiation in freezing of single ion spin fluctuations [28]. Both the peaks get almost flattened at an applied field of 50 kOe.

The shift of T_s towards the higher temperature side with an increase in the applied magnetic field is shown in **Figure 4.11**. In Ho₂Ti₂O₇, neutron spin-echo (NSE) studies establish thermally driven single ion spin relaxation process for T > 15 K, and for T < 15 K, spin dynamics is cumbersome, attributed to quantum tunneling between possible spin states [36]. The lower temperature, quantum spin relaxation feature is also manifested in Dy₂Ti₂O₇ and Dy₂Sn₂O₇. However, in Ho₂Ge₂O₇ spin relaxation feature corresponding to $T \approx 3$ K exhibits a shift upon the application of chemical pressure in the Ho₂Ge₂O₇. Coming to **Figure 4.10** where ac-susceptibility curve of Ho₂Ge₂O₇ has been plotted along with its negative chemical pressure derivative Ho₂Ti_{0.1}Ge_{1.9}O₇.



Figure 4.11 The variation in the single ion freezing temperature with a magnetic field for (a) Ho₂Ge₂O₇ and (b) Ho₂Ti_{0.1}Ge_{1.9}O₇.

With a decrease in chemical pressure, the feature corresponding to ferromagnetic correlations between spins shifts towards lower temperature, as shown in **Figure 4.12.** As discussed earlier, lattice volume expansion can weaken both the dipolar and exchange interaction due to negative chemical pressure. The spin ice system's dipolar interactions are of ferromagnetic origin, and its reduction explains the shift in T_s towards lower temperature with negative chemical pressure. It indicates the absence of quantum spin relaxation phenomenon in Ho₂Ge₂O₇ for lower T regime (\approx 3 K) probably due to the difference in crystal field environment associated with different structures compared to the cubic pyrochlore Ho₂Ti₂O₇.



Figure 4.12 The ac-susceptibility plot of Ho₂Ge₂O₇ and Ho₂Ti_{0.1}Ge_{1.9}O₇at an applied magnetic field of (a) H = 5 kOe (b) H = 7.5 kOe (c) H = 10 kOe (d) H = 20 kOe (e) H = 50 kOe.

 $Dy_2Ge_2O_7$ (tetragonal) isostructural to $Ho_2Ge_2O_7$ also marks the absence of quantum spin relaxation since both spin relaxation feature exhibits frequency dependence with temperature [92]. The interplay of dipolar and exchange interactions between Ho^{3+} atoms (across *ab* plane and along *c* axis) in the complex spin structure of $Ho_2Ge_2O_7$ can tune the spin dynamics through chemical pressure. The result demonstrates that spin freezing of $Ho_2Ge_2O_7$ is associated both with thermally activated single ion spin relaxation at 14 K and with collective relaxation of correlated clusters of spins of ferromagnetic origin at 3.2 K.

4.4 Conclusions

Detailed structural analysis of Ho₂Ge₂O₇ has been presented using the correct Wyckoff site for the P4₁2₁2 space group. Magnetic response of powdered Ho₂Ge₂O₇ (tetragonal) closely resemble that of cubic Ho₂Ti₂O₇. In pyrogermanates, the low-temperature short-range spin correlation is of ferromagnetic origin, unlike that of quantum tunneling phenomenon (spin ice configuration) in Ho₂Ti₂O₇ which is due to the difference in crystal field potential around the rare-earth ion because of dissimilar oxygen environment. The ac- susceptibility measurements suggest this system to be a classical spin ice material. With the reduction in chemical pressure in the Ho₂Ge₂O₇, the decrement of antiferromagnetic exchange interaction (J_{nn}) to be dominant over the decrease in ferromagnetic dipolar interaction (D_{nn}) has been established. It has been shown that these interactions can further be tuned through the structural distortions. Single ion spin freezing around 14 K is prevalent this spin ice materials. After studying the spin relaxation mechanism and magnetic properties of these pyrochlore oxides, we moved on to perform the electronic studies of these systems to look for its possibility in materiality applications. The same is discussed is next chapter.