

CHAPTER 1

Introduction

**Overview of Some Main Class of
Nitrogen, Oxygen and Sulphur
Containing Heterocyclic Compounds**

Overview of Some Main Class of Nitrogen, Oxygen and Sulphur Containing Heterocyclic Compounds

Heterocyclic compounds are a class of cyclic organic compounds in which one or more of the carbon atoms inside the backbone of the molecule is changed apart from carbon. The most commonly found heteroatoms in heterocyclic compounds are nitrogen, oxygen and sulphur. Heterocyclic compounds are divided into different categories on the basis of total number of atoms in the cyclic structure they have like three, four, five, six, seven and eight membered and also fused types of heterocyclic compounds are known. Out of them nitrogen, oxygen and sulphur containing five, six membered and fused heterocyclic compounds maintained the curiosity of researchers throughout the decades of historical development of organic synthesis.

Five Membered Heterocyclic Compounds: Pyrrole, furan and thiophene are the important single heteroatom containing five membered heterocyclic compounds. The common five membered heterocyclic compounds having more than one heteroatoms are isoxazole, pyrazole, imidazole, azole, thiazole, thiadiazole, oxadiazole, triazene, etc. Heterocyclic compounds which are partially reduced are often referred to as dihydro or tetrahydro derivatives of the parent unsaturated compound (**Figure 1.1**).

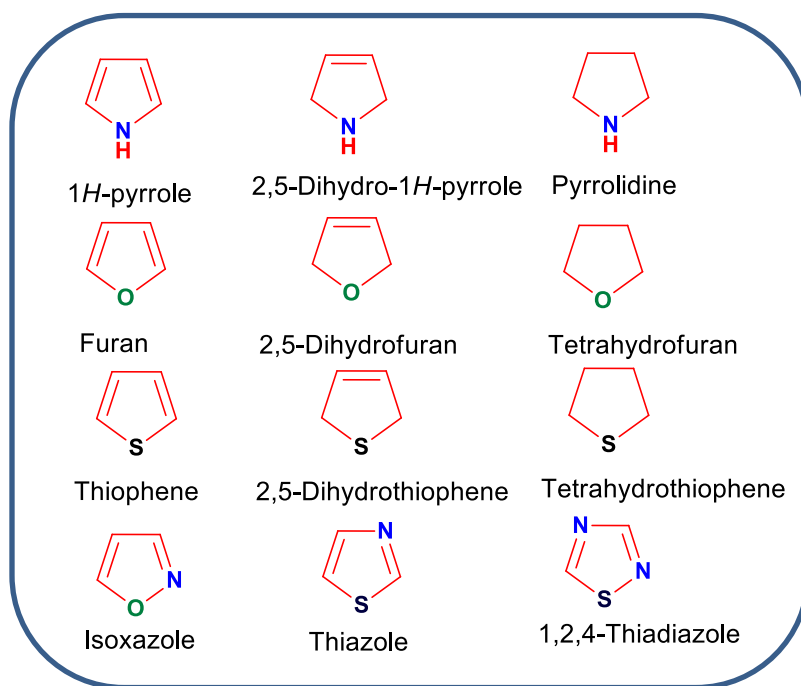


Figure 1.1: Some important five membered heterocyclic compounds.

Six Membered Heterocyclic Compounds: Pyridine, pyran and thiopyran are some common six membered heterocyclic compounds containing single heteroatom in the ring. The six membered heterocycles which comprises more than one heteroatoms are pyrazine, dioxane, dithine, oxazine, thiazine etc. Thiadiazine is a heterocyclic compound having two nitrogen atoms and one sulfur atom in six-membered ring skeleton and triazine also contains three nitrogen atoms in a six member ring (**Figure 1.2**).

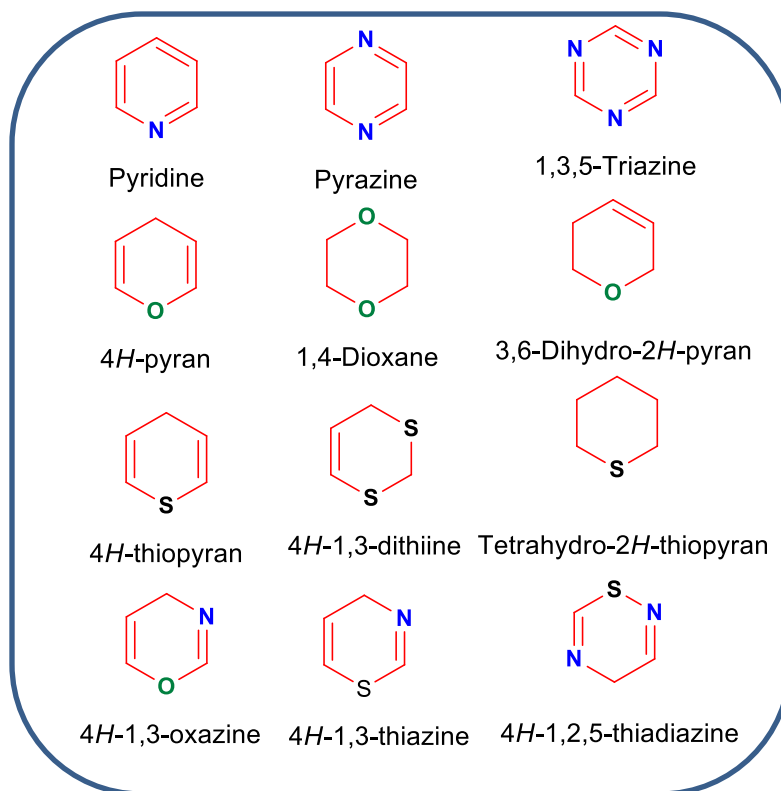


Figure 1.2: Some important six membered heterocyclic compounds.

Fused Heterocyclic Compounds: Fused heterocyclic ring structures are obtained by the fusion of one ring with other rings. Quinoline, 1H-indole, coumarin and benzothiophene are some important single heteroatom containing fused heterocyclic compounds. The common fused heterocyclic ring contains more than one heteroatoms such as imidazo[1,2-a]pyridine, benzothiazole, benzoxathiole and furopyran etc. Pyrazolooxazole is a heterocyclic compound holding one oxygen atom and three nitrogen atoms as share of the fused heterocyclic compound (**Figure 1.3**).

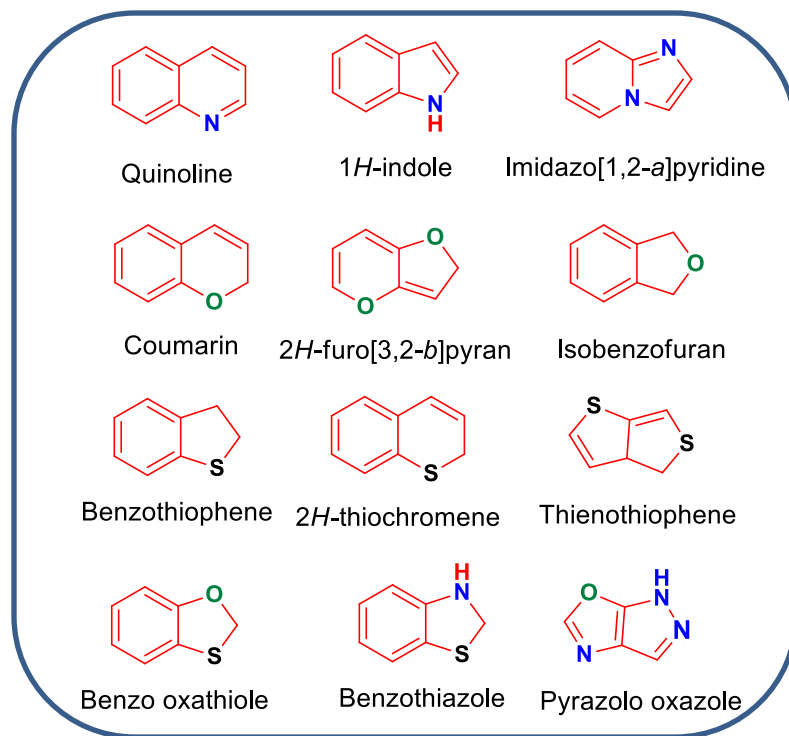


Figure 1.3: Some important fused heterocyclic compounds.

The chemical substances which comprise at least one heterocyclic moiety make them the most important class of compounds in organic chemistry. Heterocyclic compounds have attracted the attention of scientists over the years due to its vitality in human life and abundance in nature (Nagaraj et al. 2008) due to the fact that they are important structural subunits present in many herbal merchandise consisting of vitamins, hormones, antibiotics and pigments (Ju et al. 2006, Yadav et al. 2011). Thus, these derivatives have attracted considerable attention in the designing of biologically active molecules. Modern society depends on man-made heterocycles for many purposes together with drugs, pesticides,

dyes, plastics, cosmetics, solvents, antioxidants and vulcanization accelerators (Li et al. 2011, Martins et al. 2004) (Figure 1.4).

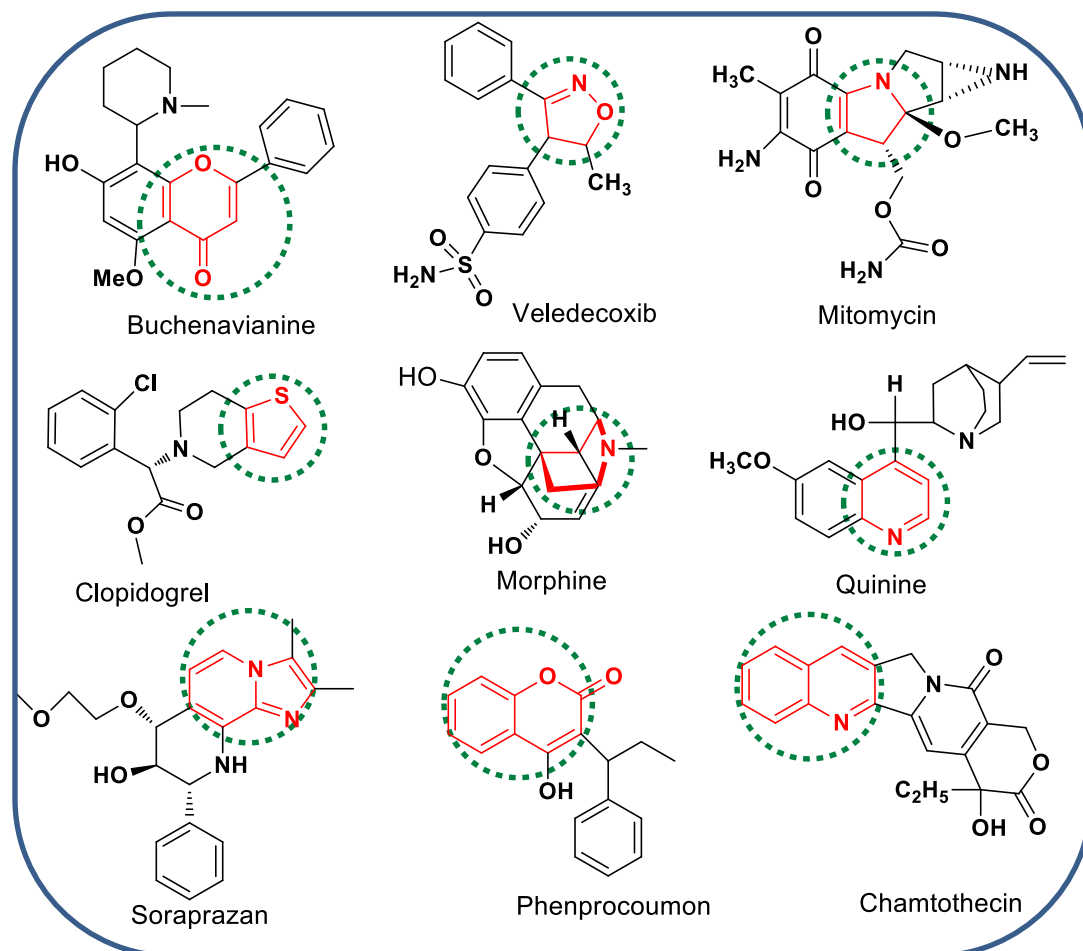


Figure 1.4: Some example of natural products and drugs containing different heterocyclic moieties.

Synthesis of heterocyclic skeleton with biologically ubiquitous functionalities is a lucrative aim for chemists. The five, six membered, fused heterocycles and related compounds have been the subject of major attraction for the researchers in the area of

synthetic organic chemistry. Therefore, chemists have been involved in extensive efforts to produce these heterocyclic compounds by rising new and effective synthetic transformation. Our main focus is on the study of 1,2,4-thiadiazole, coumarin, imidazo[1,2-*a*]pyridine and 4*H*-pyran scaffold.

1.1 Thiadiazole

Thiadiazoles are very important class of nitrogen and sulphur containing heterocyclic compounds. 1,2,3-thiadiazole, 1,2,5-thiadiazole, 1,2,4-thiadiazole and 1,3,4-thiadiazole are four isomeric forms of thiadiazole found in nature (Siddiqui et al. 2009). Out of them, we have focused on 1,2,4-thiadiazole. 1,2,4-Thiadiazole core structure are of great interest mainly because of their various biological activities and associated clinical applications (Iizawa et al. 1993, Donkor et al. 2000, Gurjar et al. 2014). They have diverse applications in different fields. A large number of synthetic 1,2,4-thiadiazole derivatives show wide range of biological activities such as thiadiazole KC 12291 which displayed the foremost proof of its cardioprotective activity (Hartmann et al. 1998), 6-(1,2,4-thiadiazol-5-yl)-3-amino pyridazine derivatives are recognized as unique angiogenesis inhibitors (Bongartz et al. 2002) and monocyclic 1,2,4-thiadiazoles have been found as useful insecticidal, herbicidal and fungicidal agents. Additionally, the properties of 1,2,4-thiadiazole as thioltrapping representatives has been reviewed recently (Tam et al. 2005) (Figure 1.5).

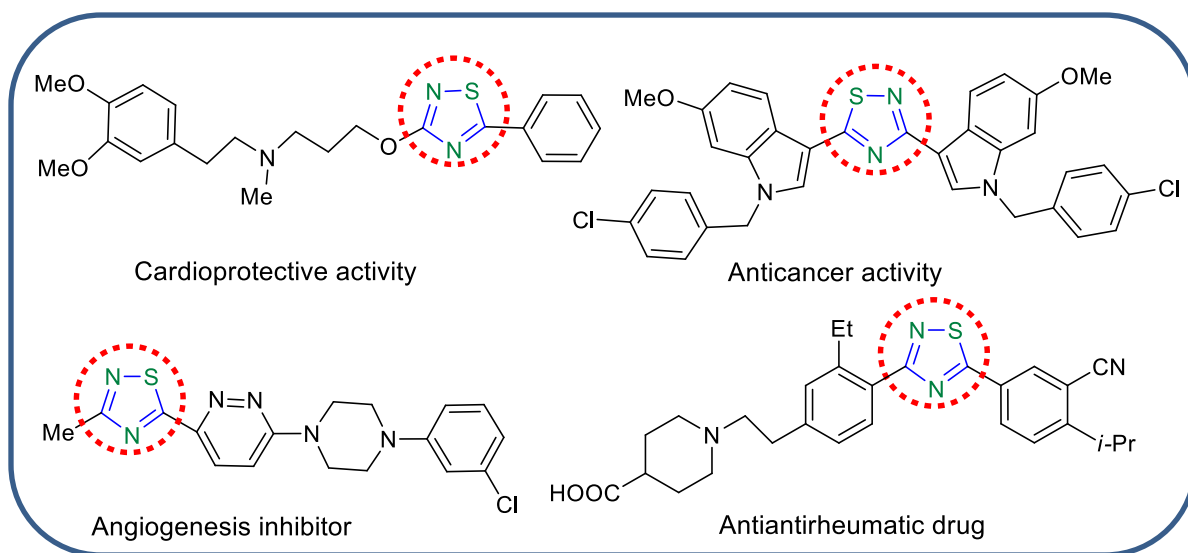
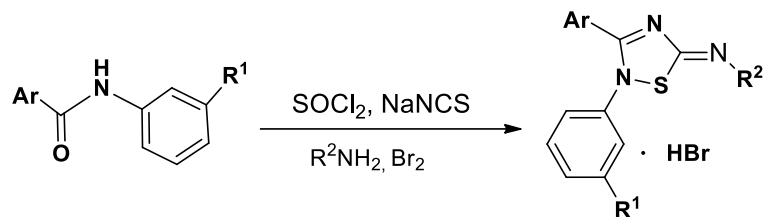


Figure 1.5: Some biological active compounds containing thiadiazole scaffold.

In this context, enormous efforts have been dedicated by chemists to establish many concise synthetic paths accessing thiadiazole framework. Synthesis of thiadiazole has been sub categorized in cyclization, dimerization and condensation reactions.

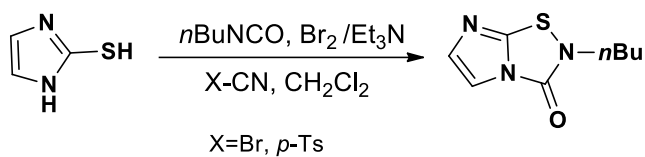
1.1.1 Synthesis of 1,2,4-Thiadiazoles by Cyclization

1,2,4-Thiadiazoles have been synthesized by cyclization of different starting materials in the presence of catalysts such as from benzamides in the presence of SOCl_2 , NaNCS , RNH_2 , Br_2 (Goblyos et al. 2005), from 2-mercaptoimidazole and *n*-butyl isocyanate via oxidative ring closure with Br_2 and triethylamine (Leung et al. 2005), from imidoylthioureas catalyzed by copper (Kim et al. 2014), imidoylthioureas mediated by phenyliodine(III) bis(trifluoroacetate) (PIFA) (Mariappan et al. 2016) (**Scheme 1.1**).

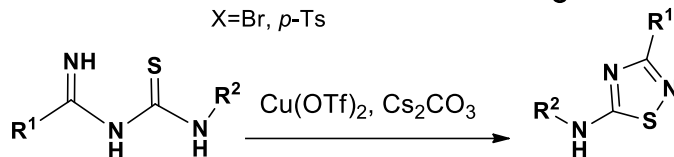


$R^1 = \text{H, Me}$

$R^2 = \text{Me, Et, Pr, Bu, 3-OH-Pr, } i\text{Pr, cPentyl, Ph, Bn}$



$X = \text{Br, } p\text{-Ts}$



$R^1 = \text{Ph, Py, Me, SBn}$

$R^2 = \text{Ph, Ar, Bn, Cyclopentyl, Bz}$



$R^1 = \text{Ph, Ar, Py, 1,2-pyrazol, MeS}$

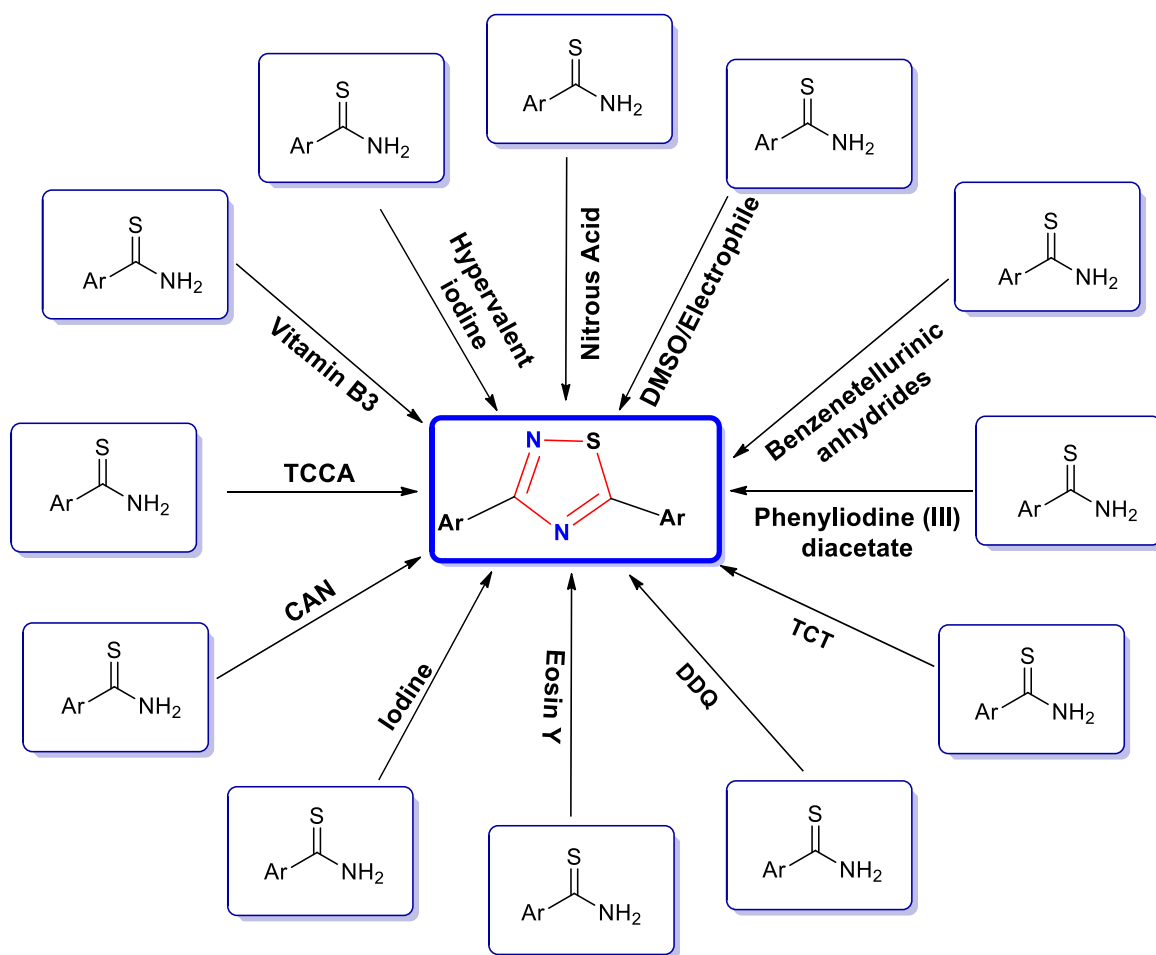
$R^2 = \text{Ph, Ar, Bn, Cyclohexyl}$

Scheme 1.1: Synthesis of 1,2,4-thiadiazoles by cyclization.

1.1.2 Synthesis of 1,2,4-Thiadiazoles by Dimerization of Thioamides

1,2,4-Thiadiazoles are also synthesized by dimerization of thioamides by using different oxidizing agents such as nitrous acid (Cronyn et al. 1952), DMSO/electrophile (Takikawa et al. 1985), benzene tellurinic mixed anhydrides

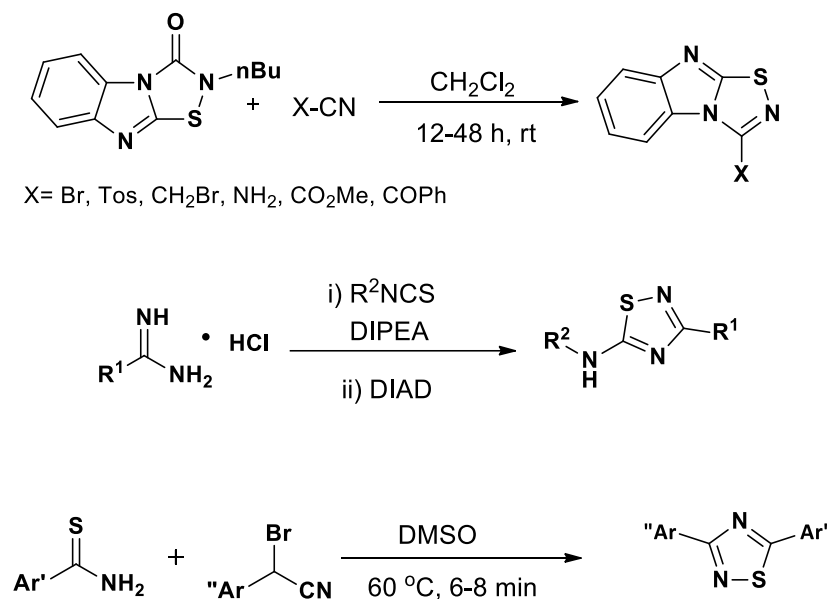
(Fukumoto et al. 1990), phenyliodine (III) diacetate (Yan et al. 2003), 2,4,6-trichloro-1,3,5-triazine (TCT) (Khosropour et al. 2010), 2,3-dichloro-5,6-dicyanobenzoquinone (Cheng et al. 2012), eosin Y in visible light (Srivastava et al. 2013), iodine (Zhao et al. 2014), ceric ammonium nitrate (Vanajatha et al. 2016), trichloroisocyanuric acid (TCCA) (Bose et al. 2017), organocatalytic approach by using vitamin B₃ (Putta et al. 2018), pseudo cyclic hypervalent iodine (Quarban et al. 2019) (**Scheme 1.2**).



Scheme 1.2: Synthesis of 1,2,4-thiadiazoles by dimerization of thiobenzamides.

1.1.3 Synthesis of 1,2,4-Thiadiazoles through Condensation Reactions

Another synthetic path to produce the useful 1,2,4-thiadiazole framework is through condensation or exchange reactions by using various precursors such as 1,2,4-thiadiazol-3(2*H*)one derivatives and substituted nitriles (e.g. cyanogen bromide or *p*-toluenesulfonyl cyanide) (Leung et al. 2005), from isothiocyanates, amidine hydrochlorides and *N,N*-diisopropylethylamine (DIPEA) followed by the addition of diisopropylazodicarboxylate (Wu et al. 2008), from benzothioamides with arylacetonitrile (Boeini et al. 2011) (**Scheme 1.3**).



Scheme 1.3: Synthesis of 1,2,4- thiadiazoles by condensation.

1.2 Coumarin

Coumarin compounds are important naturally occurring and synthetic oxygen containing heterocyclic compounds and it has benzopyrone framework. In 1820, Vogel isolated coumarin from tonka beans also known as coumarou in French (Gleye et al. 2003). This special benzopyrone structural moiety permits its derivatives to interact with a variety of enzymes and receptors in organisms through weak bond interactions and due to this property they are widely used in medicinal chemistry. Coumarins play an important role in numerous natural processes like plant physiology and also elaborate in the actions of plant growth hormones, growth regulators, control of respiration and photosynthesis (Vekariya et al. 2014). In the field of supramolecular chemistry which is important in pharmaceutical

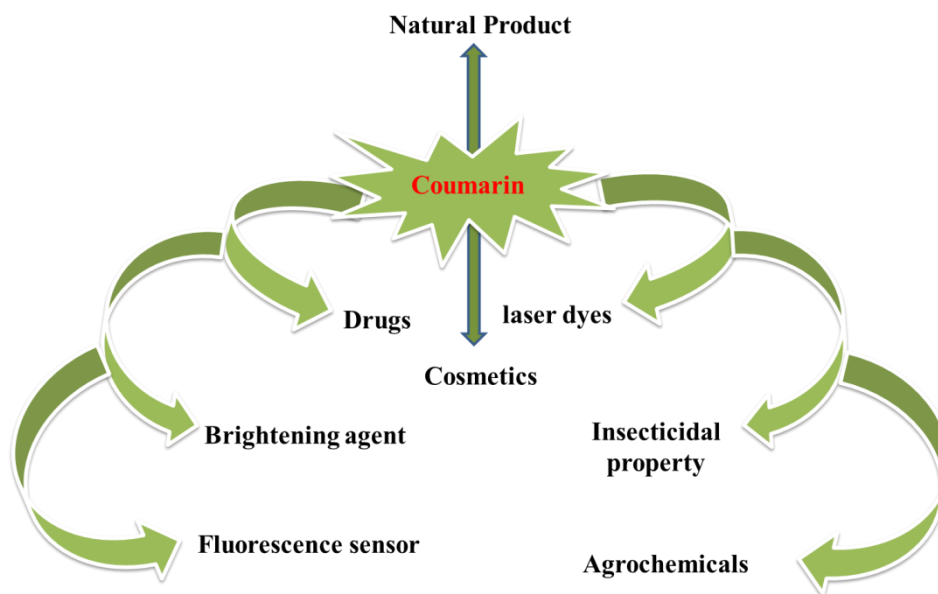


Figure 1.6: Applications of coumarin in different fields.

science, coumarin containing supramolecular medicine emerged as a new class of drugs in recent years. Coumarin has been investigated for a wide range of properties in different fields such as in fragrance & perfume, additive in food & cosmetics, laser dyes, agrochemicals, cosmetics, brightening agents, insecticides, fluorescence sensor and in drugs (Brahmachari et al. 2015) (**Figure 1.6**).

Figure 1.7 shows some coumarin containing biologically active natural products and drugs like Umbelliferone is used as sunscreen, Dicoumarol as anticoagulant drug, Seselin plays an important role as a metabolite, Calanolide A shows anti-tuberculosis activities, Chlorobiosin use as antibacterial and Lamellarin D shows cytotoxic activity against tumor cell lines.

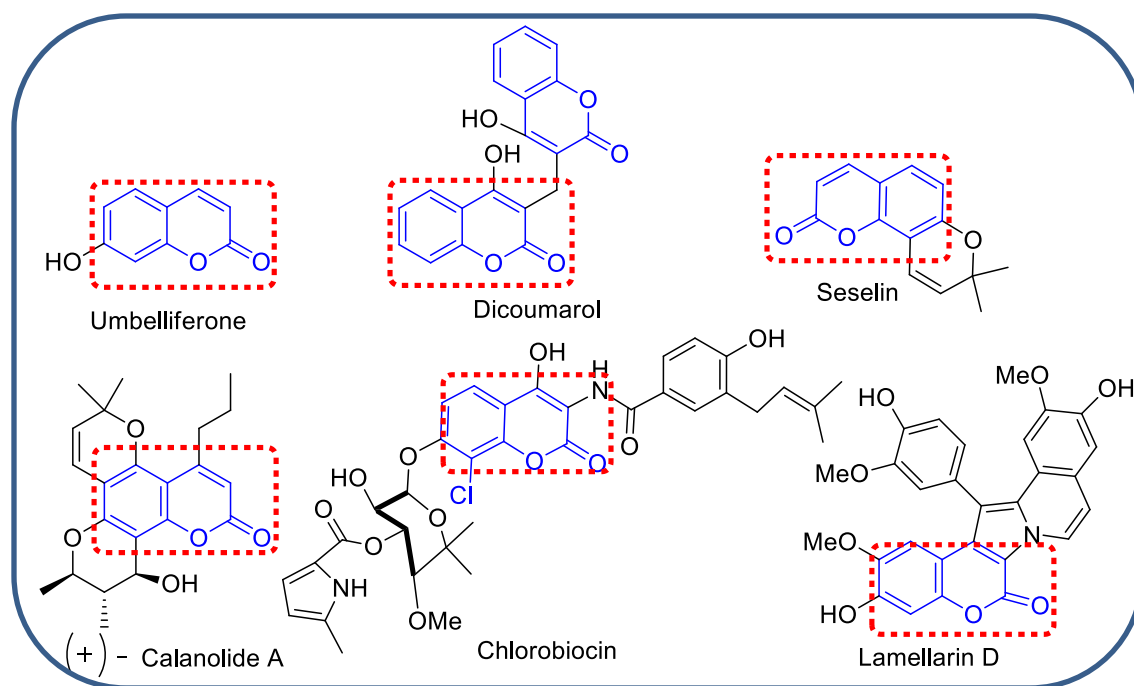
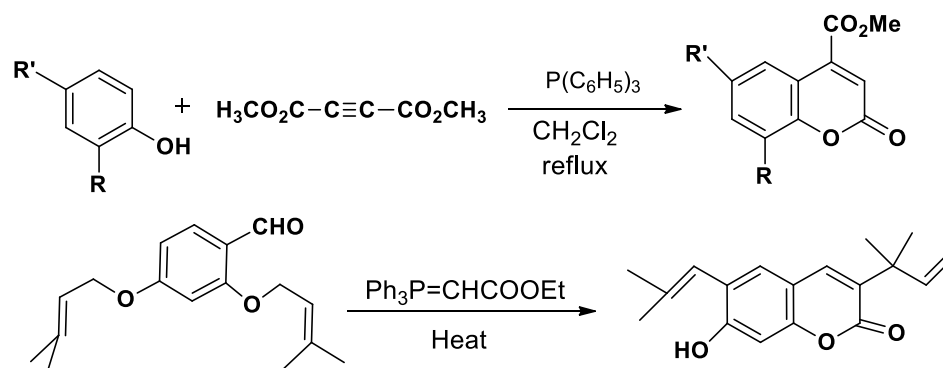


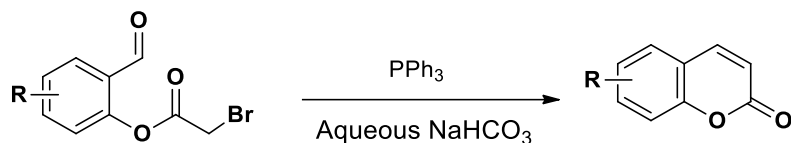
Figure 1.7: Coumarin containing biologically active compounds.

In view of the numerous applications in different fields, numerous natural coumarins are isolated from fungi, bacteria, plants and as well as obtained from chemical synthesis (Murray et al. 1991, Stefanachi et al. 2018). Synthesis of coumarin is always the subject of interest for the organic chemists. They have developed different methodologies for synthesis of this biologically important moiety. In literature, lots of methods have been described for the synthesis of coumarins by using different starting materials with different catalysts like Wittig reaction, Perkin reaction, Baylis-Hillmann reaction, Pechmann condensation and Knoevenagel condensation.

1.2.1 Wittig Reaction

Wittig reaction is tremendously used for C–C bond formation in the synthesis of natural products. Coumarins have been synthesized through Wittig reaction under different reaction conditions (Yavari et al. 1998, Patre et al. 2009, Belavagi et al. 2014) (**Scheme 1.4**).

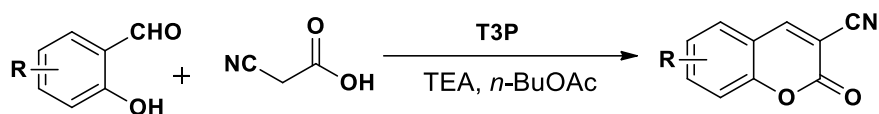




Scheme 1.4: Synthesis of coumarin by Wittig reaction.

1.2.2 Perkin Reaction

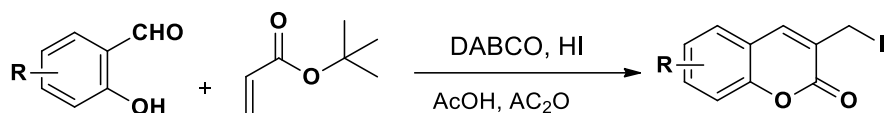
Perkin reaction is another method to synthesize coumarins. Augustine and co-workers described the synthesis of coumarins from salicylaldehyde and cyano acetic acid mediated by propylphosphonic anhydride (T3P) by Perkin reaction (Augustine et al. 2012) (**Scheme 1.5**).



Scheme 1.5: Synthesis of coumarin by Perkin reaction.

1.2.3 Baylis-Hilman Reaction

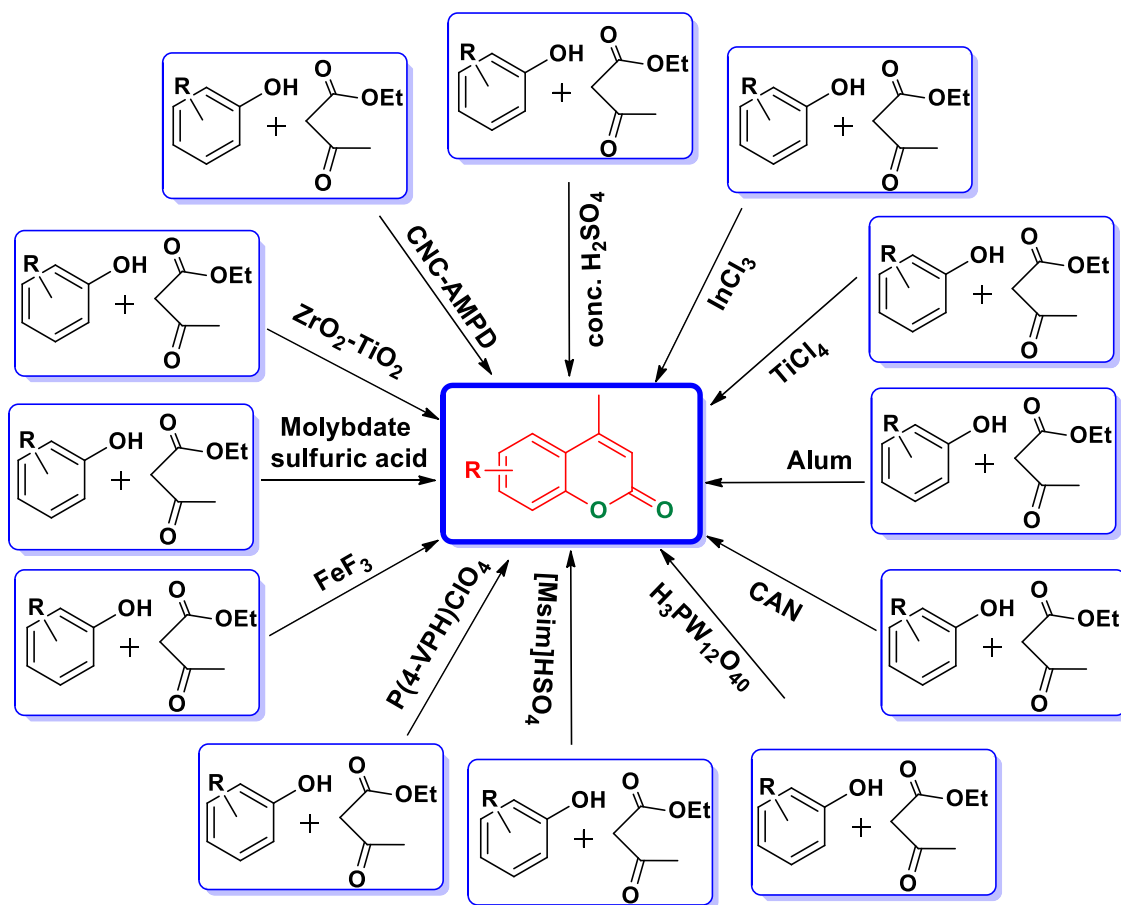
In Baylis-Hilman methodology, coumarins are synthesized by the reaction of salicylaldehyde with *t*-butyl acrylate in the presence of DABCO, HI and acetic acid (Kaye et al. 2003) (**Scheme 1.6**).



Scheme 1.6: Synthesis of coumarin by Baylis-Hilman reaction.

1.2.4 Pechmann Condensation

Traditionally coumarins have been synthesized over a decade by Pechmann reaction. In this protocol coumarins are synthesized by the reaction of phenol with β -keto esters or maleic acid in the presence of different catalysts such as conc. H_2SO_4 (Pechmann et al. 1884),

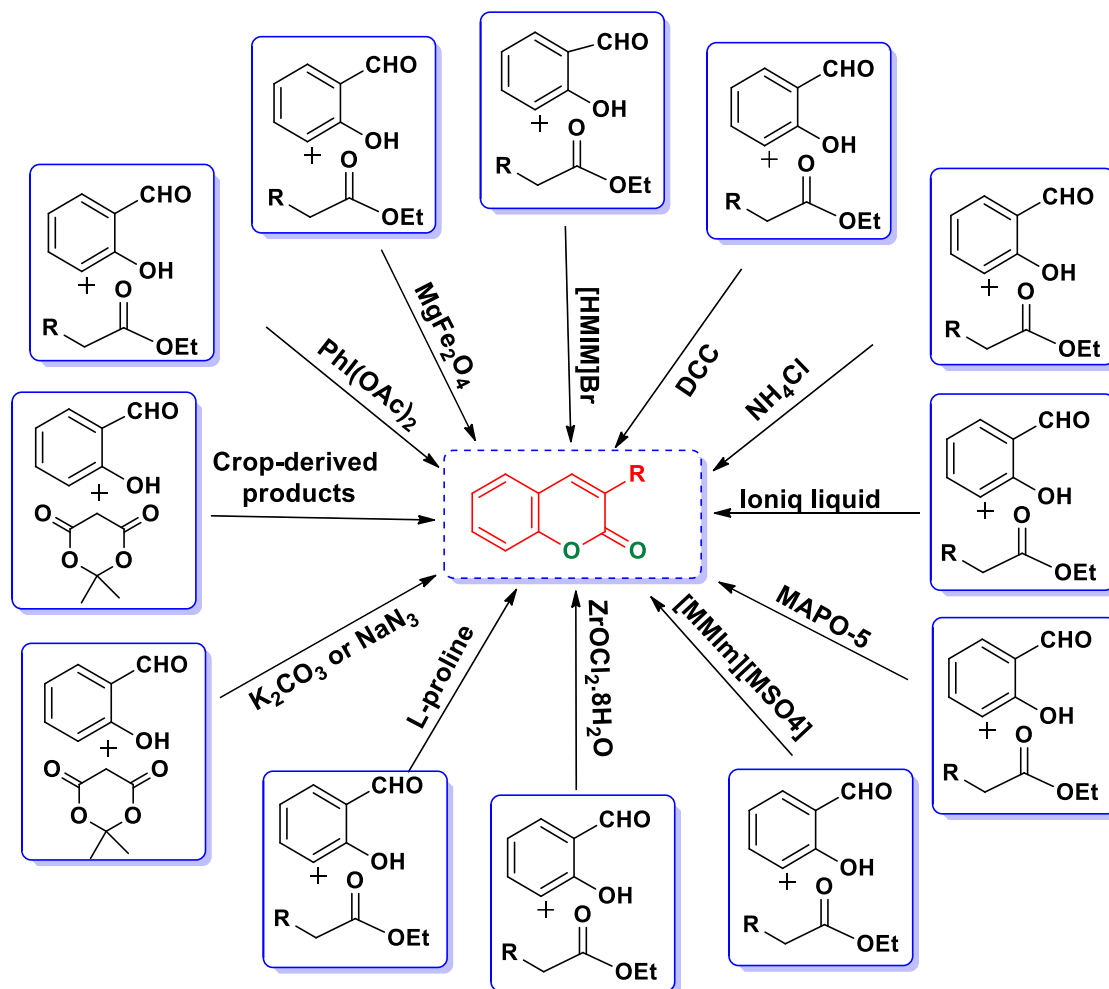


Scheme 1.7: Synthesis of coumarin by Pechmann reaction.

indium (III) chloride (Bose et al. 2002), TiCl_4 (Valizadeh et al. 2005), Alum (Dabiri et al. 2007), ceric ammonium nitrate (CAN) (Reddy et al. 2008), phosphotungstic acid (Keri et al. 2009), acidic ionic liquid (Khaligh et al. 2012), poly(4-vinylpyridinium)perchlorate (Khaligh et al. 2013), FeF_3 (Vahabi et al. 2014), molybdate sulfuric acid (Moradi et al. 2015), zirconia-based heterogeneous (Khan et al. 2016), cellulose nanocrystal supported palladium nanoparticles (Mirosanloo et al. 2018) (**Scheme 1.7**).

1.2.5 Knoevenagel Condensation

Knoevenagel condensation is most common process for the synthesis of coumarins. In this methodology coumarin have been synthesized by the reaction of *o*-hydroxybenzaldehydes *i.e.* salicylaldehyde derivatives and active methylene compounds in the presence of different catalyst such as DCC (Bonsignore et al. 1995), ammonium chloride (Valizadeh et al. 2001), ionic liquid (Ranu et al. 2006), Lewis acid metal ion-exchanged MAPO-5 molecular sieves (Gopalakrishnan et al. 2008), [MMIm][MSO₄] containing proline (Verdia et al. 2011), $\text{ZrOCl}_2 \cdot 8\text{H}_2\text{O}$ (Tasqeeruddin et al. 2013), L-proline (Srikrishna et al. 2014), K_2CO_3 or NaN_3 (Brahmachari et al. 2015), crop-derived products (Fiorito et al. 2016), $\text{PhI}(\text{OAc})_2$ (Khan et al. 2017), MgFe_2O_4 nanocatalyst (Ghomi et al. 2018), [HMIM]Br, piperidine and AcOH (Dinparast et al. 2019) (**Scheme 1.8**).



Scheme 1.8: Synthesis of coumarin by Knoevenagel condensation.

1.3 Imidazo[1,2-a]pyridine

The imidazo[1,2-a]pyridine ring system was described by Chichibabin in 1925. Imidazo[1,2-a]pyridines are an important class of nitrogen ring junction heterocyclic compounds, in this imidazole moiety is fused with the pyridine ring.

Imidazo[1,2-a]pyridine is one of the imperative fused bicyclic 5–6 heterocycles with the applications ranging from medicinal and combinatorial chemistry to material chemistry. It is also known as a ‘drug prejudice’ framework due to its broad applications in pharmaceutical chemistry. Furthermore, this scaffold is beneficial in material science due to its unique structural features. These derivatives show a wide range of biological activities like antibacterial, antitumor, antifungal, antiviral, antipyretic, analgesic activities (Bagdi et al. 2015) (**Figure 1.8**).

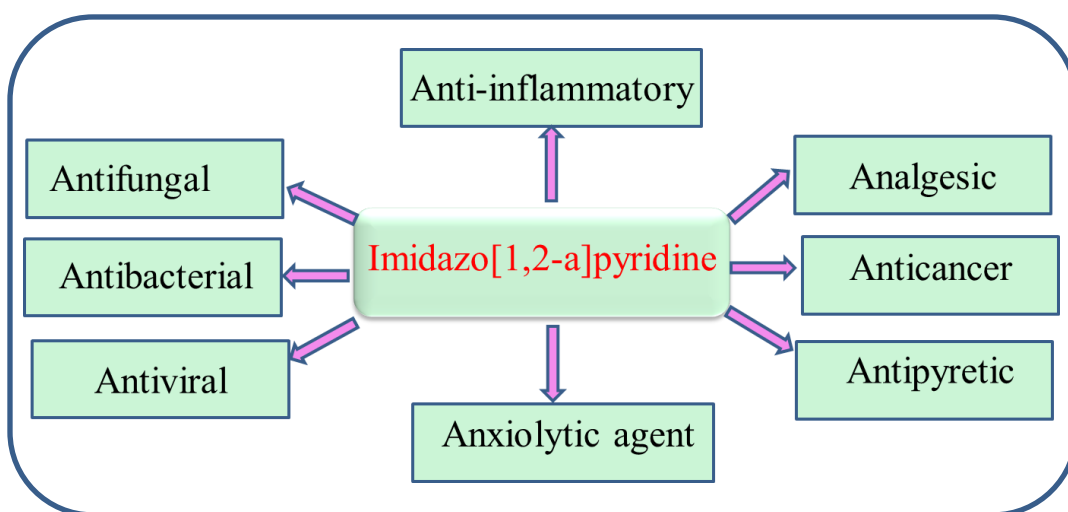


Figure 1.8: Biological activities of imidazo[1,2-a]pyridines.

Imidazo[1,2-a]pyridine moiety is present in many drugs and are used against various diseases (**Figure 1.9**). Zolpidem was the first drug launched in the market as a hypnotic, Zolimidin as antiulcer, Saripidem is commercially available drug works as an anxiolytic

agent, Miroprofen an analgesic, GSK812397 as HIV infection and P13 K alpha inhibitor as an anti-cancer agent.

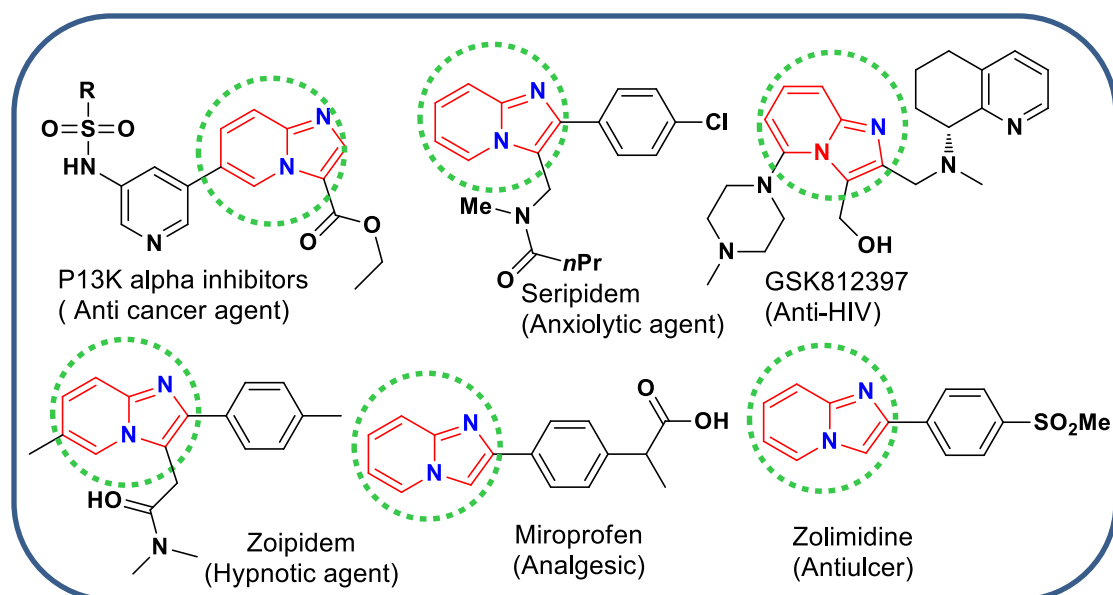


Figure 1.9: Imidazo[1,2-a]pyridine scaffold containing drugs.

Excited-state intramolecular proton transfer (ESIPT) property is observed in 2-substituted hydroxylphenyl imidazo[1,2-a]pyridines (**A**) in the majority of solvents which is responsible for strong, solid-state emission (Douhal et al. 1997). Imidazo[1,2-a]pyridines have established wide application as fluorescence sensors, laser dyes and also in molecular switches. Imidazo[1,2-a]pyridines (**B**) scaffold containing fluorescent dopamine D3 receptor ligands used as important probes for receptor visualization (Leopoldo et al. 2014).

In addition to this, metal complexes of imidazo[1,2-a]pyridine (**C**) are used in the area of electronic devices. Furthermore, 2-carbonyl-3-(pyridylamino)imidazo[1,2-a]pyridine (**D**) has been recognized as a luminous probe for mercury ion (Roopan et al. 2016) (**Figure 1.10**).

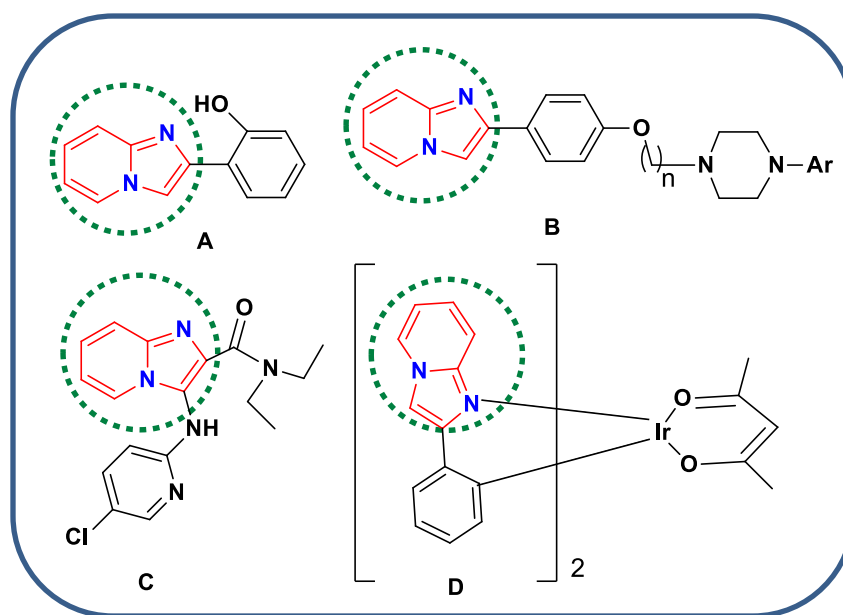


Figure 1.10: Material science application of imidazo[1,2-a]pyridine.

Due to its wide applicability in different branches of chemistry, it is desirable to synthesize this moiety from the readily available chemicals. Efforts have been focused to improve diverse synthetic approaches for the synthesis of imidazo[1,2-a]pyridines and numerous methods have been established. There are lots of methods described in literature for the synthesis of imidazo[1,2-a]pyridines and these methods can be categorized into multicomponent, tandem reaction, aminoxygenation, hydroamination, oxidative coupling

and condensation reaction. Different synthetic strategies of imidazo[1,2-a]pyridines by using various type of starting material is shown in **Figure 1.11**.

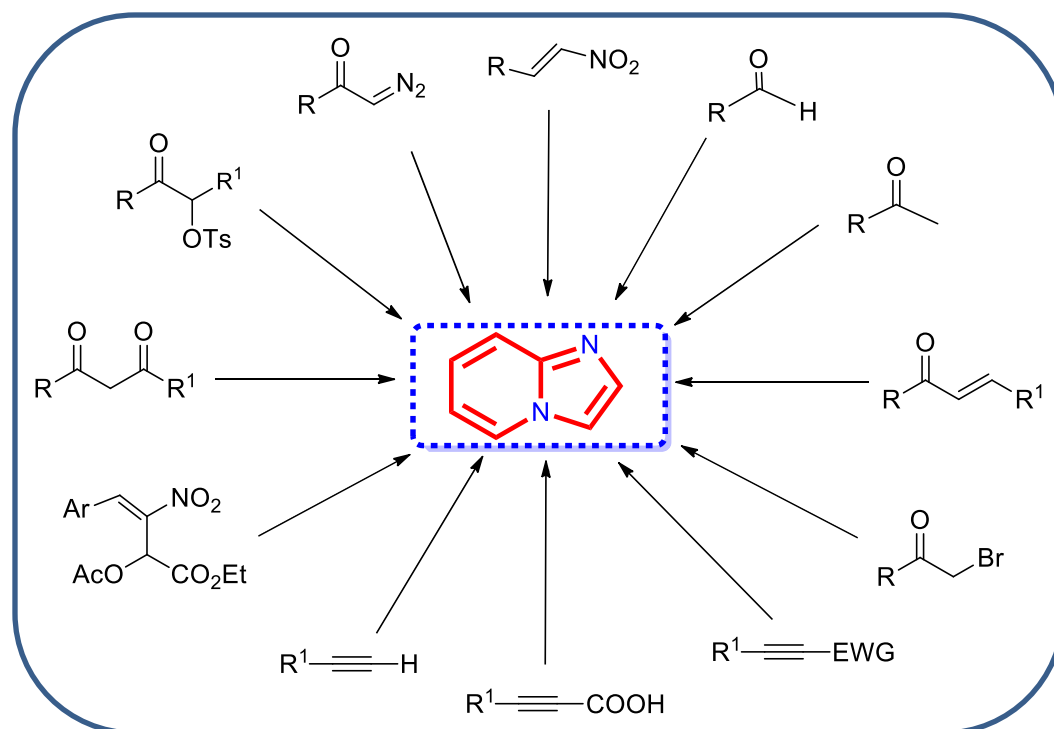
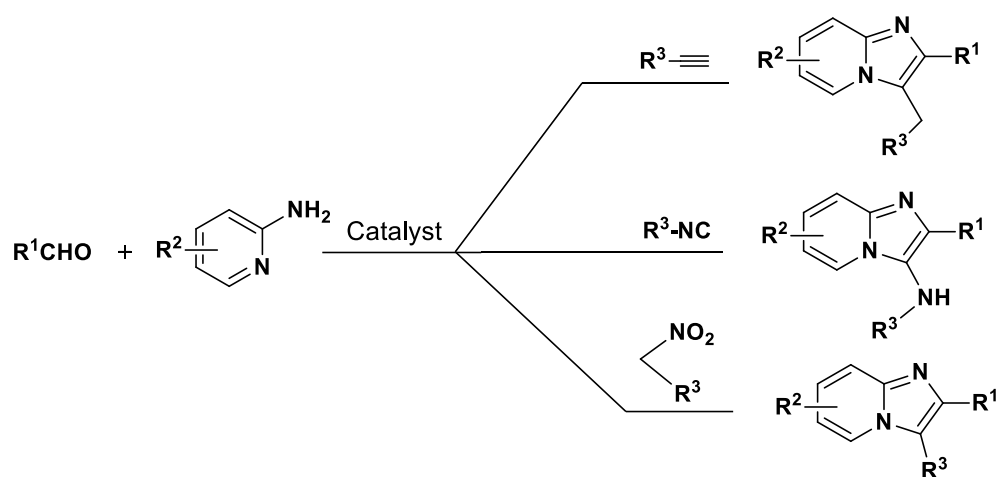


Figure 1.11: Synthesis of imidazo[1,2-a]pyridines by various starting material.

1.3.1 Multicomponent Approach

Imidazo[1,2-a]pyridines have been synthesized by the multicomponent approach through the reaction of 2-aminopyridine, aldehyde and cyanide or isocyanide in the presence of various catalysts: scandium triflate (Schwerkoske et al. 2005), copper-catalyzed (Chernyak et al. 2010), K_2CO_3 (Adib et al. 2011), bromodimethylsulfonium bromide

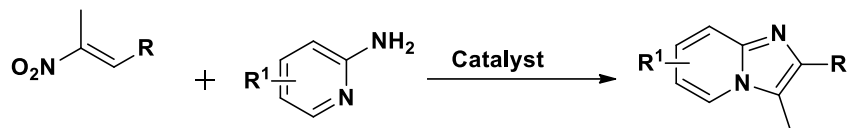
(BDMS) (Khan et al. 2012). Imidazo[1,2-a]pyridines were also achieved by the reaction of 2-aminopyridine, aldehyde and terminal alkyne catalyzed by $\text{Cu}(\text{OTf})_3$ (Chernyak et al. 2010), CuSO_4 (Liu et al. 2010), nano- $\text{Fe}_3\text{O}_4\text{-KHSO}_4\cdot\text{SiO}_2$ (Guntreddi et al. 2012) and from the reaction of 2-aminopyridine, aldehyde and nitro methane (Yan et al. 2014) (**Scheme 1.9**).



Scheme 1.9: Synthesis of imidazo[1,2-a]pyridines by multicomponent approach.

1.3.2 Tandem Reaction

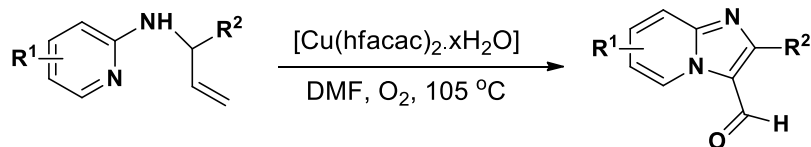
Imidazo[1,2-a]pyridines were synthesized by Tandem reaction of 2-aminopyridine with nitroalkenes in the presence of different catalyst such as iron (II) (Yan et al. 2012), iron (III) (Santra et al. 2013) (**Scheme 1.10**).



Scheme 1.10: Synthesis of imidazo[1,2-a]pyridines by Tandem reaction.

1.3.3 Aminooxygenation

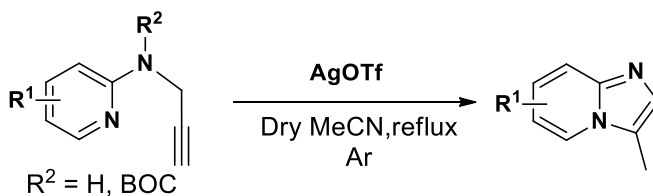
Imidazo[1,2-a]pyridines were also synthesized by copper-catalyzed intramolecular dehydrogenative aminooxygenation (Wang et al. 2011) (**Scheme 1.11**).



Scheme 1.11: Synthesis of imidazo[1,2-a]pyridines by Aminooxygenation.

1.3.4 Hydroamination

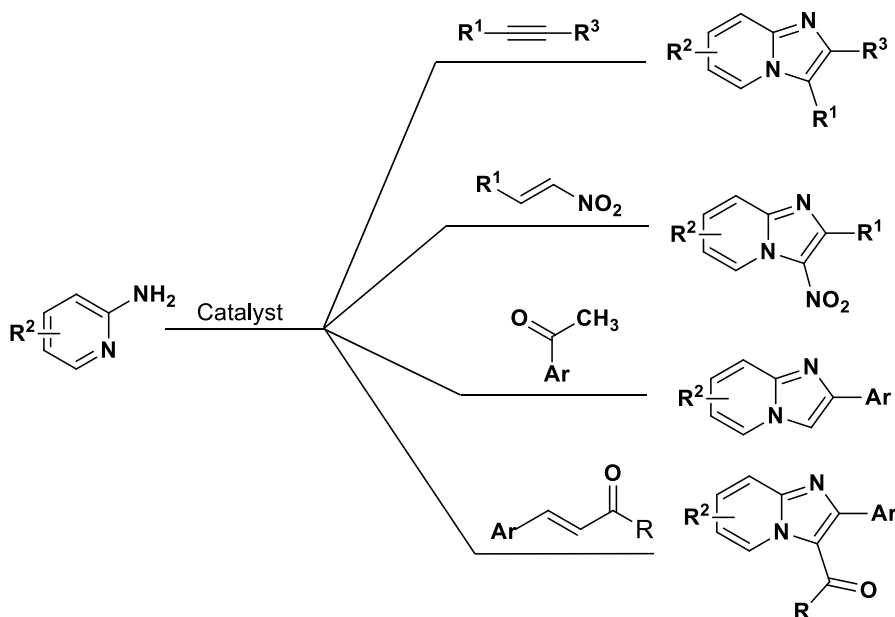
Chioua developed a method for the synthesis of 3-methylimidazo[1,2-a]pyridines through Ag-catalyzed cyclization of the N-(prop-2-yn-1-yl)pyridine-2-amines (Chioua et al. 2012) (**Scheme 1.12**).



Scheme 1.12: Synthesis of imidazo[1,2-a]pyridines by Hydroamination.

1.3.5 Oxidative Coupling

Imidazo[1,2-a]pyridines were also synthesized by oxidative coupling of 2-aminopyridine with various substrates like alkynes (Zeng et al. 2012, He et al. 2012, Gao et al. 2013), nitroolefines (Yan et al. 2012), aryl methyl ketones (Chandra et al. 2013, Pericherla et al. 2013, Meng et al. 2018), chalcones (Monir et al. 2014) etc. using different metal catalyst (Scheme 1.13).

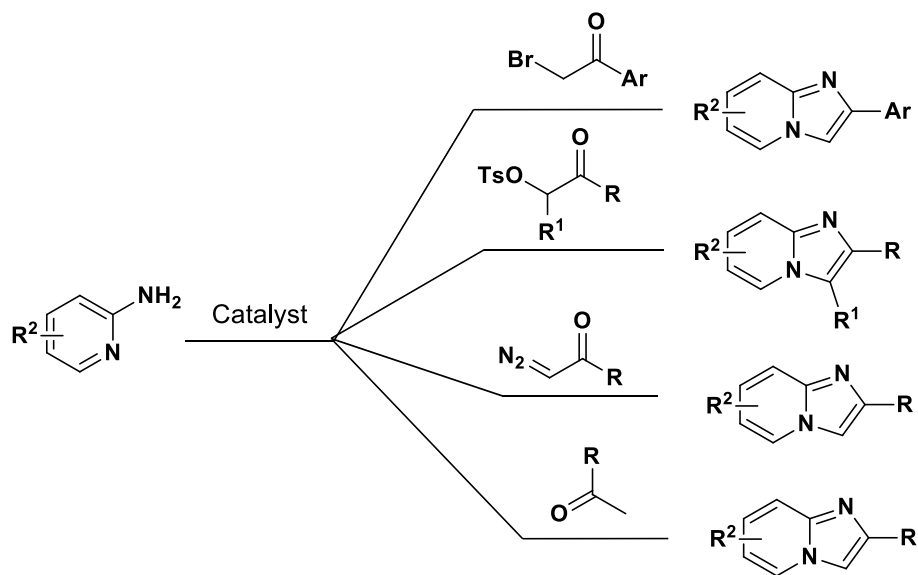


Scheme 1.13: Synthesis of imidazo[1,2-a]pyridines by oxidative coupling.

1.3.6 Condensation Reaction

Imidazo[1,2-a]pyridines were synthesized by the condensation of α -haloketones and 2-aminopyridines in the presence of different catalysts like neutral Al_2O_3 (Ponnala et al. 2005), NaHCO_3 (Hiebe et al. 2014), MgO (Patil et al. 2016), NaHCO_3 in eucalyptol

(Campos et al. 2019) and also in catalyst free condition (Burkholder et al. 2001, Zhu et al. 2009, Kong et al. 2016). Imidazo[1,2-a]pyridines were also synthesized by the reaction of 2-aminopyridine with different substrate such as α -tosyloxyketones catalyzed by BPyBF₄, Na₂CO₃ (Xie et al. 2002), α -diazoketones catalyzed by Cu (Yadav et al. 2007), methyl aryl ketones in the presence of different catalysts such as iodobenzene (Chang et al. 2010), I₂/NaOH (Stasyuk et al. 2012), graphene oxide (GO)/NaI (Kundu et al. 2015), iodine-ammonium acetate (Kour et al. 2016), NBS (Bhagat et al. 2017), I₂ in cyclohexane (Ghosh et al. 2018), HI (Feng et al. 2019), FeCl₃/I₂ (Ujwaldev et al. 2019) etc. (**Scheme 1.14**).



Scheme 1.14: Synthesis of imidazo[1,2-a]pyridines by condensation reaction.

1.4 4*H*-pyran

4*H*-pyran is oxygen containing heterocyclic compound which displays significant biological and pharmacological activities and also constitute a structural unit of a series of natural products. A number of 2-amino-4*H*-pyrans played imperative role in the discipline of drugs, photoactive materials, agrochemicals, cosmetics and pigment industries shown in Figure 1.12 (Fotouhi et al. 2007).

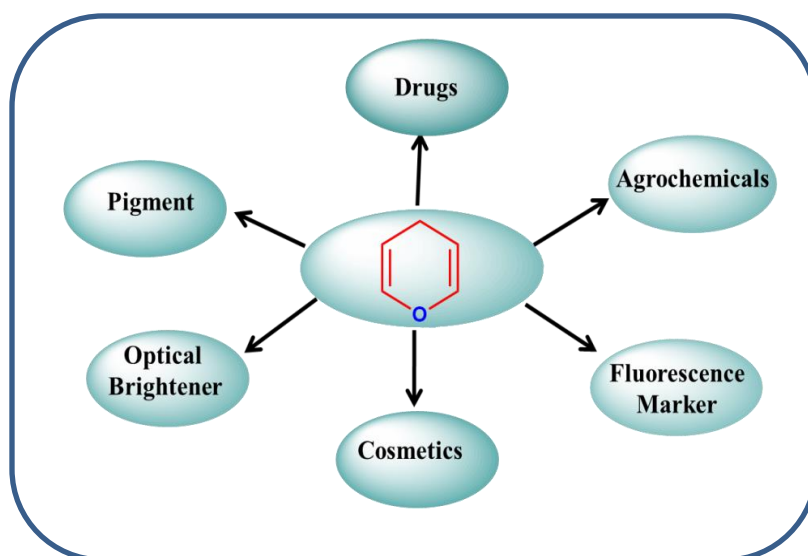


Figure 1.12: Applications of 4*H* -pyrans in different fields.

4*H*-pyrans are ubiquitous structural moiety in bioactive molecules due to their good biocompatibility. Mainly cyano functionality in 2-amino-4*H*-pyrans derivatives has an impending utility in the treatment of various diseases like cancer, rheumatoid, psoriasis (Azath et al. 2012) and also in the treatment of various neurodegenerative disease such as

alzheimer's disease, AIDS-associated dementia and down syndrome also for the treatment of schizophrenia and huntington's diseases (Thanaraj et al. 2019) (**Figure 1.13**).

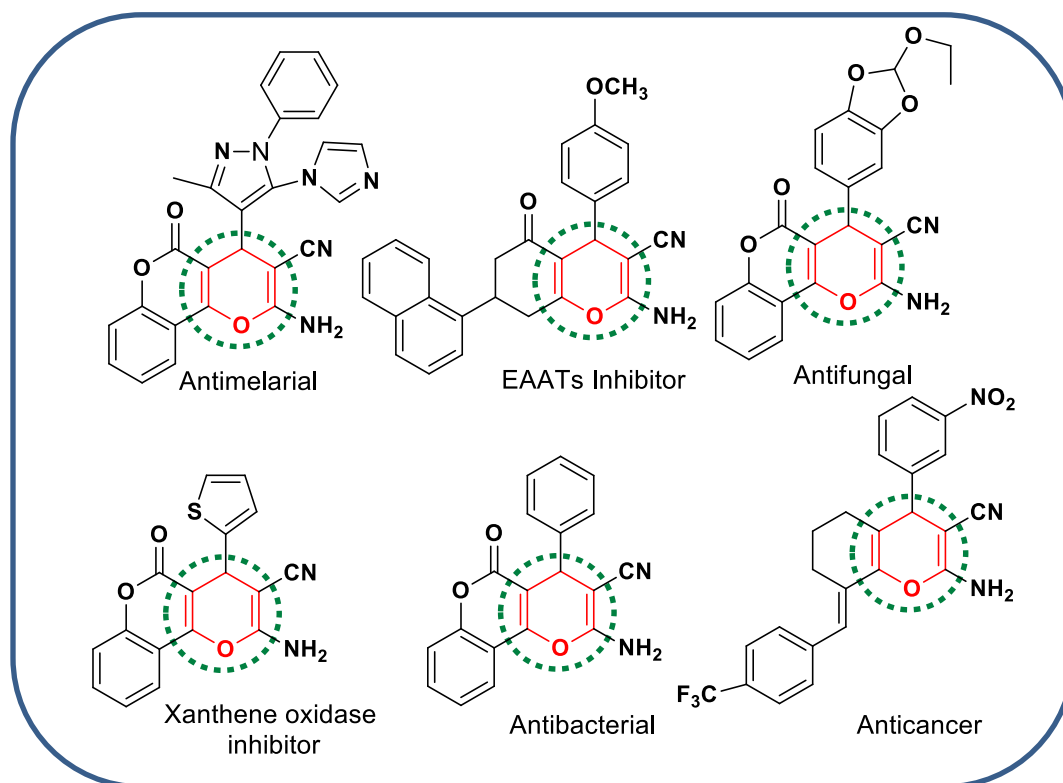
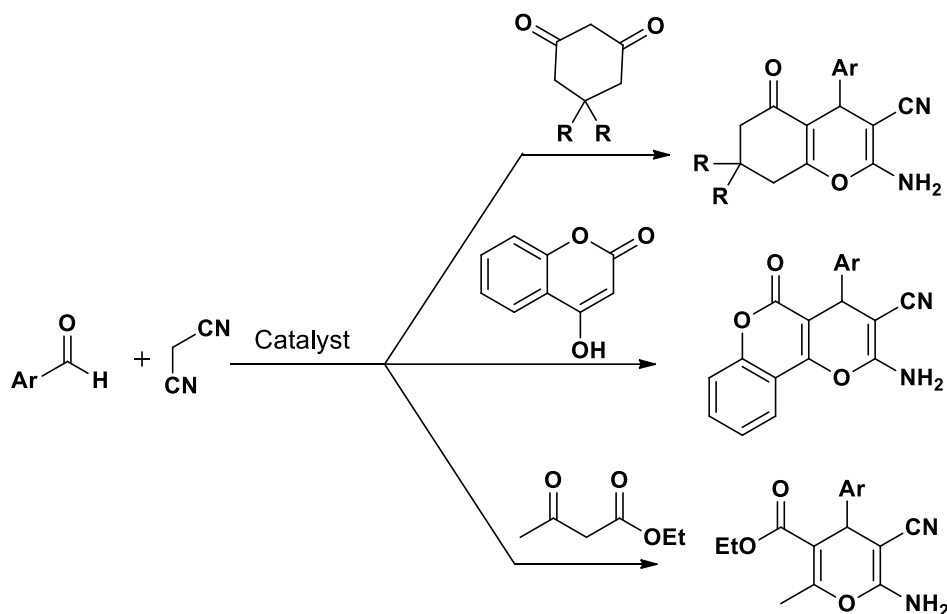


Figure 1.13: Drugs containing 4H-pyran moiety.

Considering the importance of these compounds, many methods have been reported for the synthesis of tetrahydro-4H-benzo[b]pyran derivatives. Out of the several approaches for the synthesis of 2-amino-4H-pyran, multicomponent reaction (MCR) is the most useful and preferred method for the construction of these heterocyclic compounds. The best protocol to synthesize 4H-pyrans is the Knoevenagel condensation-Michael cyclization

reaction by using aldehyde, carbonitrile and 1,3-dicarbonyl compound by one-pot multicomponent reaction.

Synthesis of 4*H*-pyran can be achieved by the reaction of benzaldehyde, malononitrile with different 1,3-dicarbonyl compounds in the presence of different catalyst such as triethyl amine (Elnagdi et al. 1989), ammonium acetate (Tu et al. 2002), KF-Al₂O₃ (Wang et al. 2003), S-proline (Balalaie et al. 2006), electrogenerated base (Fotouhi et al. 2007), tetrabutylammonium bromide (Gurumurthi et al. 2009), per-6-amino-β-cyclodextrin (Azath et al. 2012), potassium tertiary butoxide (Rao et al. 2018), TiO₂/H₁₄[NaP₅W₃₀O₁₁₀] (Azarifar et al. 2014), AuNPs@RGO-SH (Naeimi et al. 2017), NiFe₂O₄ nanoparticle-supported (Maleki et al. 2016), [pyridine-SO₃H]Cl (Sonyanaik et al. 2018),



Scheme 1.15: Common acid base catalyzed synthesis of 4*H*-pyran.

CuO-CNs (Thanaraj 2019), Cu@MNPs (Wanzheng et al. 2019). 4-*H* pyran can also be synthesized under catalyst free condition (Bandgar et al. 2007, Ponpandian et al. 2014, Elinson et al. 2015, Bakherad et al. 2019) (**Scheme 1.15**).

1.5 Objectives of Thesis Work

From this brief introduction, it is clear that heterocyclic compounds have found wide applications in various fields including organic synthesis, biochemistry, medicinal chemistry, material sciences and agriculture etc. Thus, our objective is to develop some efficient and greener methodology for synthesis of some biologically active heterocyclic compounds via conventional as well as non-conventional methods such as ultrasound, solar energy and grinding techniques which may make an encouraging contribution to the development of the green and clean chemistry.

The main focus of the current thesis work is aimed-

1. To investigate oxidative dimerization of primary thiobenzamides and selenamides into 1,2,4-thiadiazole and 1,2,4-selenadiazole by different methods under metal and catalyst-free condition using green oxidants TBN and Chloranil.
2. To develop a practical method for the synthesis of 3-functionalized coumarins by oxidative coupling of *o*-cresols and active methylene compounds under metal and catalyst-free condition in the presence of *tert*-butyl hydrogen peroxide (TBHP).

3. To explore a novel, facile, efficient and scalable route for the synthesis of imidazo[1,2-a]pyridines in solvent free conditions under grinding at room temperature using KI/TBHP.
4. To demonstrate a solar energy mediated environmentally benign, simple and efficient method for one pot multicomponent synthesis of tetrahydrobenzo[b]pyran using L-Ascorbic acid as a green biodegradable catalyst.

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