

**APPENDIX A**

**Table A1:** Data for removal efficiency with time using NaClO in bubble column (initial SO<sub>2</sub> concentration 6348 ppm, initial NO concentration 1804 ppm, initial pH 5.6, initial NaClO concentration 0.032 M, temperature 305 K)

S. No.	Time (min)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	0	0	0
2	5	45.83	97.12
3	10	62.34	97.23
4	15	76.18	97.32
5	20	84.52	97.24
6	25	85.15	97.28
7	30	86.25	97.52
8	35	91.25	98.25
9	40	91.23	98.35
10	50	89.25	98.25
11	60	90.13	98.12
12	70	90.12	98.56
13	80	89.25	97.54
14	90	89.56	97.64

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<b>15</b>	100	89.23	98.14
<b>16</b>	110	88.25	97.42
<b>17</b>	120	89.68	98.25
<b>18</b>	130	85.25	96.13
<b>19</b>	140	82.12	92.13
<b>20</b>	150	78.32	89.12
<b>21</b>	160	72.13	78.12
<b>22</b>	170	65.12	74.97
<b>23</b>	180	61.13	68.12
<b>24</b>	190	58.23	65.12
<b>25</b>	200	53.25	56.12
<b>26</b>	210	40.12	50.35
<b>27</b>	220	35.12	48.23
<b>28</b>	220	23.45	45.28
<b>29</b>	230	18.25	33.89
<b>30</b>	240	11.23	29.56
<b>31</b>	250	6.12	23.28
<b>32</b>	260	0	19.63
<b>33</b>	270	0	18.23
<b>34</b>	280	0	15.23
<b>35</b>	290	0	12.36

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<b>36</b>	300	0	8.25
<b>37</b>	310	0	4.67
<b>38</b>	320	0	2.63
<b>39</b>	330	0	1.13
<b>40</b>	340	0	0.89
<b>41</b>	350	0	0

**Table A2:** Data for removal efficiency with NaClO concentration in bubble column (initial SO<sub>2</sub> concentration 6348 ppm, initial NO concentration 1804 ppm, temperature 305 K, initial pH 5.6)

<b>S. No.</b>	<b>NaClO concentration (M)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	0.004	82.73	96.14
<b>2</b>	0.008	83.71	97.12
<b>3</b>	0.012	85.97	97.15
<b>4</b>	0.016	86.71	98.12
<b>5</b>	0.02	87.97	98.25
<b>6</b>	0.024	89.71	98.12

<b>7</b>	0.028	90.55	98.05
<b>8</b>	0.032	90.97	98.35
<b>9</b>	0.036	90.97	98.15
<b>10</b>	0.04	90.97	98.62

**Table A3:** Data for removal efficiency with absorbent temperature using NaClO in stirred vessel (initial SO<sub>2</sub> concentration 6348 ppm, initial NO concentration 1804 ppm, initial pH 5.6, initial NaClO concentration 0.032 M)

<b>S. No.</b>	<b>Temperature (K)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	283	82.291234	97.9823
<b>2</b>	294	86.91452097	97.58174905
<b>3</b>	305	91.9690926	97.84790875
<b>4</b>	315	87.56278214	97.80988593
<b>5</b>	324	87.41190006	97.88593156

**Table A4:** Data for removal efficiency with time using NaClO in stirred vessel (initial SO<sub>2</sub> concentration 6340 ppm, initial NO concentration 816 ppm, initial NaClO concentration 0.01 M, temperature 313 K)

S. No.	Time (min)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	0	0.00	0.00
2	20	99.81	92.56
3	40	99.58	89.66
4	60	98.97	85.81
5	80	98.52	80.36
6	100	98.10	75.84
7	120	97.38	71.45

**Table A5:** Data for removal efficiency with NaClO concentration in stirred vessel (initial SO<sub>2</sub> Concentration 6340 ppm, initial NO Concentration 816 ppm, temperature 313 K)

<b>S. No.</b>	<b>NaClO concentration (M)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
1	0.002	84.73	97.34
2	0.005	88.71	99.32
3	0.008	91.85	99.64
4	0.010	92.56	99.85
5	0.012	92.58	99.81

**Table A6:** Data for removal efficiency with absorbent temperature using NaClO in stirred vessel (initial SO<sub>2</sub> concentration 6340 ppm, initial NO concentration 816 ppm, initial NaClO concentration 0.01 M)

S. No.	Temperature (K)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	293	87.91	99.58
2	303	90.97	99.85
3	313	92.56	99.81
4	323	91.10	99.89

**Table A7:** Data for removal efficiency with initial SO<sub>2</sub> concentration using NaClO in stirred vessel (initial NO concentration 816 ppm, initial NaClO concentration 0.01 M, temperature 313 K)

S. No.	SO <sub>2</sub> concentration (ppm)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	4171	84.78	99.71
2	5124	87.51	99.77
3	6340	92.56	99.81
4	7136	93.26	99.83

**Table A8:** Data for removal efficiency with initial NO concentration using NaClO in stirred vessel (initial SO<sub>2</sub> concentration 6340 ppm, initial NaClO concentration 0.01 M, temperature 313 K)

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<b>S. No.</b>	<b>NO concentration (ppm)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	508	87.10	99.71
<b>2</b>	614	89.12	99.77
<b>3</b>	721	91.28	99.81
<b>4</b>	817	92.56	99.83

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**Table A9:** Data for removal efficiency with initial pH of NaClO in stirred vessel (initial SO<sub>2</sub> concentration 6340 ppm, initial NO concentration 816 ppm, initial NaClO concentration 0.01 M, temperature 313 K)

S. No.	Initial pH	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	4	88.14	99.58
2	5	91.27	99.78
3	5.8	92.12	99.85
4	6.8	86.67	99.81
5	7.6	77.45	99.82
6	9	54.23	99.27

**Table A10:** Data for removal efficiency with  $\text{Ca}(\text{OCl})_2$  concentration in stirred vessel (initial  $\text{SO}_2$  concentration 6126 ppm, initial NO concentration 876 ppm, temperature 313 K)

<b>S. No.</b>	<b><math>\text{Ca}(\text{OCl})_2</math> concentration (g/100ml)</b>	<b>NO removal efficiency (%)</b>	<b><math>\text{SO}_2</math> removal efficiency (%)</b>
1	1	74.73	97.34
2	2	78.71	99.32
3	4	80.97	99.85
4	5	84.47	99.89
5	6	84.68	99.91
6	7	84.97	99.95

**Table A11:** Data for removal efficiency with absorbent temperature using  $\text{Ca}(\text{OCl})_2$  in stirred vessel (initial  $\text{SO}_2$  concentration 6126 ppm, initial NO concentration 876 ppm, absorbent concentration 5 g/100ml)

S. No.	Temperature (K)	NO removal efficiency (%)	$\text{SO}_2$ removal efficiency (%)
1	293	84.91	99.58
2	303	87.97	99.85
3	313	89.56	99.81
4	323	90.10	99.89

**Table A12:** Data for removal efficiency with mole ratio of absorbent blend NaOH/NaClO in magnetic stirred vessel (initial SO<sub>2</sub> concentration 6829 ppm, initial NO concentration 850 ppm, temperature 313 K, initial pH 5.9, NaClO concentration 0.024 M)

S. No.	Mole ratio of NaClO/NaOH	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	0.5	80.73	98.14
2	1	81.71	98.12
3	1.5	82.97	98.15
4	2	83.71	98.12
5	2.5	85.97	98.25
6	3	87.71	98.12
7	3.5	87.97	98.35
8	4	87.96	98.35

**Table A13:** Data for removal efficiency with absorbent temperature in magnetic stirrer vessel using NaOH/NaClO blend (initial SO<sub>2</sub> concentration 6829 ppm, initial NO concentration 850 ppm, initial pH 5.9, mole ratio of NaOH/NaClO 3)

S. No.	Temperature (K)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	293	81.91	97.58
2	303	84.97	97.85
3	313	87.56	97.81
4	323	87.41	97.89

**Table A14:** Data for estimation of  $k_{SO_2g}$  with NaClO

S. No.	Absorption rate ( $\times 10^{-6}$ mole/m <sup>2</sup> ·s)	Partial pressure of SO <sub>2</sub> ( $\times 10^{10}$ .Pa)
1	1.67	2.37
2	2.06	2.92
3	2.54	3.61
4	2.86	4.06

**Table A15:** Data for estimation of reaction order for NO with NaClO

S. No.	$-\log (C_{NOi})$	$\log (N_{NO})$
1	9.92	12.13
2	9.72	11.92
3	9.56	11.73
4	9.44	11.60

**Table A16:** Data for enthalpy change with temperature using NaClO

S. No.	Temperature (K)	$\Delta H$ (kJ·mol <sup>-1</sup> )
1	283	3171.63
2	293	3178.24
3	303	3184.84
4	313	3191.45
5	323	3198.05
6	333	3204.65
7	343	3211.26
8	353	3217.86

**Table A17:** Data for Gibbs free energy change with temperature using NaClO

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S. No.	Temperature (K)	$\Delta G$ (kJ·mol <sup>-1</sup> )
1	303	2932.05
2	313	2703.36
3	323	2585.47
4	333	2499.16
5	343	2427.68
6	353	2364.69

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**Table A18:** Data for removal efficiency with time using NaClO in spray column (gas flow rate 600 mL/min, liquid flow rate 1500 mL/min, initial NaClO concentration 0.024 M, distributor diameter 1 mm, temperature 313 K)

S. No.	Time (min)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	0	0.00	0.00
2	20	82.56	97.81
3	40	75.66	97.58
4	60	71.81	96.97
5	80	66.36	96.52
6	100	63.84	96.10
7	120	57.45	95.38
8	140	51.81	90.97
9	160	46.36	86.52
10	180	38.84	81.10
11	200	31.45	75.38

**Table A19:** Data for removal efficiency with absorbent (NaClO) concentration in spray column (gas flow rate 600 mL/min, liquid flow rate 1500 mL/min, temperature 313 K, distributor diameter 1 mm, time 120 min)

<b>S. No.</b>	<b>NaClO concentration (M)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
1	0.004	64.73	91.34
2	0.008	68.71	93.32
3	0.012	70.97	95.85
4	0.016	74.71	97.32
5	0.02	78.97	97.85
6	0.024	79.71	98.32
7	0.028	81.97	98.85
8	0.032	82.56	98.85

**Table A20:** Data for removal efficiency with reaction temperature using NaClO in spray column (gas flow rate 600 mL/min, liquid flow rate 1500 mL/min, initial NaClO concentration 0.024 M, distributor diameter 1 mm, time 120 min)

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<b>S. No.</b>	<b>Temperature (K)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	293	77.91	97.58
<b>2</b>	303	80.97	97.85
<b>3</b>	313	82.56	97.81
<b>4</b>	323	81.10	97.89

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**Table A21:** Data for removal efficiency with initial SO<sub>2</sub> concentration using NaClO in spray column (gas flow rate 600 mL/min, liquid flow rate 1500 mL/min, initial NaClO concentration 0.024 M, distributor diameter 1 mm, temperature 313 K, time 120 min)

S. No.	SO <sub>2</sub> concentration (ppm)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	2432	81.56	97.81
2	3528	81.66	97.58
3	4635	82.81	96.97
4	5712	86.36	96.52
5	6817	87.84	96.10

**Table A22:** Data for removal efficiency with initial NO concentration using NaClO in spray column (gas flow rate 600 mL/min, liquid flow rate 1500 mL/min, initial NaClO concentration 0.024 M, distributor diameter 1 mm, temperature 313 K, time 120 min)

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<b>S. No.</b>	<b>NO concentration (ppm)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	432	80.56	97.81
<b>2</b>	528	81.66	97.58
<b>3</b>	635	84.81	96.97
<b>4</b>	712	86.36	96.52
<b>5</b>	817	87.84	96.10

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**Table A23:** Data for removal efficiency with initial pH of absorbent using NaClO in spray column (gas flow rate 600 mL/min, liquid flow rate 1500 mL/min, initial NaClO concentration 0.024 M, distributor diameter 1 mm, temperature 313 K, time 120 min)

S. No.	Initial pH	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	4.1	82.56	97.81
2	4.9	84.66	97.58
3	5.4	85.81	96.97
4	6.7	76.36	96.52
5	7.8	63.84	96.10
6	9.2	51.45	95.38

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**Table A24:** Data for removal efficiency with addition of CO<sub>2</sub> to simulated gas stream using NaClO in spray column (initial NO concentration 600 ppm, initial SO<sub>2</sub> concentration 2600 ppm, initial CO<sub>2</sub> concentration 50000 ppm, initial NaClO concentration 0.024 M, temperature 313 K, distributor diameter 1 mm)

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<b>S. No.</b>	<b>Time (min)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>	<b>CO<sub>2</sub> removal efficiency (%)</b>
1	0	0.00	0.00	0.00
2	20	82.56	97.81	41.24
3	40	75.66	97.58	38.12
4	60	71.81	96.97	37.33
5	80	66.36	96.52	36.94
6	100	63.84	96.10	35.82
7	120	57.45	95.38	34.62
8	140	51.81	90.97	33.12
9	160	46.36	86.52	30.24
10	180	38.84	81.10	26.12
11	200	31.45	75.38	22.32

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**Table A25:** Data for enthalpy change with temperature using NaClO/NH<sub>3</sub>

S. No.	Temperature (K)	-ΔH (SO <sub>2</sub> ) (kJ·mol <sup>-1</sup> )	-ΔH (NO) (kJ·mol <sup>-1</sup> )
1	303	819.17	503.70
2	313	821.50	503.86
3	323	823.82	504.02
4	333	826.14	504.18
5	343	828.47	504.34
6	353	830.79	504.50

**Table A26:** Data for Gibbs free energy change with temperature using NaClO/NH<sub>3</sub>

S. No.	Temperature (K)	-ΔG (SO <sub>2</sub> ) (kJ·mol <sup>-1</sup> )	-ΔG (NO) (kJ·mol <sup>-1</sup> )
1	303	913.98	409.20
2	313	999.36	400.74
3	323	1045.71	394.97
4	333	1080.95	389.96
5	343	1110.96	385.32
6	353	1137.98	380.87

**Table A27:** Data for estimation of equilibrium time in semi batch stirred with NaClO/NH<sub>3</sub> (initial SO<sub>2</sub> concentration 1807 ppm, initial NO concentration 1020 ppm, temperature 313 K, initial pH 5.5, NaClO concentration 0.032 M, mole ratio of NH<sub>3</sub> to NaClO 1)

S. No.	Time (min)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	0	0	0
2	5	52.83	98.25
3	10	75.24	98.24
4	15	83.86	97.92
5	20	89.21	97.12
6	25	91.25	98.87
7	30	91.39	99.15
8	35	92.23	99.45
9	40	91.86	99.42
10	50	91.96	99.23
11	60	92.24	99.14
12	70	91.43	98.23
13	80	91.56	98.56
14	90	92.03	98.42
15	100	91.85	98.41

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<b>16</b>	110	91.53	98.23
<b>17</b>	120	91.08	98.15
<b>18</b>	130	91.13	98.65
<b>19</b>	140	91.42	98.1
<b>20</b>	150	90.43	97.45
<b>21</b>	160	90.12	97.86
<b>22</b>	170	90.21	97.12
<b>23</b>	180	90.1	97.02
<b>24</b>	190	87.43	95.14
<b>25</b>	200	81.45	93.65
<b>26</b>	210	73.84	91.23
<b>27</b>	220	65.12	86.14
<b>28</b>	220	59.46	84.12
<b>29</b>	230	51.78	81.23
<b>30</b>	240	42.43	78.96
<b>31</b>	250	30.06	73.43
<b>32</b>	260	17.85	68.59
<b>33</b>	270	9.58	61.32
<b>34</b>	280	2.32	58.23
<b>35</b>	290	0	54.12
<b>36</b>	300	0	50.13

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<b>37</b>	310	0	46.75
<b>38</b>	320	0	40.39
<b>39</b>	330	0	34.67
<b>40</b>	340	0	29.13
<b>41</b>	350	0	23.48
<b>42</b>	360	0	18.25
<b>43</b>	370	0	13.56
<b>44</b>	380	0	10.45
<b>45</b>	390	0	5.32
<b>46</b>	400	0	2.89
<b>47</b>	410	0	1.24
<b>48</b>	420	0	0

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**Table A28:** Data for estimation of optimal molar ratio in semi batch stirred with NaClO/NH<sub>3</sub> (initial SO<sub>2</sub> concentration 1807.58 ppm, initial NO concentration 1020.51 ppm, temperature 313 K, initial pH 5.5, NaClO concentration 0.032 M, time 180 min)

<b>S. No.</b>	<b>Mole ratio of NaClO/NH<sub>3</sub></b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
1	0.08	75.43	95.62
2	0.15	77.28	95.89
3	0.23	80.14	96.48
4	0.30	83.56	97.82
5	0.38	88.53	98.56
6	0.46	90.42	98.32
7	0.53	90.25	98.16
8	0.61	91.23	98.53
9	0.68	91.86	98.46
10	0.76	92.13	98.31
11	0.84	92.48	98.43
12	0.91	92.43	98.25
13	0.99	93.65	98.68
14	1.06	93.42	98.64
15	1.14	93.13	98.13

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<b>16</b>	1.22	93.03	97.43
<b>17</b>	1.29	92.77	97.16
<b>18</b>	1.37	92.56	96.73
<b>19</b>	1.44	92.34	96.30
<b>20</b>	1.52	90.65	95.88
<b>21</b>	1.60	89.85	95.45
<b>22</b>	1.67	88.86	95.03
<b>23</b>	1.75	88.13	89.25
<b>24</b>	1.82	87.42	87.83
<b>25</b>	1.90	86.43	84.25
<b>26</b>	1.98	85.86	82.42
<b>27</b>	2.05	85.12	80.32
<b>28</b>	2.13	83.89	78.42
<b>29</b>	2.20	83.59	77.52
<b>30</b>	2.28	82.85	75.86
<b>31</b>	2.35	82.11	75.54
<b>32</b>	2.43	81.37	75.22
<b>33</b>	2.51	80.63	74.90
<b>34</b>	2.58	79.90	74.58
<b>35</b>	2.66	79.16	74.26
<b>36</b>	2.73	78.42	73.94

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<b>37</b>	2.81	77.68	73.62
<b>38</b>	2.89	76.94	73.30
<b>39</b>	2.96	76.21	72.98
<b>40</b>	3.04	75.47	72.66

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**Table A29:** Data for estimation of optimal temperature in semi batch stirred with NaClO/NH<sub>3</sub> (initial SO<sub>2</sub> concentration 1807.58 ppm, initial NO concentration 1020.51 ppm, initial pH 5.5, NaClO concentration 0.032 M, time 180 min, mole ratio of NH<sub>3</sub> to NaClO 1)

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<b>S. No.</b>	<b>Temperature (K)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	283	78.96	97.14
<b>2</b>	293	83.14	98.16
<b>3</b>	303	88.42	98.56
<b>4</b>	313	91.89	99.45
<b>5</b>	323	91.12	98.89
<b>6</b>	333	87.42	97.25

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**Table A30:** Data for estimation of optimal pH in semi batch stirred with NaClO/NH<sub>3</sub> (initial SO<sub>2</sub> concentration 1807.58 ppm, initial NO concentration 1020.51 ppm, temperature 313 K, NaClO concentration 0.032 M, time 180 min, mole ratio of NH<sub>3</sub> to NaClO 1)

S. No.	pH	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	3.5	69.83	96.74
2	4	74.25	97.38
3	4.5	79.12	97.89
4	5	85.78	98.41
5	5.5	92.14	99.13
6	6	90.81	99.24
7	6.5	88.12	99.02
8	7	82.23	99.01
9	7.5	80.23	98.52
10	8	77.25	98.25
11	8.5	75.23	98.21
12	9	71.25	98.45
13	9.5	70.25	97.82
14	10	70.12	98.35

**Table A31:** Data for removal efficiency with initial SO<sub>2</sub> concentration in semi batch stirred with NaClO/NH<sub>3</sub> (initial NO concentration 1020.51 ppm, temperature 313 K, NaClO concentration 0.032 M, initial pH 5.5, time 180 min, mole ratio of NH<sub>3</sub> to NaClO 1)

<b>S. No.</b>	<b>SO<sub>2</sub> concentration (ppm)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
1	214	92.84	99.12
2	428	92.42	98.25
3	615	91.89	99.45
4	843	92.13	99.01
5	1026	91.83	98.45
6	1202	90.89	98.42
7	1434	91.24	97.23
8	1617	92.89	97.56
9	1808	91.65	98.42
10	2087	90.84	99.21
11	3103	90.25	97.13
12	4265	90.42	96.14
13	5028	90.37	97.02
14	6043	86.24	96.89

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<b>15</b>	7126	83.25	95.24
<b>16</b>	8216	80.12	92.35

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**Table A32:** Data for removal efficiency with initial NO concentration in semi batch stirred with NaClO/NH<sub>3</sub> (initial SO<sub>2</sub> concentration 1807.58 ppm, temperature 313 K, NaClO concentration 0.032 M, initial pH 5.5, time 180 min, mole ratio of NH<sub>3</sub> to NaClO 1)

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<b>S. No.</b>	<b>NO concentration (ppm)</b>	<b>NO removal efficiency (%)</b>	<b>SO<sub>2</sub> removal efficiency (%)</b>
<b>1</b>	200	88.32	98.81
<b>2</b>	421	89.89	97.62
<b>3</b>	663	92.92	96.97
<b>4</b>	821	92.48	96.52
<b>5</b>	1021	91.83	96.01
<b>6</b>	1276	90.95	96.53
<b>7</b>	1412	89.45	96.28
<b>8</b>	1626	89.58	95.14
<b>9</b>	1845	88.64	95.23

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**Table A33:** Data for NO removal efficiency change in both semi batch and spray column with NaClO/NH<sub>3</sub>

<b>S. No.</b>	<b>Parameters</b>	<b>Semi Batch</b>	<b>Spray</b>
1	SO <sub>2</sub> concentration	92.89	89.42
2	NO concentration	92.92	88.13
3	Temperature	91.89	87.28
4	Mole ratio	93.65	89.35
5	Time	92.24	88.13
6.	Average removal efficiency	92.72	88.46
7.	Removal drop		4.59

**Table A34:** Data for SO<sub>2</sub> removal efficiency change in both semi batch and spray column with NaClO/NH<sub>3</sub>

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<b>S. No.</b>	<b>Parameter</b>	<b>Semi Batch</b>	<b>Spray</b>
1	SO <sub>2</sub> concentration	99.45	98.23
2	NO concentration	98.81	97.24
3	Temperature	99.45	97.89
4	Mole ratio	98.68	96.98
5	Time	99.45	98.02
6.	Average removal efficiency	99.17	97.67
7.	Removal drop		1.51

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## Appendix B

**Table B1:** Standard Gibbs free energy, enthalpy, entropy change and specific heat capacities of the components (**Dean 1970**)

S. No.	Species	Physical State	$\Delta H$ , $\text{kJ}\cdot\text{mol}^{-1}$	$\Delta G^\circ$ , $\text{kJ}\cdot\text{mol}^{-1}$	$\Delta S^\circ$ , $\text{J}\cdot\text{K}^{-1}$ $\text{mol}^{-1}$	$\Delta C_p$ , $\text{J}\cdot\text{K}^{-1}$ $\text{mol}^{-1}$
1.	SO <sub>2</sub>	g	-296.81	-300.13	-248.223	39.88
2.	NO	g	91.29	87.60	210.76	29.85
3.	HClO standard state	aq.	-120.9	79.9	142	37.15
4.	NaClO standard state	aq.	-347	-298	100	
5.	NaCl standard State	aq.	-407.27	-393.17	115.5	-90.0
6.	Na <sub>2</sub> SO <sub>4</sub>	aq	-1387	-1270.2	149.6	128.2
7.	NaNO <sub>3</sub>	aq.	-1389.51	-1268.40	-138.1	-201
8.	Cl <sub>2</sub>	g	0	0	233	33.95
9.	H <sub>2</sub> O	l	-285.8	-237.14	69.95	75.28
10.	SO <sub>4</sub> <sup>-2</sup>	aq.	-909.34	-744.5	18.50	-293.0
11.	NO <sub>3</sub> <sup>-</sup> (aq.)	aq.	-206.85	-111.3	146.70	-86.6

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<b>12.</b>	Cl <sup>-</sup> standard state (aq.)	aq.	-167.08	-131.3	56.60	-136.4
<b>14.</b>	NH <sub>4</sub> <sup>+</sup> standard state (aq.)	aq.	-133.26	-79.37	111.17	79.9
<b>15.</b>	NH <sub>4</sub> OH standard state (aq.)	aq.	-361.2	-254.0	165.5	-68.6

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## Appendix C

**Table C1:** Data for removal efficiency with initial SO<sub>2</sub> concentration using NaClO in bubble column (initial NO concentration 1804 ppm, initial pH 5.6, initial NaClO concentration 0.032 M, temperature 305 K, Time 2 h)

S. No.	SO <sub>2</sub> concentration (ppm)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	4324	86.18	99.12
2	5481	88.23	99.14
3	6348	91.46	99.28
4	7238	92.26	99.73

**Table C2:** Data for removal efficiency with initial NO concentration using NaClO in bubble column (initial SO<sub>2</sub> concentration 6348 ppm, initial pH 5.6, initial NaClO concentration 0.032 M, temperature 305 K, Time 2 h)

S. No.	NO concentration (ppm)	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	1456	88.10	99.52

2	1702	89.59	99.67
3	1804	91.01	99.76
4	1941	91.29	99.81

**Table C3:** Data for removal efficiency with initial pH of NaClO in bubble column (initial SO<sub>2</sub> concentration 6348 ppm, initial NO concentration 1804 ppm, initial NaClO concentration 0.032 M, temperature 305 K, time 2 h)

S. No.	Initial pH	NO removal efficiency (%)	SO <sub>2</sub> removal efficiency (%)
1	4.3	87.28	99.41
2	5.1	90.81	99.73
3	5.6	91.62	99.82
4	6.4	87.67	99.88
5	7.9	80.25	99.91
6	9.1	62.42	99.49