

# A new occurrence of two-pyroxene granulites at Chicholi from Betul supracrustal belt in Central Indian Tectonic Zone (CITZ), MP, India

Manish Srivastava<sup>1,\*</sup>, S B Dwivedi<sup>1</sup> and S P  $Singh^2$ 

<sup>1</sup>Department of Civil Engineering, Indian Institute of Technology (BHU), Varanasi 221 005, India.

<sup>2</sup>Department of Geology, Institute of Earth Sciences, Bundelkhand University, Jhansi 284 128, India. \*Corresponding author. e-mail: manishs.rs.civ16@itbhu.ac.in

MS received 28 December 2018; revised 25 March 2019; accepted 2 May 2019; published online 22 July 2019

We report the new occurrence of two-pyroxene granulites from Chicholi, the Betul Group of the Central Indian Tectonic Zone (CITZ). The common mineral assemblage observed within different thin sections is orthopyroxene–clinopyroxene–hornblende–plagioclase–biotite–quartz. The textural relationship of these mineral phases shows the reaction: hornblende + quartz = orthopyroxene + clinopyroxene + plagio-clase. The estimated P-T condition of metamorphism of the two-pyroxene granulites is 901 ± 30°C and 8.68 ± 1.4 kbar.

Keywords. Pyroxene; granulite; CITZ; metamorphism; P-T condition.

# 1. Introduction

Granulites are considered as an important window to the lower crust as they provide us an important constraint for understanding their probable nature and composition. They also provide clues for the important processes - tectonic and chemical evolution which are responsible for crustal accretion, crustal stabilization and continental growth (Harley 1989). They are considered as the major modern analogues for the study of geodynamic evolution and tectonic history of the Earth. The coexistence of two pyroxenes in a granulite has been studied from several geological formations in India such as Southern Granulite Belt (Tsunogae and Santosh 2006 and references therein), Eastern Ghats Granulite Belt (Dasgupta et al. 1993 and references therein), Chhotanagpur Granite Gneiss Complex (Karmakar et al. 2011; Chatterjee 2018 and references therein) and Shillong Meghalaya Gneissic Complex (Dwivedi and Theunuo 2011 and

references therein). The supracrustal rocks of the Betul Group contain low-to-medium grade metamorphic rocks formed under amphibolite facies (Roy et al. 2003). It is an important factor which provides the clue that the protoliths might have been metamorphosed up to granulite facies, as Betul supracrustals are an active part of the Central Indian Tectonic Zone (CITZ) (Bhowmik and Roy 2003; Santosh et al. 2009; Naganjaneyulu and Santosh 2010; Vansutre and Hari 2010 and references therein). The CITZ has been considered in earlier studies as a trans-continental suture between the northern Bundelkhand Craton and Southern Bastar Craton (Harris 1993). The granulite belts have already been described from northern Bundelkhand Craton (Sharma 1988; Guha and Bhattacharya 1995; Dasgupta et al. 1997; Singh and Dwivedi 2009 and references therein) and from southern Bastar Craton (Narsimha Prakash et al. 1996; Bhowmik et al. 1999; Bhowmik and Dasgupta 2004; Das et al. 2008 and references therein). We report the occurrence of two-pyroxene granulites from the area of Chicholi of the Betul Group within CITZ. The petrography, mineral chemistry and geothermobarometry have been discussed in detail which provide imprints of granulite facies metamorphism within the given rock suite of the Betul Group.

## 2. Geological setting

The Central Indian region is characterised by the occurrence of small-sized granulite belts within the CITZ and adjoining cratons. The CITZ is a Proterozoic mobile belt whose regional trend is ENE–WSW. It is the zone of amalgamation of the two major cratonic blocks namely, Bundelkhand craton in the north and Bastar Craton in the south (Radhakrishna 1989; Naganjaneyulu and Santosh 2010; Acharvya and Roy 2011) (figure 1). The CITZ is sandwiched between the Son Narmada North Fault in the north and the Central Indian Shear (CIS) zone in the south. It incorporates at least three distinct supracrustal belts namely, Mahakoshal, Betul and Sausar which are composed of low-to-medium-grade metamorphosed supracrustal rocks set in largely undifferentiated gneisses and granitoids (Rov *et al.* 2003; Vansutre and Hari 2010). Within these supracrustals of CITZ, four prominent granulite belts are present namely:

- Balaghat–Bhandara Granulite Belt (BBG) of Sausar Series.
- Ramakona–Katangi Granulite Belt (RKG) of Sausar Series.
- Chhatuabhavna Granulite (CBG) of Bilaspur– Raigarh Belt.
- Makrohar Granulite Belt (MGB) of Mahakoshal supracrustals.

Chicholi village  $(21^{\circ}59'603''N; 77^{\circ}41'140''E)$  is located in the Betul supracrustal belt, whose regional trend is ENE–WSW. The rock formations of the belt designated as the Betul Group are considered as a prominent division of CITZ (figure 1). The structural characteristics reveal that the supracrustal rocks have been deformed by three episodes. The first generation structures (D<sub>1</sub>) are represented by isoclinal to reclined folds; the second generation folds (D<sub>2</sub>) are tight to isoclinal, upright to inclined which define the ENE–WSW regional tectonic trend of the belt. The D<sub>2</sub> structures have been affected by superposed folding  $(D_3)$  that has given rise to N–S trending broad open folds. The  $D_1$  and  $D_2$  deformations are accompanied by greenschist to lower amphibolite facies metamorphism (Roy *et al.* 2003).

# 3. Petrography

The two-pyroxene granulites of the study area are dark grey to greyish black in colour and are medium to coarse-grained (figure 2a). The rock exhibits granoblastic/granulitic texture with granular mosaic of orthopyroxene, clinopyroxene, hornblende and plagioclase. Biotite is also present in few thin sections which exhibit weak foliation. The twopyroxene granulites contain the following mineral assemblage: (i) orthopyroxene–clinopyroxene–hornblende–plagioclase–biotite–quartz.

*Hornblende* is strongly pleochroic with green, vellowish-green, bluish-green, dark green or greenish-brown pleochroic colours (figure 2b and c). The pleochroic colour in hornblende varies with a change in chemical composition which is given in table 3. Hornblende is of two generations, hornblende1 and hornblende2 in which hornblende1 is corroded and occurs as inclusion within orthopyroxene providing evidence of breakdown of hornblende1 in the presence of quartz into two pyroxenes (figure 2b). Hornblende2 is formed in the late stage during retrogression of pyroxenes (figure 2c). The mineral chemistry of hornblende is given in table 4 which shows variation in chemical composition.

Orthopyroxene is hypersthene which is colourless to pale green, slightly pleochroic and the grains are sub-idioblastic to idioblastic. Hypersthene shows low birefringence in polarized light and first-order interference colour, parallel extinction under crossed nicols of the petrological microscope. Orthopyroxene contains the inclusion of corroded hornblende (Hbl1) and quartz (figure 2b), which shows a prograde reaction as:

hornblende1 + quartz = orthopyroxene

$$+ \text{clinopyroxene} + \text{plagioclase} + \text{H}_2\text{O}.$$
 (I)

At some places, orthopyroxene is also partially encircled by hornblende2 (figure 2c), which shows a retrograde reaction as:

$$orthopyroxene + clinopyroxene + plagioclase$$

$$+ H_2O = hornblende2 + quartz.$$
 (II)







Figure 2. (a) Photograph of the two-pyroxene granulites in field which occur in the form of patches. (b) Photomicrograph of coexisting mineral phases in orthopyroxene coexisting with clinopyroxene and plagioclase contain inclusions of corroded hornblende and quartz under plane-polarised light. (c) Photomicrograph shows the inclusions of orthopyroxene, clinopyroxene and plagioclase in hornblende coexisting with quartz (plane-polarised light). (d) Photomicrograph of coexisting mineral phases shows inclusions of biotite and quartz in orthopyroxene under plane-polarised light. Where, Opx = orthopyroxene; Cpx = clinopyroxene, Hbl = hornblende, Plag = plagioclase, Bt = biotite, Kfs = alkali-feldspar and <math>Qz = quartz (Whitney and Evans 2010).

Orthopyroxene also rims flakes of biotite along with quartz inclusions (figure 2d), which indicates a prograde reaction as:

Biotite + quartz = orthopyroxene + alkali feldspar+ vapour. (III)

*Clinopyroxene* is diopside. It is colourless to pale yellow, slightly pleochroic and shows inclined extinction. Reaction (I) can also be evidenced by the occurrence of inclusions of hornblende and plagioclase in clinopyroxene (figure 2b).

*Plagioclase* occurs as coarse aggregates, lathshaped grains and it is characterized by polysynthetic/lamellar twinning. *Biotite* is also a dominant mineral phase and it occurs as medium-grained, subidioblastic flakes. It shows strong pleochroism from colourless to light brown to dark brown to light yellow.

## 4. Mineral chemistry

Different mineral phases and assemblages were identified by detailed petrographic studies (table 1). The coexisting mineral phases have been analysed by using an electron probe micro analyser (EPMA). The analytical work was performed by using an EPMA CAMECA SXFive instrument at DST-SERB National Facility, Department of

Table 1. Representative approximate modal composition (in percentage) of the two-pyroxene granulites observed under a petrological microscope through Leica Qwin software.

Serial	Sample	Minerals						
no.	no.	Opx	Cpx	Hbl	Plg	Bt	Qtz	Kfs
1	CH-01	25	20	10	20	10	10	5
2	CH-02	25	20	15	25	5	5	5
3	CH-03	20	20	10	20	15	5	5
4	CH-04	30	25	15	20	5	5	0
5	CH-05	25	30	25	15	0	5	0
6	CH-06	20	30	15	25	0	10	0
7	CH-07	15	35	15	20	10	5	0
8	CH-08	30	30	10	15	10	5	0
9	CH-09	25	25	15	20	10	5	0
10	CH-10	30	20	10	10	15	10	5
11	CH-11	25	30	15	20	5	5	0
12	CH-12	35	25	10	25	0	5	0
13	CH-13	25	35	15	15	10	0	0
14	CH-14	30	30	20	10	5	0	0
15	CH-15	25	20	15	25	10	5	0
16	CH-16	30	25	20	15	5	5	0
17	CH-17	25	25	10	25	15	0	0
18	CH-18	20	30	20	10	10	5	5
19	CH-19	30	25	15	25	0	5	0

Table 2. EPMA data and structural formula of orthopyroxene from two-pyroxene granulites on six oxygen basis.

Sample no.	CH	-03	CH	-04	CH	-06	CH	-12
Oxides	Core $1/1$	$\operatorname{Rim} 1/1$	Core $2/1$	$\operatorname{Rim} 2/1$	Core $3/1$	$\operatorname{Rim} 3/1$	$\overline{\text{Core } 4/1}$	Rim $4/1$
$SiO_2$	50.02	50.35	50.23	50.00	49.90	50.14	50.33	50.50
$TiO_2$	0.06	0.07	0.09	0.09	0.09	0.08	0.13	0.11
$Al_2O_3$	1.08	1.26	1.16	1.06	1.12	1.02	1.24	1.13
$Cr_2O_3$	0.01	0.00	0.02	0.00	0.02	0.01	0.04	0.04
$\rm FeO$	31.08	31.58	30.45	30.81	31.22	31.10	29.82	30.13
MnO	0.71	0.96	0.55	0.56	0.94	0.94	0.61	0.65
MgO	15.12	15.41	16.04	15.59	15.51	15.67	16.72	16.29
CaO	0.76	0.72	0.70	0.68	0.90	0.71	0.82	1.17
$Na_2O$	0.08	0.07	0.09	0.17	0.17	0.12	0.00	0.01
$K_2O$	0.00	0.00	0.00	0.04	0.09	0.02	0.00	0.00
Total	99.19	100.63	99.45	99.14	100.18	100.08	99.95	100.37
$X_{\rm Mg}$	0.46	0.46	0.48	0.47	0.46	0.47	0.49	0.49
Si	1.97	1.96	1.96	1.96	1.95	1.96	1.95	1.96
Ti	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.05	0.05	0.05	0.04	0.05	0.04	0.05	0.05
$\mathbf{Cr}$	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
$\mathrm{Fe}^{2+}$	1.02	1.02	0.99	1.01	1.02	1.01	0.96	0.97
Mn	0.02	0.03	0.01	0.01	0.03	0.03	0.02	0.02
Mg	0.88	0.89	0.93	0.91	0.90	0.91	0.96	0.94
Ca	0.03	0.03	0.02	0.02	0.03	0.03	0.03	0.04
Na	0.00	0.00	0.00	0.01	0.01	0.00	0.00	0.00
Κ	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Total	4.00	4.01	4.00	4.01	4.02	4.01	4.01	4.00

 $X_{\rm Mg} = {\rm Mg}/({\rm Mg} + {\rm Fe}).$ 

Sample no.	CH-03		CH-04		CH-06		CH-12	
Oxides	Core $1/1$	Rim $1/1$	Core $2/1$	$\operatorname{Rim} 2/1$	Core $3/1$	Rim $3/1$	Core $4/1$	Rim $4/1$
$SiO_2$	50.47	50.23	50.57	49.94	50.01	50.32	49.85	49.77
$TiO_2$	0.27	0.22	0.23	0.30	0.25	0.26	0.28	0.26
$Al_2O_3$	2.12	2.13	2.21	2.53	2.52	2.50	2.23	1.85
$Cr_2O_3$	0.00	0.02	0.00	0.00	0.00	0.03	0.04	0.06
FeO	12.88	12.64	14.09	14.55	14.05	13.91	12.31	13.66
MnO	0.27	0.25	0.34	0.32	0.43	0.47	0.29	0.24
MgO	11.19	11.00	11.20	10.75	11.01	10.76	11.00	11.17
CaO	21.40	21.52	20.78	19.66	20.29	21.02	21.86	21.21
Na <sub>2</sub> O	0.39	0.65	0.41	0.52	0.50	0.43	0.48	0.44
$K_2O$	0.00	0.04	0.00	0.04	0.00	0.00	0.03	0.00
Total	99.14	99.15	100.04	98.85	99.40	100.00	98.69	98.89
$X_{\mathrm{Mg}}$	0.60	0.60	0.58	0.56	0.58	0.57	0.61	0.59
Si	1.93	1.93	1.93	1.93	1.92	1.92	1.92	1.92
Ti	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.09	0.09	0.10	0.11	0.11	0.11	0.10	0.08
$\operatorname{Cr}$	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
$\mathrm{Fe}^{2+}$	0.41	0.40	0.45	0.47	0.45	0.44	0.39	0.44
Mn	0.00	0.00	0.01	0.01	0.01	0.01	0.00	0.00
Mg	0.64	0.63	0.63	0.62	0.63	0.61	0.63	0.64
Ca	0.88	0.88	0.85	0.81	0.83	0.86	0.90	0.88
Na	0.02	0.04	0.03	0.03	0.03	0.03	0.03	0.03
Κ	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Total	4.01	4.03	4.02	4.01	4.02	4.02	4.02	4.03

Table 3. EPMA data and structural formula of clinopyroxene from two-pyroxene granulites on six oxygen basis.

 $X_{\rm Mg} = {\rm Mg}/({\rm Mg} + {\rm Fe}).$ 

Geology (Center of Advanced Study), Institute of Science, Banaras Hindu University. A total of 20-nm thick carbon coating was polished on thin sections for electron probe micro analyses with the help of LEICA-EM ACE200. EPMA was operated at a voltage of 15 kV and a current of 10 nA with the help of SXFive software. The source used for the generation of electron beam was  $LaB_6$ . The internal standard used to verify positions of crystals (SP1-TAP, SP2-LiF, SP3-LPET, SP4-LTAP and SP5-PET) was and radite, which is a natural silicate mineral. The verified position was with respect to the corresponding wavelength dispersive spectrometers (SP#) in a CAMECA SX-Five instrument. The following X-ray lines were used in the analyses: F-K $\alpha$ , Na-K $\alpha$ , Mg-K $\alpha$ , Al-K $\alpha$ , Si-K $\alpha$ , P-Ka, Cl-Ka, K-Ka, Ca-Ka, Ti-Ka, Cr-Ka, Mn-Ka, Fe-K $\alpha$  and Ni-K $\alpha$ . The natural mineral standards used during analysis are: fluorite, halite, periclase, corundum, wollastonite, apatite, orthoclase, rutile, chromite, rhodonite, hematite and pure Ni metal standard supplied by CAMECA-AMETEK, which were used for routine calibration and quantification.

Routine calibration, acquisition, quantification and data processing were carried out by using SxSAB version 6.1 and SX-result software of CAMECA (Sharma *et al.* 2018). EPMA data of the coexisting mineral phases with their calculated structural formula are given in tables 2–6.

The EPMA data of pyroxene were plotted in a triangular end-member  $\text{CaSiO}_3-\text{MgSiO}_3-\text{Fe}^{2+}\text{SiO}_3$  diagram (figure 3). The orthopyroxene plots at  $\text{En}_{46-49}$  near the hypersthene, which is the solid solution between Mg and Fe end-members of orthopyroxene ( $X_{\text{Mg}} = 0.46-0.49$ ) and the Al content of the orthopyroxene varies between 0.04 and 0.05 per formula unit (pfu, on the basis of six oxygens). The clinopyroxene plots in the field of salite near the diopside end of the diagram ( $X_{\text{Mg}} = 0.59-0.61$ ) and the Al content ranges from 0.08 to 0.11 pfu. The Ca content of clinopyroxene varies between 0.81 and 0.90 pfu.

The hornblende in two-pyroxene granulites is of two generation which is clearly indicated in mineral chemistry obtained through EPMA. The Mg content of Hbl1 ranges from 2.00 to 2.03 pfu in its core portion and it is a part of the prograde metamorphic reaction. The Mg content of Hbl2 ranges from 1.92 to 1.94 pfu in its core portion and it is retrograded in nature. The overall Al content of hornblende varies from 1.85 to 2.05 pfu from the structural formula calculated at 23 oxygen basis. The  $X_{Mg}$  of hornblende varies from 0.42 to 0.46. The Ti content of hornblende varies between 0.24 and 0.31 pfu and wt% of TiO<sub>2</sub> varies from 2.12 to 2.75. Hornblende is calcic in nature, which is evident from high calcium content ranging from 1.87 to 1.98 pfu. The EPMA data plot for amphibole classification (Leake *et al.* 1997) is shown in figure 4.

The  $X_{\rm Mg}$  content of biotite ranges from 0.43–0.45 which shows predominance of iron over magnesium. The TiO<sub>2</sub> content ranges from 3.56% to 3.96%. The Al<sub>2</sub>O<sub>3</sub> content ranges from 13.67% to 14.81% while Al<sup>IV</sup> content lies in between 2.42 and 2.52 pfu and Al<sup>VI</sup> ranges from 0.11 to 0.20 pfu.

The EPMA data and the calculated structural formula indicate the presence of labradorite type

of plagioclase which is rich in calcium than potassium. The An-content of plagioclase ranges from  $An_{50}$  to  $An_{58}$  which was calculated using the relation Ca/(Ca + Na + K).

#### 5. Geothermobarometry

The metamorphic condition was estimated by quantification of P-T condition for the granulites by orthopyroxene–clinopyroxene conventional exchange geothermobarometers (table 6), and also from an internally consistent dataset with the help of Thermocalc v. 3.21 (Holland and Powell 1998) at 9 kbar (table 7). The estimates of temperature of the coexisting orthopyroxene–clinopyroxene from the granulites range between 845 and 962°C for the core and 779–870°C for the rim. The average temperature of the core is  $889 \pm 35^{\circ}$ C and the average temperature of the rim is  $800 \pm 38^{\circ}$ C. The average P-T condition of metamorphism

Table 4. EPMA data and structural formula of hornblende from two-pyroxene granulites on 23 oxygen basis.

Sample no.	CH-04 (Hbl1)		CH-06 (Hbl1)		CH-12	(Hbl2)	CH-03 (Hbl2)		
Oxides	Core $2/1$	$\operatorname{Rim} 2/1$	Core $3/1$	$\operatorname{Rim} 3/1$	Core $4/1$	$\operatorname{Rim} 4/1$	Core $1/1$	Rim $1/1$	
$SiO_2$	40.81	41.15	40.64	40.39	39.76	40.42	40.24	41.31	
$TiO_2$	2.37	2.75	2.12	2.55	2.65	2.37	2.20	2.30	
$Al_2O_3$	10.99	10.22	11.25	11.10	11.01	10.69	10.73	10.71	
$Cr_2O_3$	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
FeO	18.88	19.33	18.84	19.12	19.20	19.28	19.45	18.32	
MnO	0.08	0.19	0.16	0.29	0.25	0.21	0.27	0.28	
MgO	8.67	9.01	8.81	8.34	8.31	8.19	8.23	9.05	
CaO	11.37	11.77	11.68	11.42	11.48	11.73	11.78	11.31	
$Na_2O$	1.30	1.22	1.24	1.17	1.43	1.22	1.18	1.17	
$K_2O$	1.77	1.70	1.83	1.84	1.91	1.77	1.80	1.77	
$\rm ZrO_2$	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Total	96.91	97.97	97.51	96.97	96.81	96.55	96.57	96.86	
$X_{\rm Mg}$	0.44	0.45	0.45	0.43	0.43	0.42	0.43	0.46	
Si	6.34	6.34	6.30	6.29	6.23	6.33	6.32	6.39	
Ti	0.27	0.31	0.24	0.29	0.31	0.27	0.25	0.26	
Al	2.01	1.85	2.05	2.04	2.03	1.97	1.98	1.95	
$\operatorname{Cr}$	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
$\mathrm{Fe}^{2+}$	2.45	2.49	2.44	2.49	2.51	2.52	2.55	2.37	
Mn	0.01	0.02	0.02	0.03	0.03	0.02	0.03	0.03	
Mg	2.00	2.07	2.03	1.94	1.94	1.91	1.92	2.09	
Ca	1.88	1.94	1.94	1.90	1.93	1.97	1.98	1.87	
Na	0.39	0.36	0.37	0.35	0.43	0.37	0.36	0.35	
Κ	0.35	0.33	0.36	0.36	0.38	0.35	0.00	0.34	
$\operatorname{Zr}$	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Total	15.74	15.75	15.78	15.74	15.83	15.75	15.78	15.70	

 $X_{\rm Mg} = {\rm Mg}/({\rm Mg} + {\rm Fe}).$ 

Sample no.	CH-03		CH	-04	CH-12		
Oxides	Core $1/1$	$\operatorname{Rim} 1/1$	Core $2/1$	$\operatorname{Rim} 2/1$	Core $3/1$	Rim $3/1$	
$SiO_2$	35.04	34.99	35.11	35.02	35.12	34.96	
$TiO_2$	3.56	3.76	3.79	3.96	3.72	3.61	
$Al_2O_3$	13.89	14.05	13.88	13.67	13.80	14.81	
$Cr_2O_3$	0.04	0.03	0.02	0.01	0.01	0.06	
FeO	21.25	22.09	21.22	21.56	21.22	21.96	
MnO	0.07	0.07	0.06	0.13	0.00	0.12	
MgO	10.05	9.78	9.98	9.59	9.80	9.88	
CaO	0.02	0.00	0.00	0.04	0.04	0.01	
BaO	0.00	0.00	0.00	0.00	0.00	0.00	
Na <sub>2</sub> O	0.23	0.21	0.19	0.28	0.31	0.13	
$K_2O$	8.30	8.08	8.09	8.30	8.30	8.26	
Cl	0.26	0.31	0.31	0.38	0.39	0.26	
F	0.37	0.04	0.18	0.37	0.34	0.36	
Total	93.37	93.59	92.91	93.36	93.31	94.75	
$X_{\mathrm{Mg}}$	0.45	0.44	0.45	0.43	0.45	0.44	
Si	5.56	5.52	5.56	5.55	5.57	5.47	
$Al^{IV}$	2.43	2.47	2.43	2.44	2.42	2.52	
Cr	0.00	0.00	0.00	0.00	0.00	0.00	
$Al^{VI}$	0.15	0.14	0.15	0.11	0.15	0.20	
Ti	0.42	0.44	0.45	0.47	0.44	0.42	
$\mathrm{Fe}^{2+}$	2.81	2.91	2.81	2.86	2.81	2.87	
Mn	0.01	0.01	0.00	0.01	0.00	0.01	
Ba	0.00	0.00	0.00	0.00	0.00	0.00	
Mg	2.37	2.30	2.35	2.27	2.31	2.30	
Ca	0.00	0.00	0.00	0.00	0.00	0.00	
Na	0.07	0.06	0.05	0.08	0.09	0.04	
K	1.68	1.62	1.63	1.68	1.68	1.65	
Cl	0.07	0.08	0.08	0.10	0.10	0.07	
F	0.18	0.02	0.09	0.18	0.17	0.17	
Total	15.55	15.52	15.49	15.52	15.52	15.53	

Table 5. EPMA data and structural formula of biotite from two-pyroxene granulites on 22 oxygen basis.

 $X_{\rm Mg} = {\rm Mg}/({\rm Mg} + {\rm Fe}).$ 

 $(P-T_{\rm av})$  was estimated using Thermocalc v. 3.21 with EPMA data of two-pyroxene granulites and with phases involving orthopyroxene, clinopyroxene, hornblende and plagioclase. The estimated average P-T condition of granulite facies metamorphism is 901 ± 30°C/8.68 ± 1.4 kbar and the independent set of reactions involved are given in table 8.

# 6. Discussion

The two-pyroxene granulites have been studied and identified based on petrography and mineral chemistry. The different thin sections were examined and mineral phases were identified with the help of their characteristic optical properties. The mutual relationship of different mineral phases led to the establishment of definite metamorphic reactions. Mineral chemistry was studied with the help of the EPMA analytical technique. The twopyroxene granulites of the Chicholi region are found to contain orthopyroxene-clinopyroxenehornblende-plagioclase-biotite-quartz as an essential mineral assemblage. Based on the EPMA analytical studies, it has been proved that the orthopyroxene is hyperstheme  $(En_{46-49})$  and the clinopyroxene is salite which is a characteristic feature of basic granulites. The anorthite content of plagioclase, An<sub>49</sub>–An<sub>51</sub> suggests the presence of labradorite. The textural relationship of minerals shows the evidence of the following reactions: (i) hornblende1 + quartz = orthopyroxene + clino $pyroxene + plagioclase + H_2O$ , which is evident from the relicts of hornblende within the hyper-

Table 6. EPMA and structural formula of plagioclase from two-pyroxene granulites on eight oxygen basis.

Sample no.	CH	-03	CH-04		CH-06 CH		CH-12
Oxides	Core $1/1$	Rim $1/1$	Core $2/1$	$\operatorname{Rim} 2/1$	Core $3/1$	Rim $3/1$	Core $4/1$
$SiO_2$	52.81	54.68	54.59	54.84	55.00	55.00	51.10
$Al_2O_3$	27.28	27.08	27.82	27.01	27.65	27.80	30.43
FeO	0.15	0.00	0.00	0.00	0.00	0.00	0.04
CaO	11.63	11.06	10.89	10.82	10.95	10.48	12.24
$Na_2O$	5.71	5.53	5.29	5.71	5.42	5.58	4.67
$K_2O$	0.59	0.13	0.11	0.13	0.10	0.08	0.18
BaO	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Total	98.73	98.77	98.84	98.76	99.22	99.11	98.99
$X_{\rm Ca}$	0.51	0.52	0.52	0.50	0.52	0.50	0.58
Si	9.80	10.02	9.96	10.04	10.00	10.00	9.40
Al	5.96	5.85	5.98	5.83	5.92	5.96	6.60
$\mathrm{Fe}^{2+}$	0.02	0.00	0.00	0.00	0.00	0.00	0.00
Ca	2.31	2.17	2.13	2.12	2.13	2.04	2.41
Na	2.05	1.96	1.87	2.02	1.91	1.96	1.66
Κ	0.14	0.03	0.02	0.03	0.02	0.02	0.04
Ba	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Total	20.30	20.04	19.98	20.06	20.00	20.00	20.14

 $X_{\rm Ca} = {\rm Ca}/({\rm Ca} + {\rm K} + {\rm Na}).$ 



Figure 3. Representation of EPMA data of pyroxene shown in pyroxene quadrilateral (part of the triangular diagram  $CaSiO_3$ -MgSiO\_3-FeSiO\_3). The clinopyroxene coexisting with orthopyroxene is joined by a line.

sthene (orthopyroxene) and (ii) biotite + quartz = orthopyroxene + K-feldspar +  $H_2O$  which is shown by the inclusions of biotite flakes within orthopyroxene grains and both the reactions are prograde reactions. It is clear from the reactions that hornblende becomes unstable in the presence of quartz and recrystallises in orthopyroxene, clinopyroxene and plagioclase in response to the changing P-T condition. Similarly, biotite gives the stable phases of orthopyroxene and alkali feldspar by the above reaction in response to the metamorphic process. In few thin sections, textural relations where both orthopyroxene and clinopyroxene are partially rimmed by hornblende suggest the retrograde reaction: Orthopyroxene + clinopyroxene + plagioclase + H<sub>2</sub>O = hornblende2 + quartz. Based on the



Figure 4. Representation of the EPMA data of amphibole (after Leake et al. 1997).

Table 7.	Temperature	estimates of	of two-pyroxene	granulites	from	conventional	exchange	geothermometer	at
assumed	pressure.								

Estimate of geothermometers (temp. in $^{\circ}$ C) at 9 kbar								
Models	Wood and Banno (1973)		Wells	(1977)	Powell (1978)			
Domain	Core (°C)	Rim (°C)	Core (°C)	Rim (°C)	Core (°C)	Rim (°C)		
CH-03	845	793	891	847	899	851		
CH-04	869	825	902	856	903	870		
CH-06	872	780	898	849	909	845		
CH-12	844	779	880	796	962	849		
Average	$857\pm25$	$794 \pm 30$	$892\pm25$	$837 \pm 42$	$918\pm40$	$853\pm30$		

Average temperature for core =  $889 \pm 35^{\circ}$ C. Average temperature for rim =  $828 \pm 38^{\circ}$ C.

Average temperature for  $\min = 620 \pm 50$  C.

Table 8. Pressure and temperature estimates of two-pyroxene granulites using internally consistent dataset by Thermocalc v3.21.

Sample no.	Average pressure in kbar $(P_{\rm av})$	Average temperature in °C $(T_{av})$	Average pressure and temperature $(P-T_{av})$
CH-03	$10.55 \pm 1.2$	$845 \pm 42$	$894 \pm 45^{\circ}C/9.31 \pm 1.5$ kbar
CH-04	$9.43 \pm 1.3$	$963 \pm 45$	$879 \pm 55^{\circ}C/8.44 \pm 1.6$ kbar
CH-06	$8.87 \pm 1.5$	$896 \pm 35$	$905 \pm 40^{\circ} \text{C}/8.73 \pm 1.2 \text{ kbar}$
CH-12	$9.12 \pm 1.5$	$904 \pm 50$	$927 \pm 35^{\circ}\text{C}/8.24 \pm 1.5 \text{ kbar}$
Average	$9.49 \pm 1.5$	$902 \pm 52$	$901 \pm 30^{\circ} \text{C}/8.68 \pm 1.4 \text{ kbar}$

Independent set of reactions involved in P-T estimation:

1. 5ts + 10hed + 10q = 3tr + 2fact + 10an.

2. ts + en + 2hed + 2q = tr + 2an + fs.

3. 2ts + 5fs + 4di + 4q = 2fact + 4an + 5en.

4. parg + di + 5q = tr + an + ab.

5.  $6fact + 16ts + 12ab + 38cats = 12parg + 58an + 15fs + 10H_2O.$ 

6. parg + 2fs + hed + 4q = fact + ab + 2en + cats.

Symbols. di: diopside, hed: hedenbergite, en: enstatite, fs: ferrosilite, tr: tremolite, fact: ferroactinolite, ts: tschermakite, parg: pargasite, an: anorthite, ab: albite, q: quartz, and  $H_2O$ : water fluid.

geothermobarometric calculations it has been derived that the two-pyroxene granulites were formed under the ultra-high temperature (UHT) conditions (at 900-1100°C and 7-13 kbar) of metamorphism (Harley 1998). The average pressure of the two-pyroxene granulites of the study area is  $\sim 8.75$  kbar, which corresponds to about 30 km (3.5 km/kbar) depth of burial of the protoliths and the present average crustal thickness of the central Indian region is  $\sim 35$  km. This indicates that the crust was doubly thickened to  $\sim 65$  km at the time of the thermal peak of metamorphism during the evolution of granulites (Ellis 1987) and suturing of the Northern Indian block and Southern Indian block along the CITZ occurred. Apart from this, lateral transportation of lower crustal rocks may have contributed to the crustal thickening in addition to under plate and intraplate magmatism, metamorphism and tectonism. These processes might also have facilitated the evolution of the granulites (Dessai *et al.* 2010). Further, they might have been exhumed onto the surface due to tectonic processes and have suffered denudation by the effect of geological agencies which clearly attribute to their occurrence on the surface as patches.

#### 7. Conclusion

Based on the work carried out, it has been conrmed that the rock samples collected from the area of Chicholi are two-pyroxene granulites which occur within the supracrustal rocks of the Betul group. The common mineral assemblage inferred from the studies is orthopyroxene–clinopyroxene– hornblende–plagioclase–biotite–quartz. Based on geothermobarometry, it is evident that the protoliths have been metamorphosed under high-grade granulite facies. The pressure–temperature condition of two-pyroxene granulites is 901  $\pm$  30°C and 8.68  $\pm$  1.4 kbar.

#### Acknowledgements

We are thankful to the Director, Indian Institute of Technology (BHU) for providing infrastructures and funds for the completion of this work. We also extend our gratitude towards the department for providing valuable Research Support Grant needed for research purpose. We appreciate the efforts of both anonymous reviewers for their constructive comments to improve the quality of the manuscript.

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